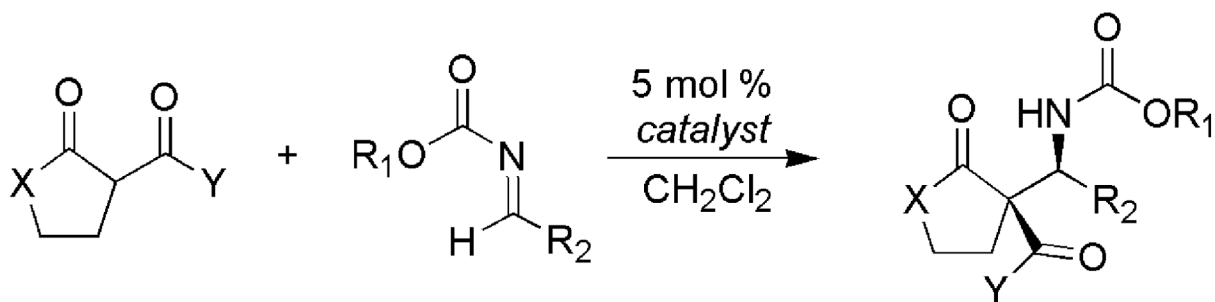


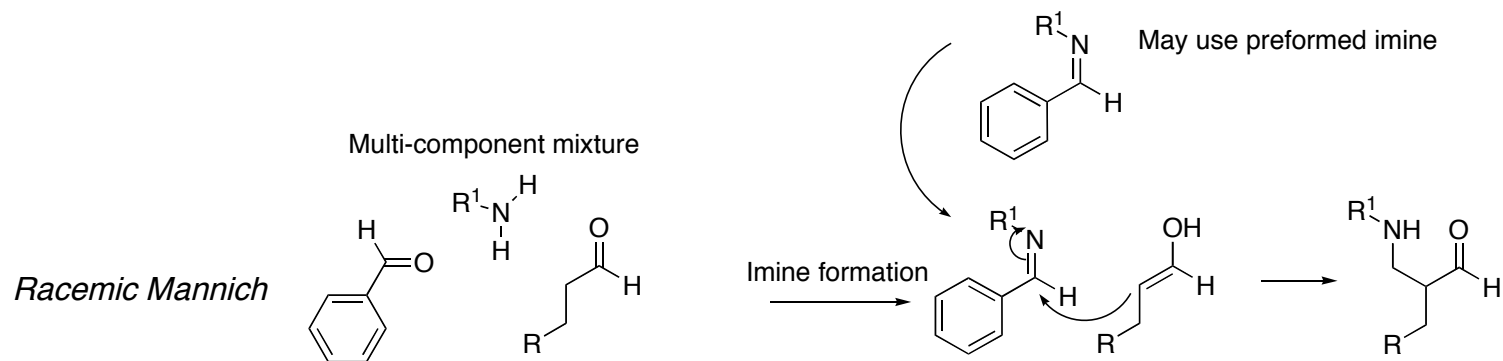
Highly Diastereoselective Asymmetric Mannich Reactions of 1,3-Dicarbonyls with Acyl Imines



