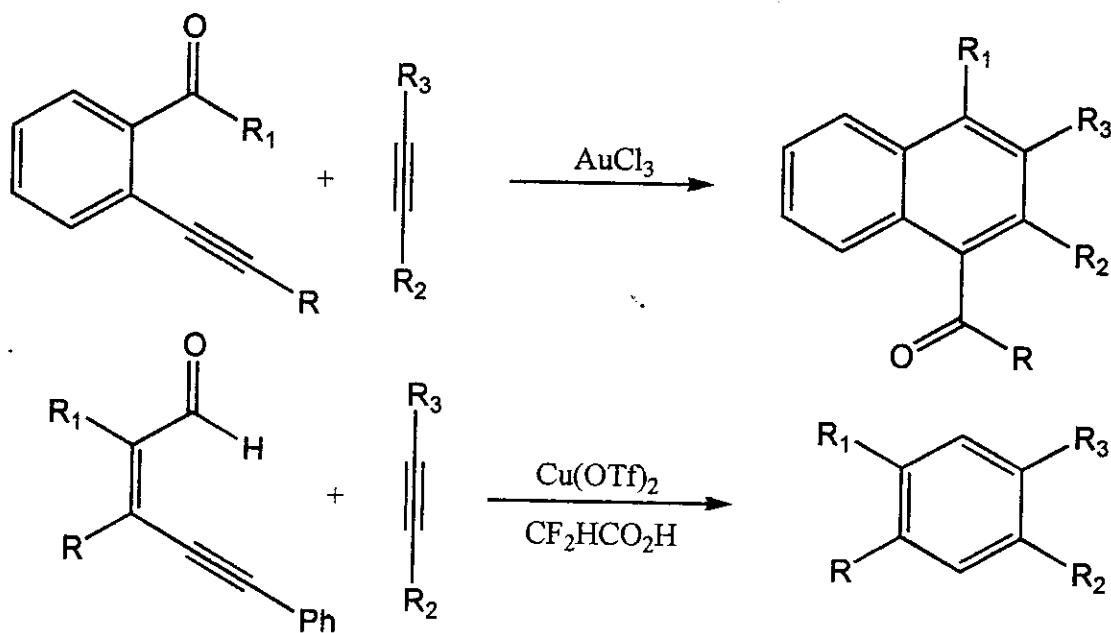


Lewis Acid-Catalyzed Benzannulation via Unprecedented [4+2] Cycloaddition of *o*-Alkynyl(oxo)benzenes and Enynals with Alkynes

Asao, N.; Nogami, T.; Lee, S.; Yamamoto, Y.
J. Am. Chem. Soc. **2003**, *124*, ASAP

AuCl₃-Catalyzed Benzannulation: Synthesis of Naphthyl Ketone Derivatives from *o*-Alkynylbenzaldehydes with Alkynes

Asao, N.; Takahashi, K.; Lee, S.; Kasahara, T.; Yamamoto, Y.
J. Am. Chem. Soc. **2002**, *124*, 12650-12651



Contents

Introduction (Transition Metal-Catalyzed Benzannulation)

(1) Zr-Catalyzed [2+2+2] Trimerization

(2) Ti-Catalyzed [2+2+2] Trimerization

(3) Pd-Catalyzed Benzannulation

(4) W-Catalyzed Benzannulation

(5) Rh-Catalyzed Benzannulation

(6) Ir-Catalyzed Benzannulation

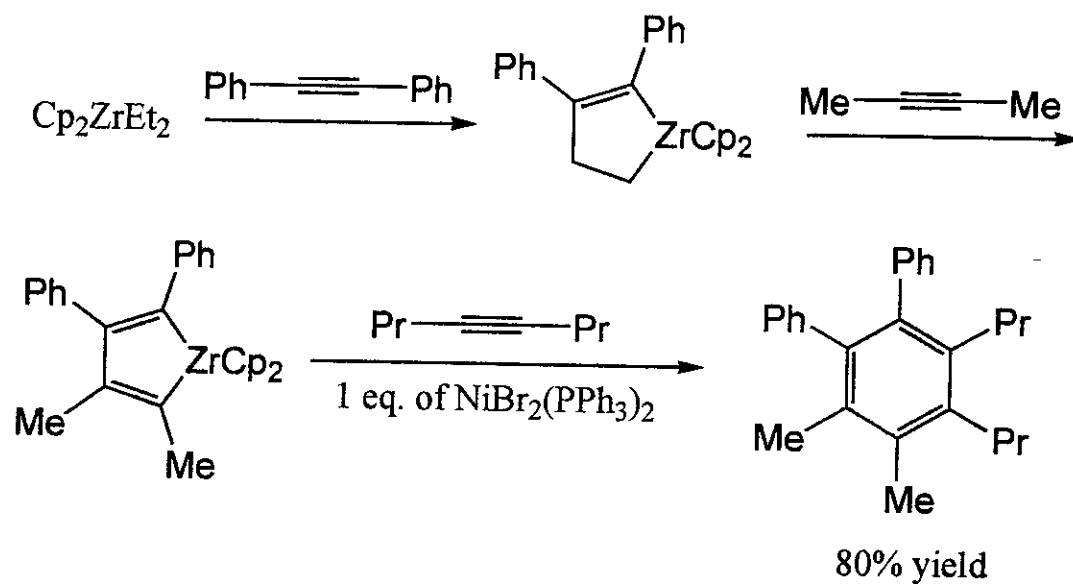
Today's topic (Lewis Acid-Catalyzed Benzannulation)

(1) Au-Catalyzed Benzannulation

(2) Cu-Catalyzed Benzannulation

New type of acetylen trimerization-1

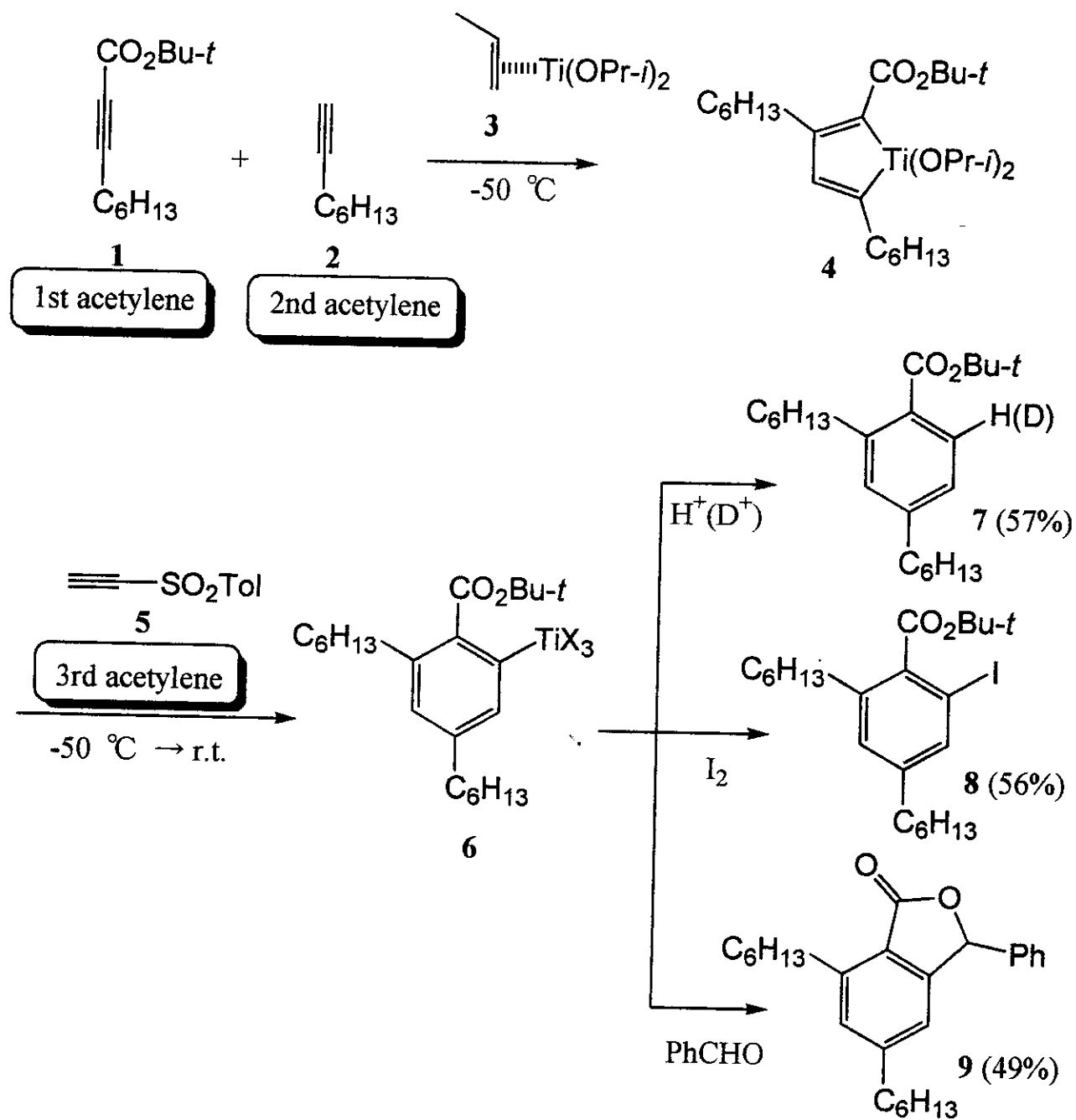
Benzannulation through Zirconacycles



Takahashi, T etc. *J. Am. Chem. Soc.* **1999**, *121*, 11093-11100

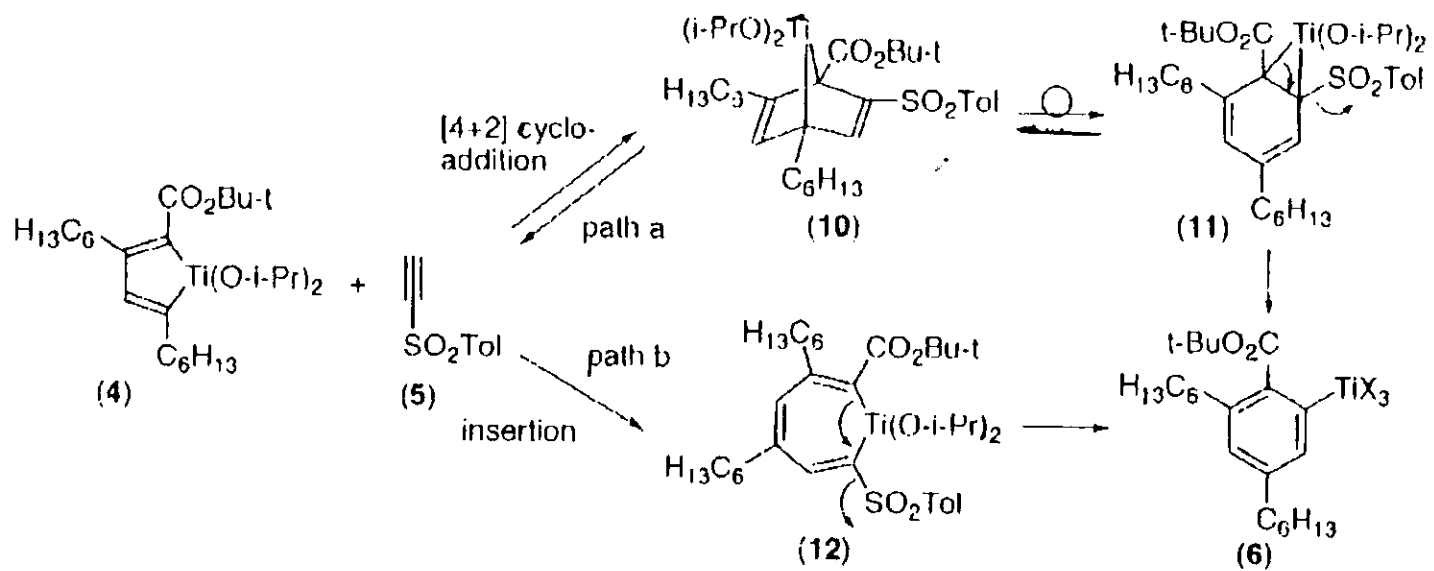
New type of acetylen trimerization-2

Benzannulation via Titanacycles



Sato, F. etc. *J. Am. Chem. Soc.* **2001**, *123*, 7925-7926

Reaction Mechanism