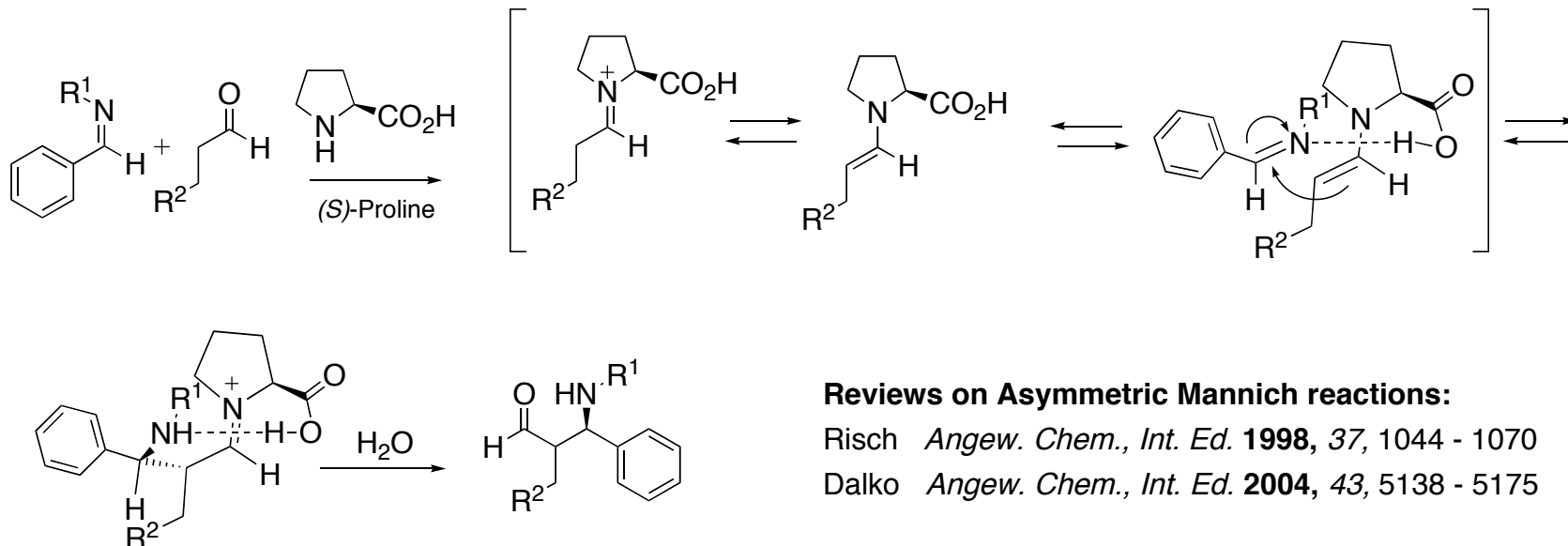
Amal Ting, Sha Lou, and Scott E. Schaus
Org. Lett., **2006**, *8*, 2003-2006

Presented by:
John Maciejewski
July 1, 2006

General Mannich Reaction



Asymmetric Mannich



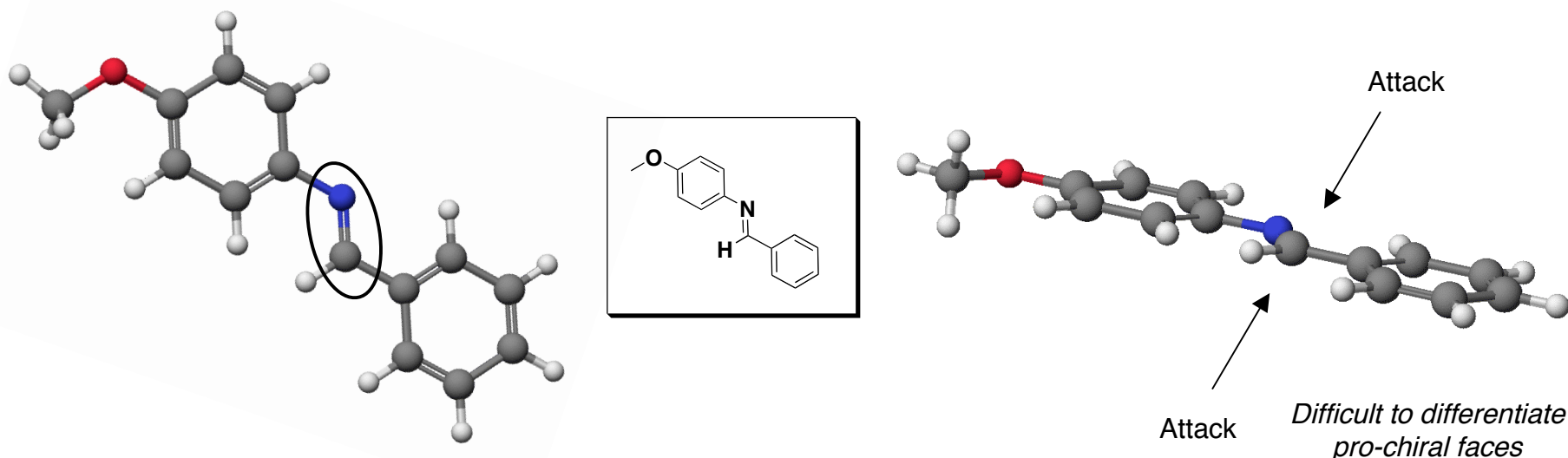
Modest diastereoselectivity with good enantioselectivities

Reviews on Asymmetric Mannich reactions:

Risch *Angew. Chem., Int. Ed.* **1998**, *37*, 1044 - 1070

Dalko *Angew. Chem., Int. Ed.* **2004**, *43*, 5138 - 5175

Asymmetric Nucleophilic Addition to Imines



- Selective addition to pro-chiral face of imine continues to be studied extensively
- Chiral Proline derivatives have been used to invoke enantioselective addition to imines
- Other nucleophiles (ex. β -keto esters, β -diketones) require source of chirality for asymmetric addition to imines