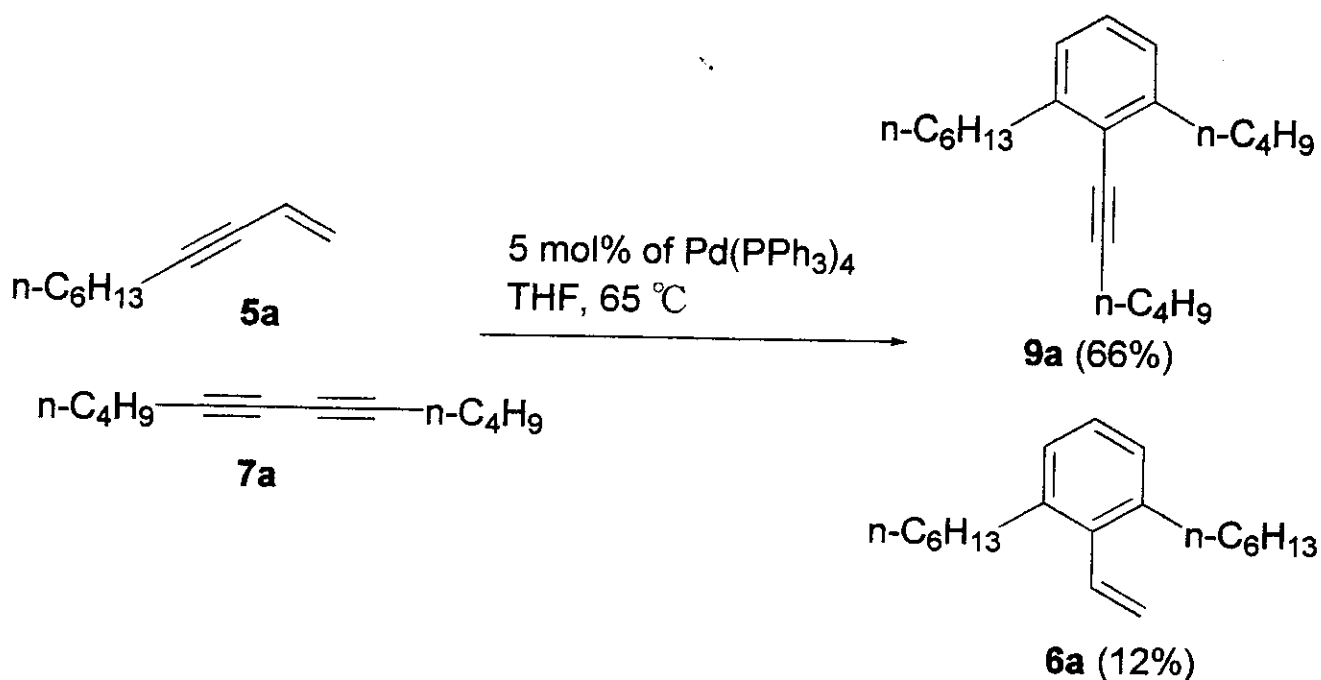
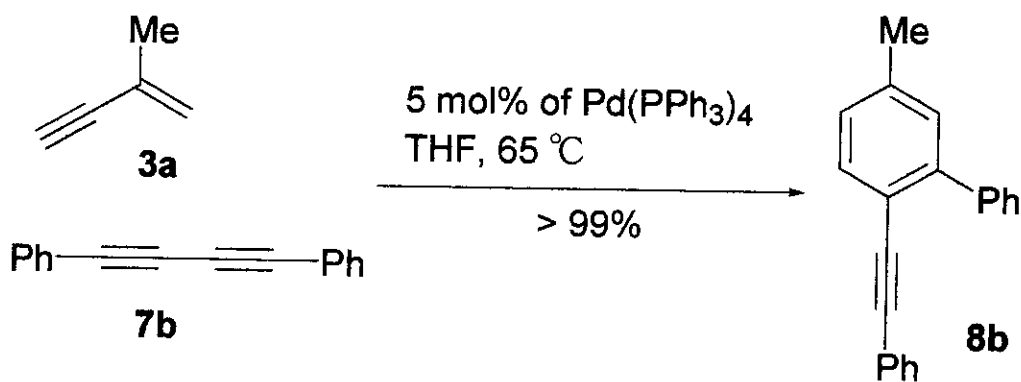
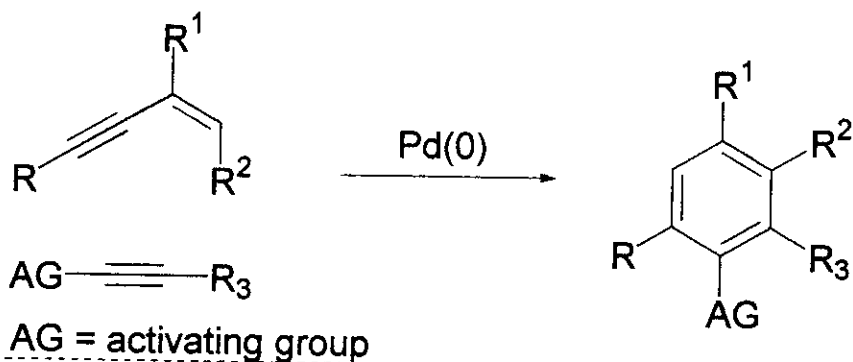


Transition metal-catalyzed benzannulation (1)

Palladium-Catalyzed [4+2] Cross-Benzannulation reaction of Conjugated Enynes with Diynes

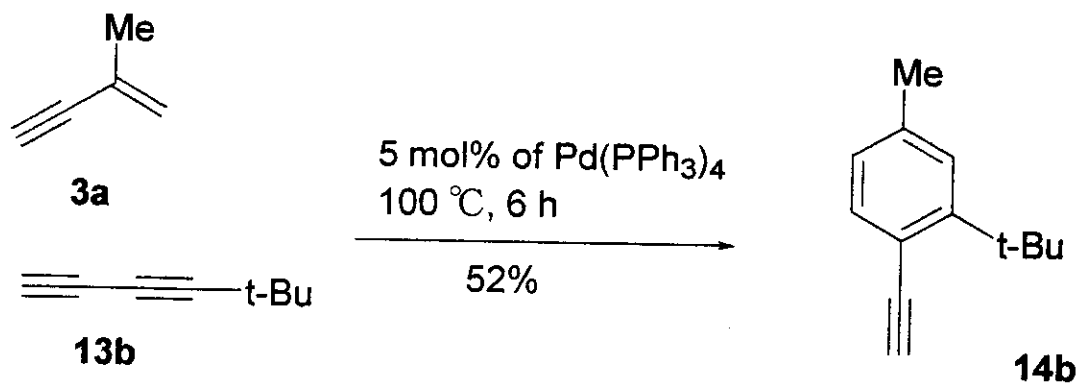
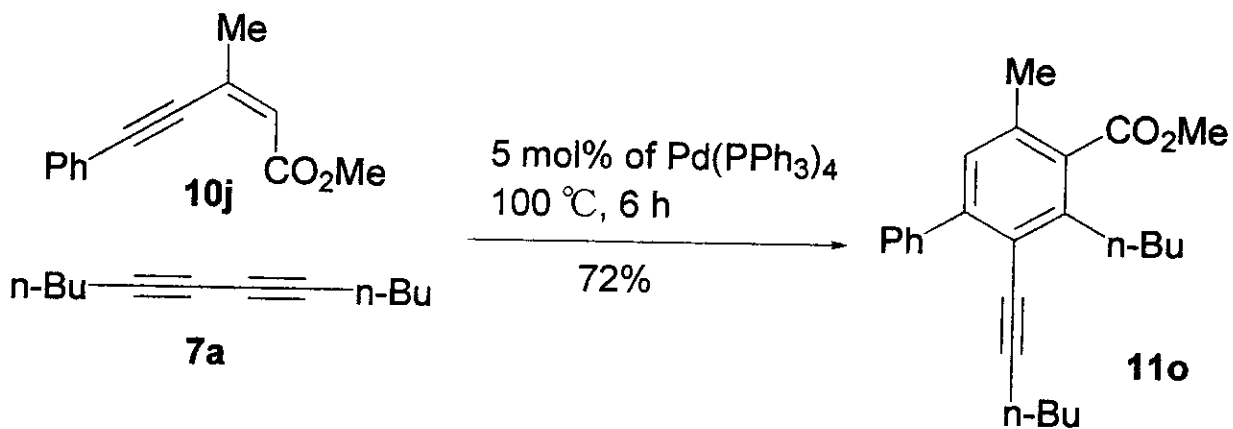
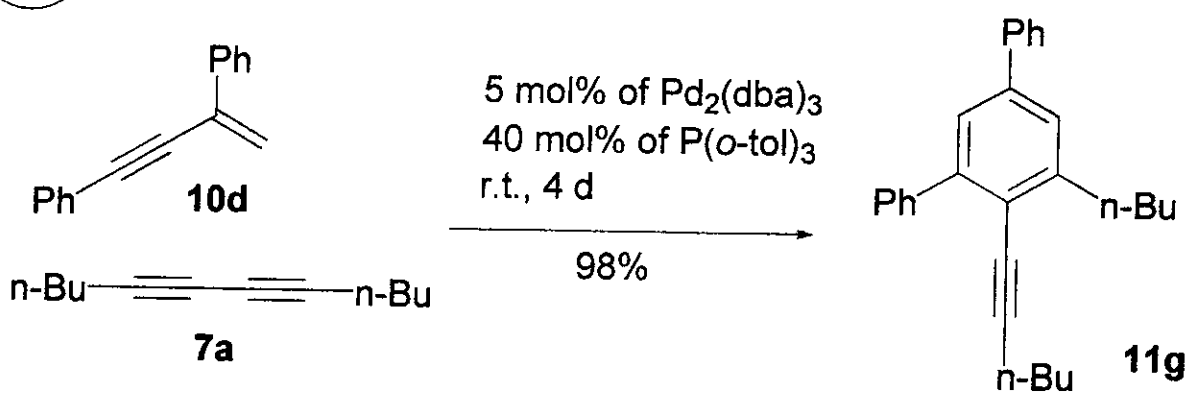
Yamamoto, Y. etc. *J. Am. Chem. Soc.* **1999**, *121*, 6391-6402

Pd



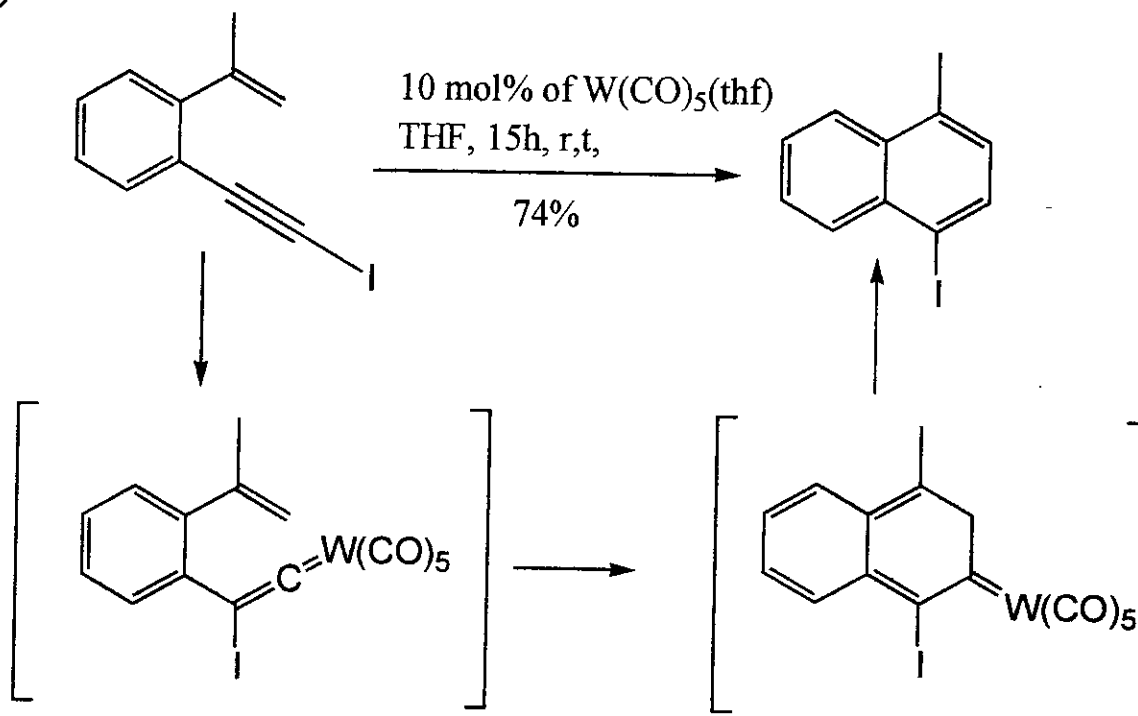
Pd

Transition metal-catalyzed benzannulation (1)



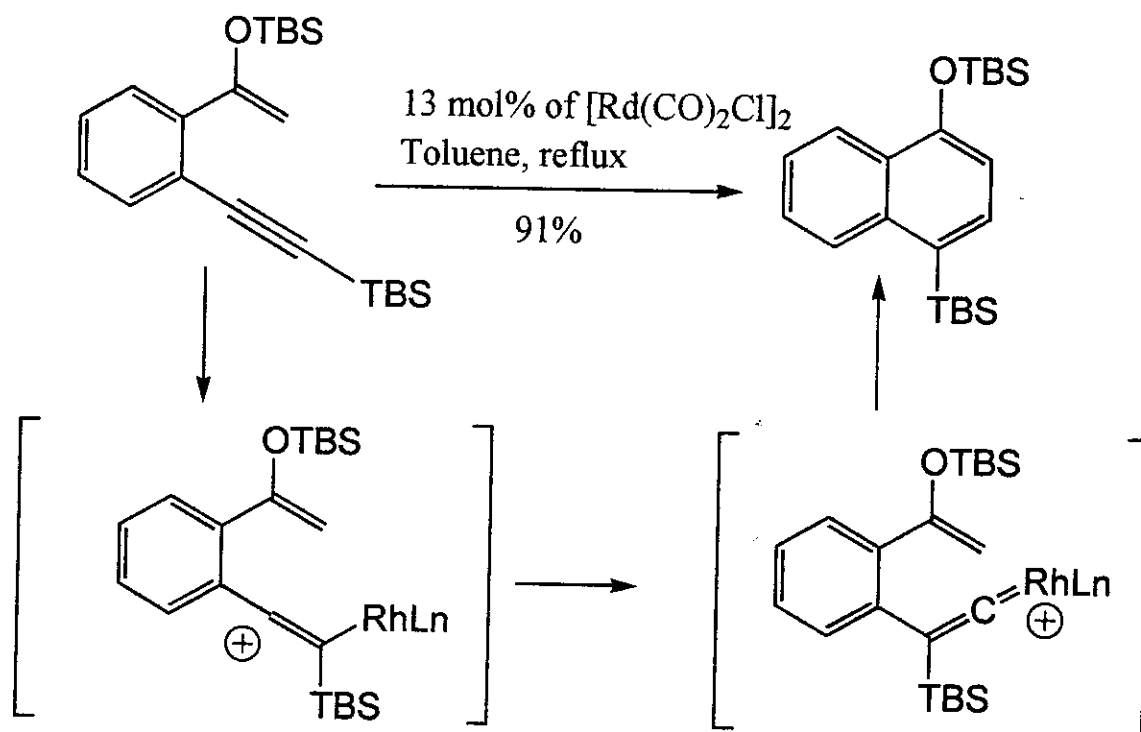
Transition metal-catalyzed benzannulation (2)