Importance of asymmetric Mannich Reactions

- Ability to form quaternary carbon stereocenters
- Form valuable enantioenriched synthons
- Organocatalytic

Selected organocatalytic examples:

(S)-Proline - Parrish *J. Org. Chem.*, **1974**, *39*, 1615 - 1621

Modified prolines - Barbas *J. Am. Chem. Soc.*, **2006**, *128*, 1040-1041

Cinchona alkaloids - Schaus *J. Am. Chem. Soc.*, **2005**, *127*, 11256-11257

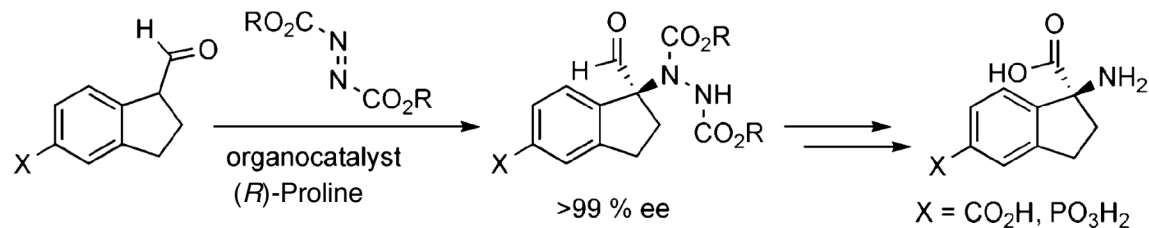
Organometallic Catalytic

Shibasaki - *Tet. Lett.* **1999**, *40*, 307-310

Mannich Reactions

Applications of asymmetric synthesis

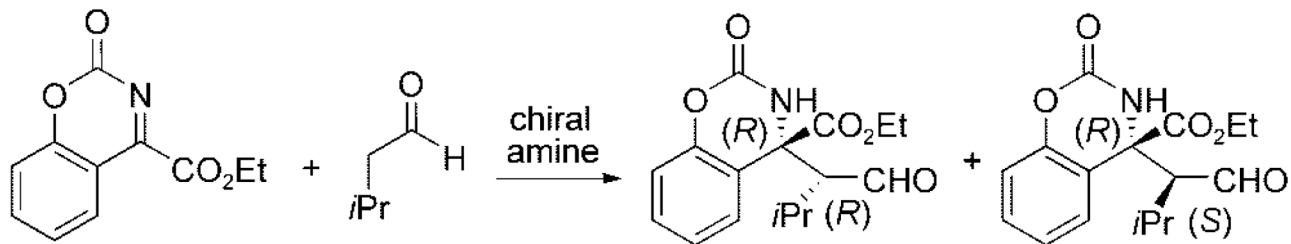
α -Amination



Barbas *Org. Lett.*, **2005**, 7, 3885-3888

Compounds used to treat neurodegenerative diseases