W Iwasawa, N. etc. *J. Am. Chem. Soc.* 2002, 124, 518-519



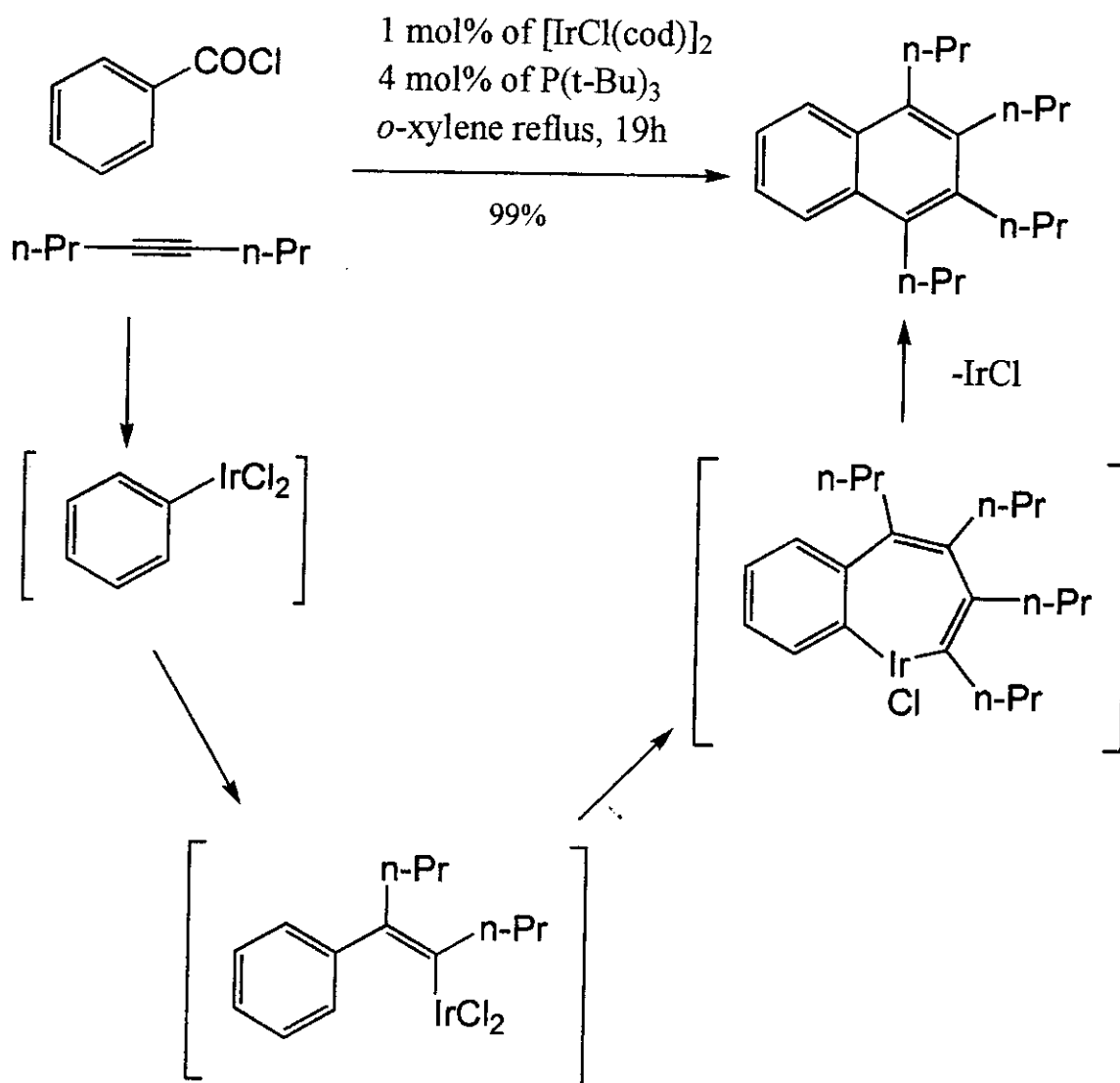
Transition metal-catalyzed benzannulation (3)

(Rh) Dankwardt, J. W. *Tetrahedron Lett.* 2001, 42, 5809-5812



Transition metal-catalyzed benzannulation (4)

Ir Miura, M. etc. *J. Am. Chem. Soc.* **2002**, *124*, 12680-12681



AuCl₃-Catalyzed Benzannulation: Synthesis of Naphthyl Ketone Derivatives from *o*-Alkynylbenzaldehydes with Alkynes

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 Department of Chemistry, Graduate School of Science, Tohoku University, Sendai 980-8578, Japan
 Received August 13, 2002

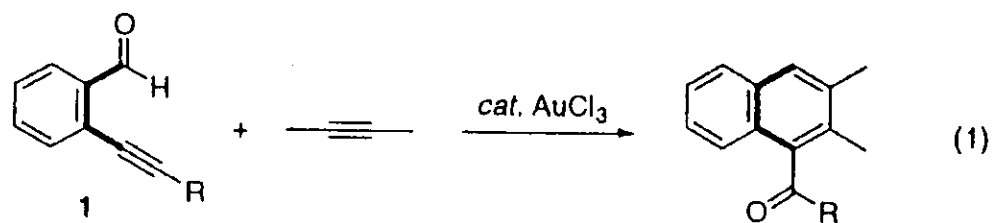
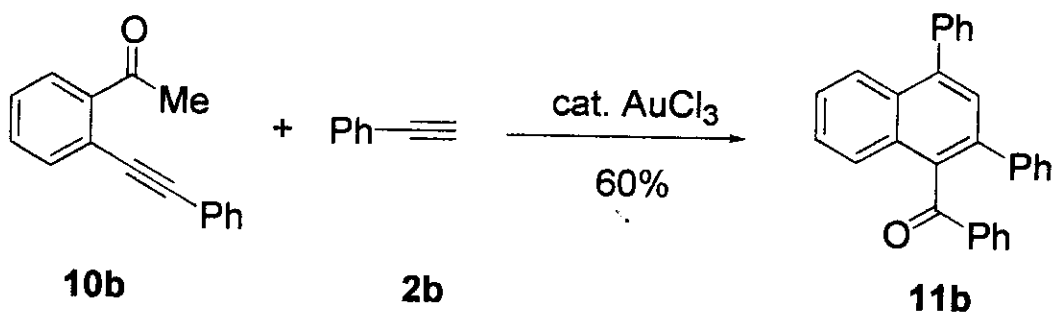
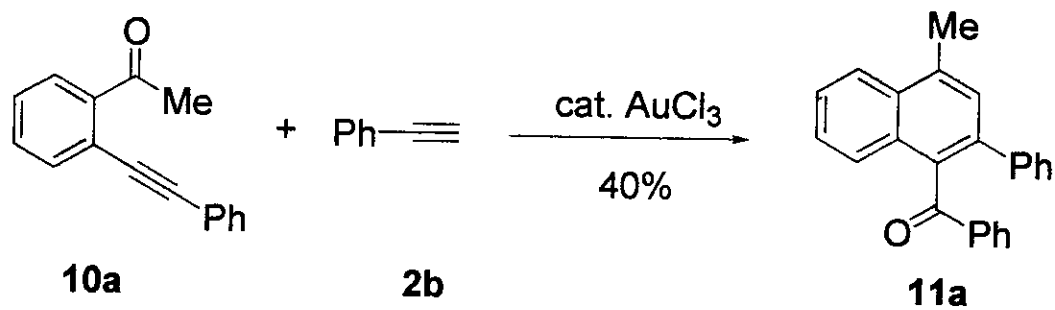
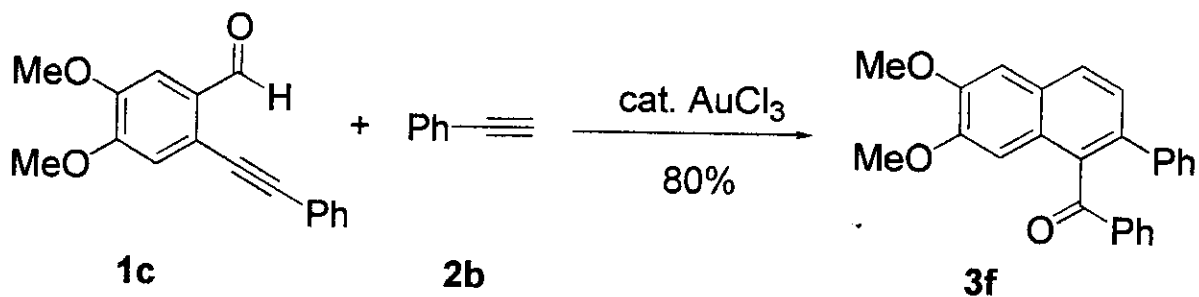


Table 1. The AuX₃-Catalyzed Reaction of *o*-Alkynylbenzaldehydes **1** with Alkynes **2**^a

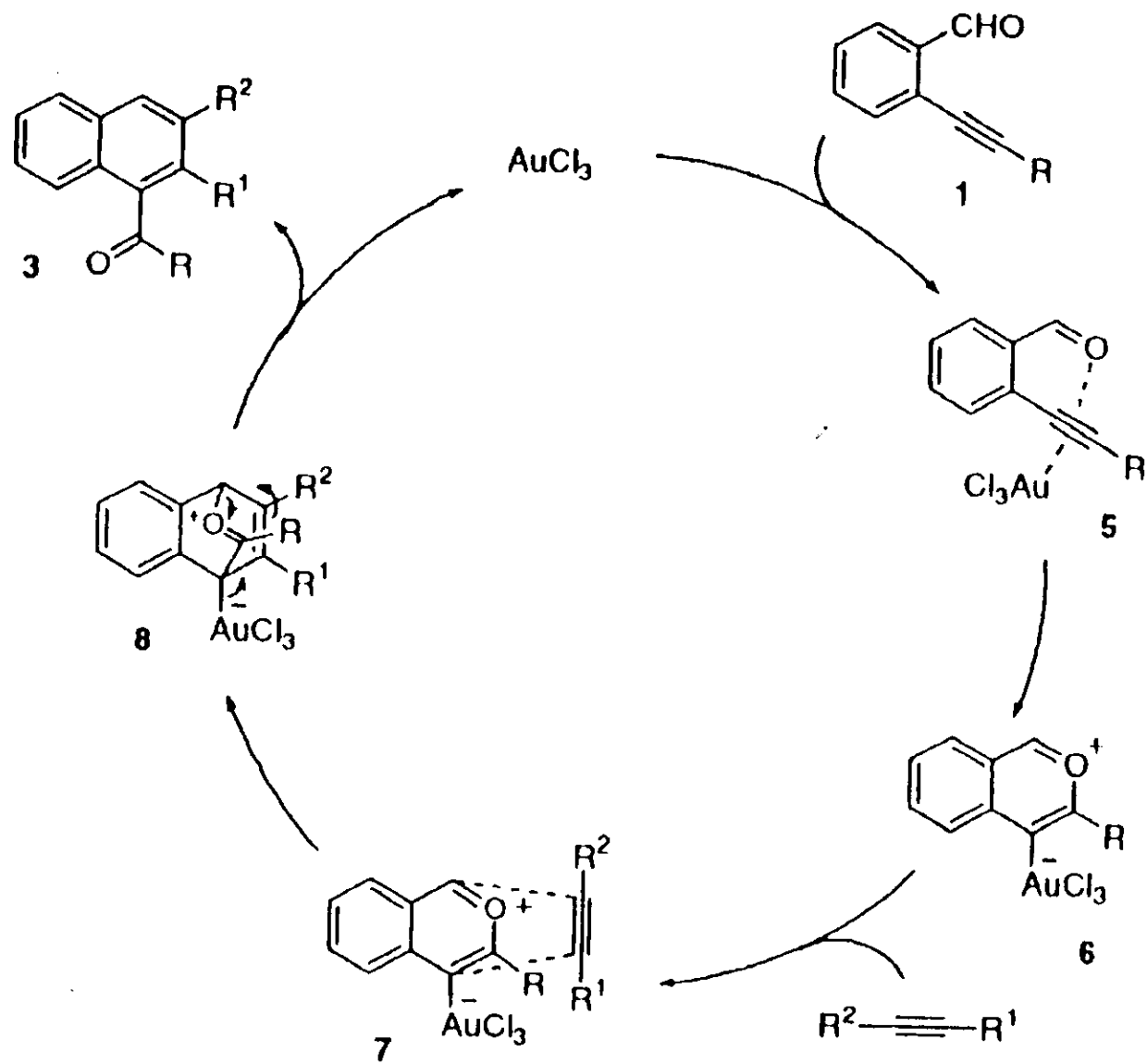
entry	1	R	2	R ¹ —C≡C—R ²		catalyst	time			yield (%) ^c
				R ¹	R ²			ratio ^b		
1 ^d	1a	Ph	2a	C ₃ H ₇	H	AuCl ₃	1 d	3a:4a = 95:5	57	
2	1a	Ph	2a	C ₃ H ₇	H	AuCl ₃	1.5 h	3a:4a = 92:8	91	
3 ^e	1a	Ph	2a	C ₃ H ₇	H	AuCl ₃	3 h	3a:4a = 89:11	72	
4	1a	Ph	2a	C ₃ H ₇	H	AuBr ₃	1.5 h	3a:4a = 93:7	100	
5 ^f	1a	Ph	2b	Ph	H	AuCl ₃	2.5 h	3b:4b = 99:<1	96	
6 ^f	1a	Ph	2b	Ph	H	AuBr ₃	0.7 h	3b:4b = 99:<1	100	
7	1a	Ph	2c	Me ₃ Si	H	AuCl ₃	6 h	3c:4c = 16:84	82	
8	1a	Ph	2d	CO ₂ Et	H	AuCl ₃	3 h	3d:4d = 18:82	72	
9	1a	Ph	2e	COCH ₃	H	AuCl ₃	3.5 h	3e:4e = <1:99	75	
10	1a	Ph	2f	C ₃ H ₇	C ₃ H ₇	AuCl ₃	2.5 h	3f(=4f)	52	
11	1a	Ph	2f	C ₃ H ₇	C ₃ H ₇	AuBr ₃	2.5 h	3f(=4f)	70	
12	1a	Ph	2g	Ph	Me ₃ Si	AuCl ₃	2 h	3g:4g = 99:<1	92	
13	1a	Ph	2h	Ph	Me	AuCl ₃	3 h	3h:4h = 99:<1	89	
14	1b	C ₆ H ₁₃	2b	Ph	H	AuCl ₃	1.5 h	3i:4i = 92:8	91	

^a The reaction was performed using *o*-alkynylbenzaldehydes **1** (1 equiv) and alkynes **2** (3 equiv) in the presence of AuX₃ (3 mol %) in (CICH₂)₂ at 80 °C unless otherwise noted. ^b Determined by ¹H NMR. ^c Combined isolated yield. ^d The reaction was carried out at 30 °C in CH₂Cl₂. ^e The reaction was carried out in the presence of 1 mol % of AuCl₃. ^f The reaction was carried out using 1.2 equiv of **2b**.

AuCl₃-Catalyzed Benzannulation: another examples

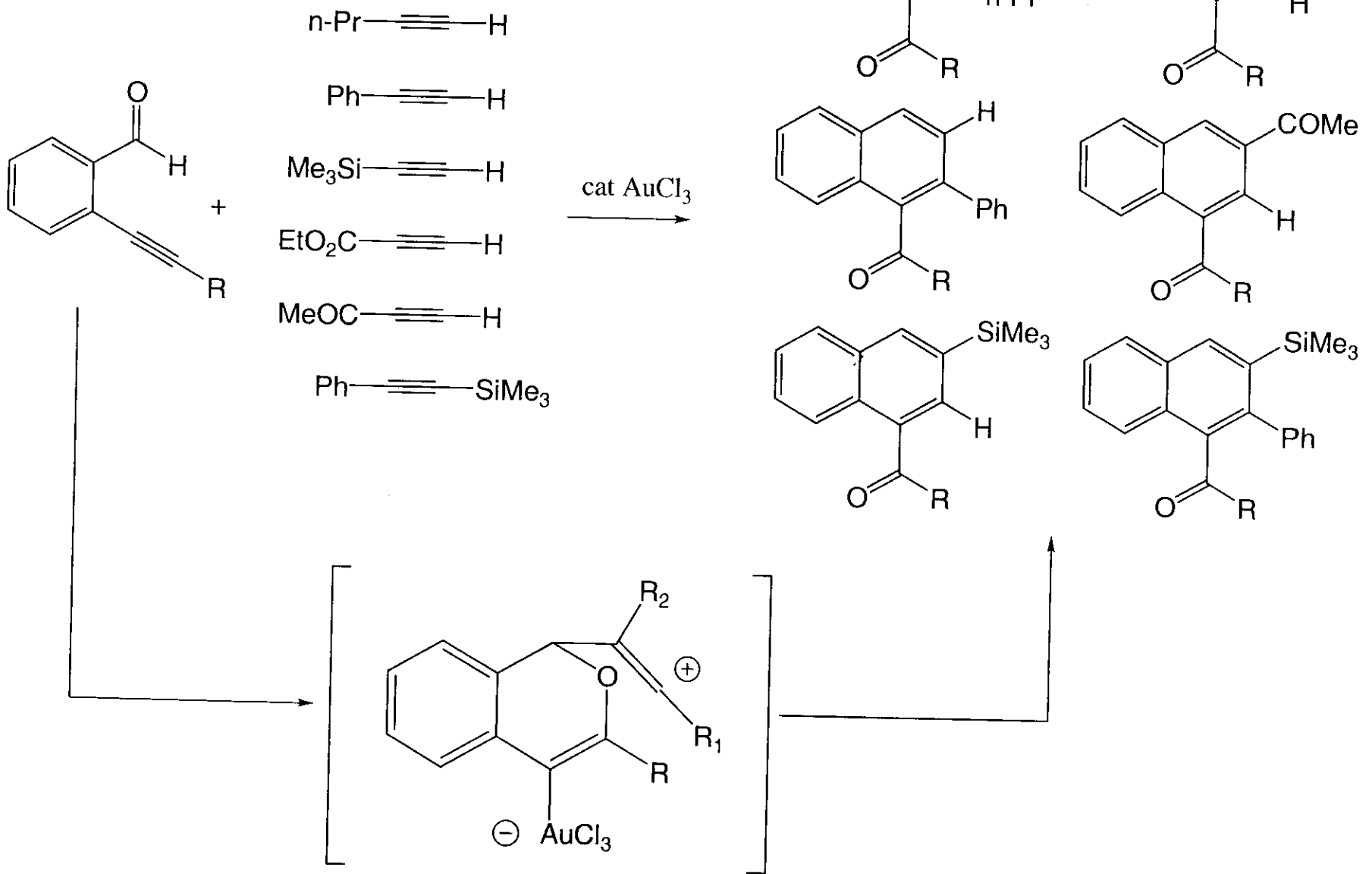


Reaction Mechanism



AuCl₃-Catalyzed Benzannulation

Regioselectivity



Cu(OTf)₂-Catalyzed Benzannulation

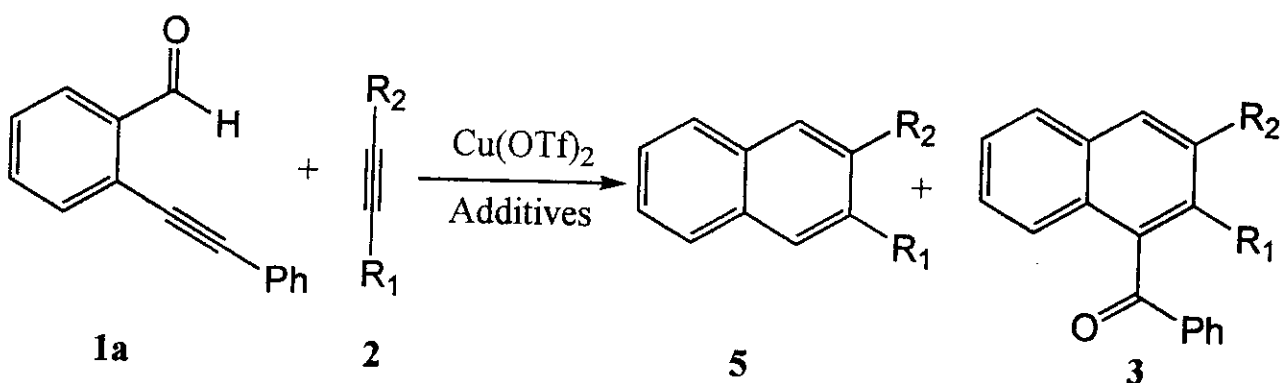
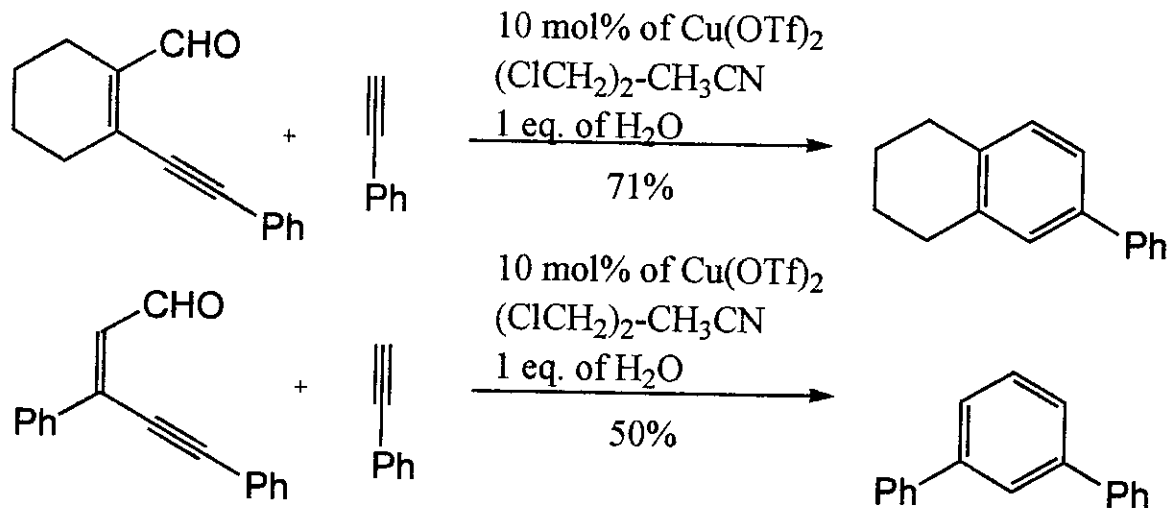


Table 2. The *Cu(OTf)₂*-Catalyzed Reaction of o-(Phenylethynyl)benzaldehyde **1a** with Alkynes **2^a**

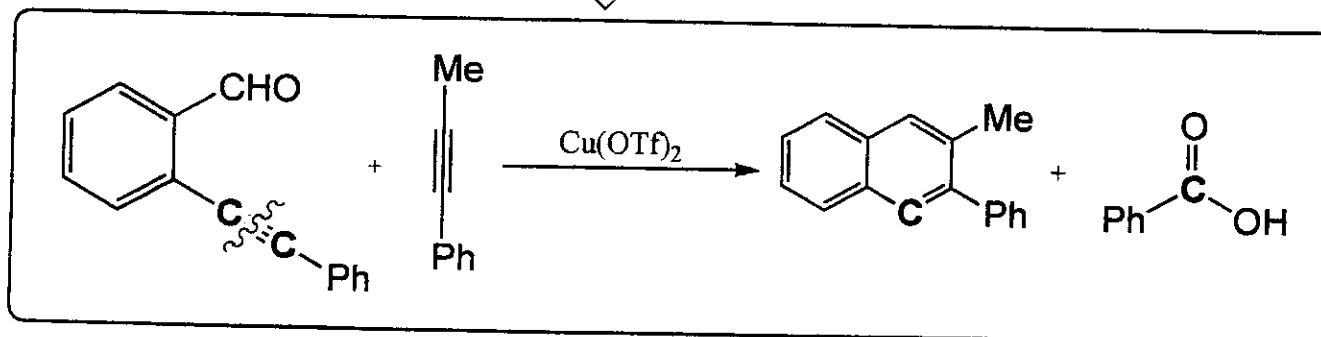
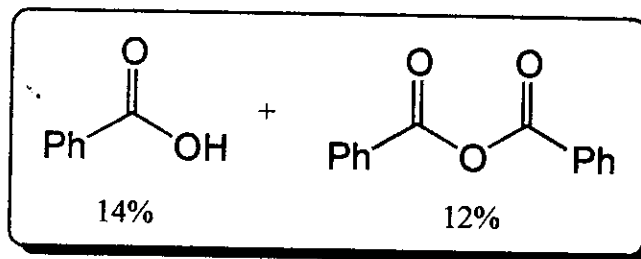
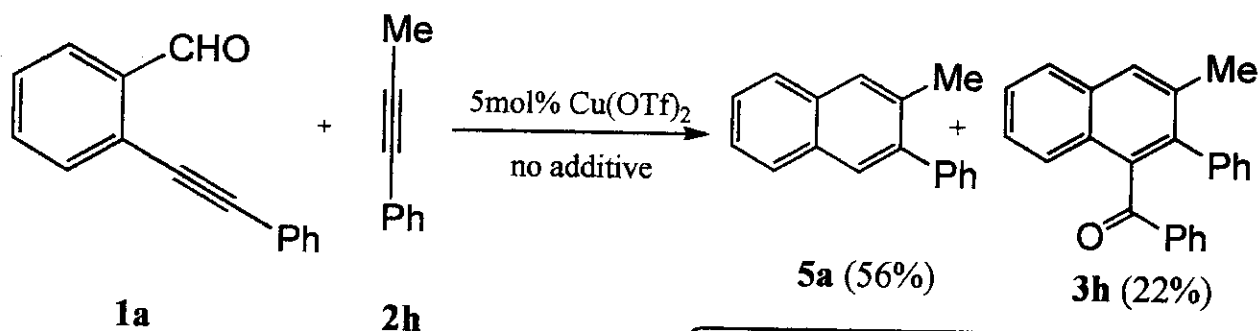
entry	2	R ¹	R ²	additive	conditions	yield (%) ^b	
						5	3
1	2h	Ph	Me	none	80 °C, 0.5 h	5a 56	3h 22
2	2h	Ph	Me	H ₂ O	80 °C, 19 h	5a 52	3h 12
3	2h	Ph	Me	MeOH	80 °C, 1 h	5a 44	3h 13
4	2h	Ph	Me	HCO ₂ H	80 °C, 0.5 h	5a 79	3h 8
5	2h	Ph	Me	CH ₃ CO ₂ H	80 °C, 0.5 h	5a 67	3h 17
6	2h	Ph	Me	CF ₂ HCO ₂ H	80 °C, 0.5 h	5a 82	3h trace
7	2h	Ph	Me	CF ₃ CO ₂ H	80 °C, 0.5 h	5a 78	3h trace
8 ^c	2h	Ph	Me	CF ₂ HCO ₂ H	80 °C, 15 h	5a 6	3h 0
9	2h	Ph	Me	CF ₂ HCO ₂ H	100 °C, 0.25 h	5a 86	3h 0
10	2b	Ph	H	CF ₂ HCO ₂ H	100 °C, 0.25 h	5b 90	3b 0
11	2i	C ₄ H ₉	H	CF ₃ CO ₂ H	100 °C, 0.25 h	5c 72	3j 8
12	2j	Ph	Ph	CF ₂ HCO ₂ H	100 °C, 0.25 h	5d 85	3k 0
13	2f	C ₃ H ₇	C ₃ H ₇	CF ₂ HCO ₂ H	100 °C, 0.25 h	5e 74	3f 0
14	2k	Ph	Br	CF ₂ HCO ₂ H	100 °C, 0.25 h	5f 73	3l 0
15	2l	Ph	PhS	CF ₂ HCO ₂ H	100 °C, 0.25 h	5g 60	3m 0
16	2d	CO ₂ Et	H	CF ₃ CO ₂ H	100 °C, 0.25 h	5h 60	3d 0 ^d

^a The reaction was carried out using **1a** (1 equiv) and **2** (1.2 equiv) in the presence of *Cu(OTf)₂* (5 mol %) and additives (1 equiv) in (CICH₂)₂ unless otherwise noted. ^b Isolated yield. ^c The reaction was performed in the presence of 10 mol % of TfOH instead of *Cu(OTf)₂*. ^d **4d** was obtained in 14% yield.

Cu(OTf)₂-Catalyzed Benzannulation: another examples



Entry 1



Reaction Mechanism