Organocatalysis on ketimines

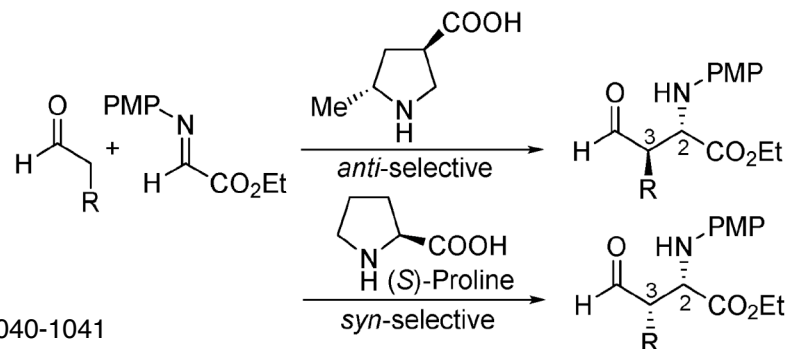


Jørgensen *Angew. Chem., Int. Ed.*, **2004**, 43, 4476-4478

Used L-proline -derived catalysts to obtain selectivity

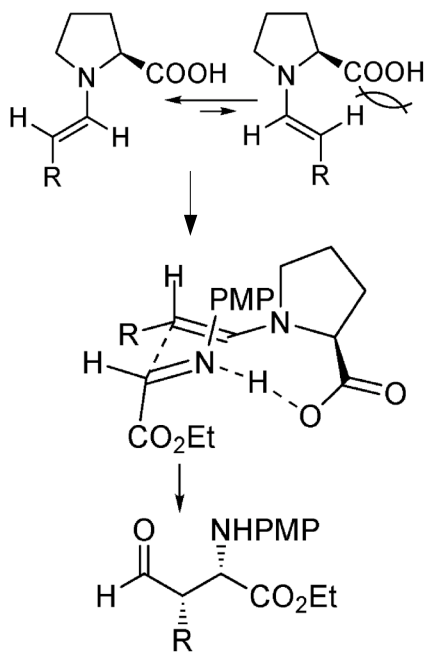
Mannich Reactions

Applications of asymmetric synthesis

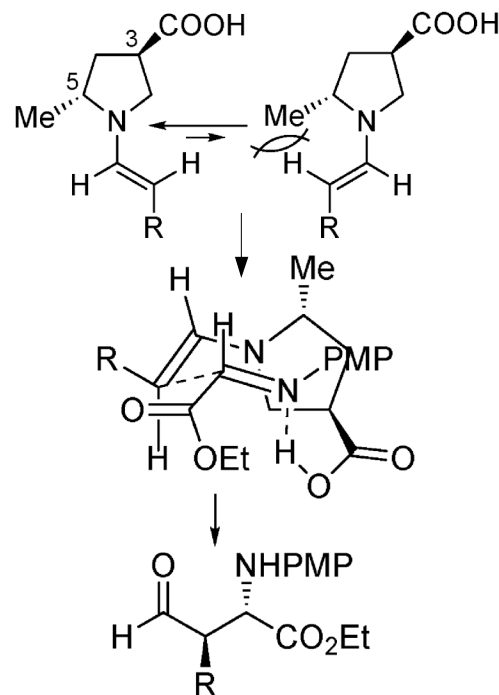


Barbas *J. Am. Chem. Soc.*, **2006**, *128*, 1040-1041

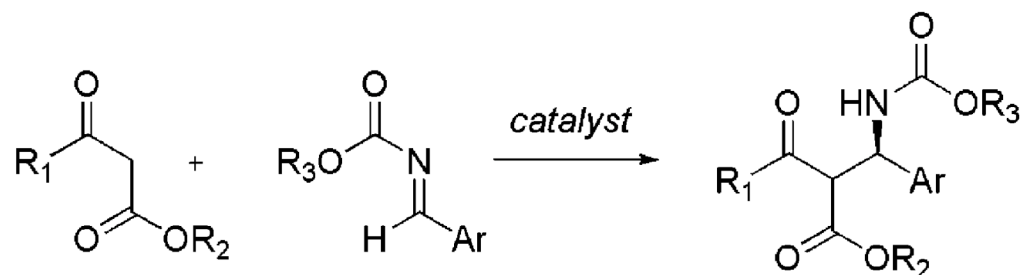
Syn-pathway



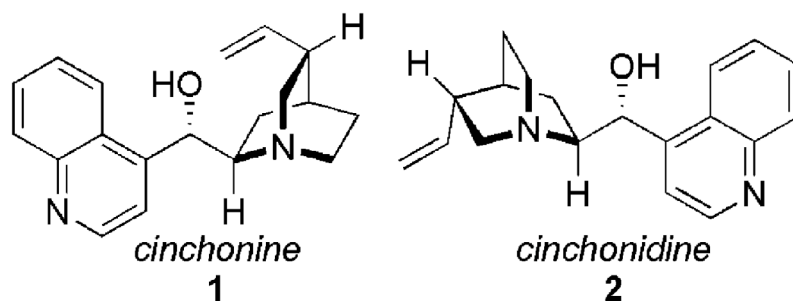
Anti-pathway



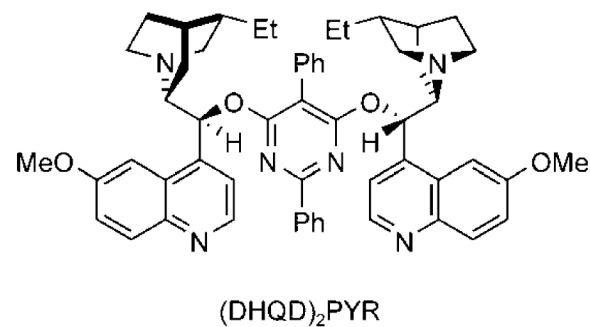
Mannich Reaction Using Chiral Lewis Bases



Modest diastereoselectivity with good enantioselectivities



Schaus *J. Am. Chem. Soc.*, **2005**, *127*, 11256-11257

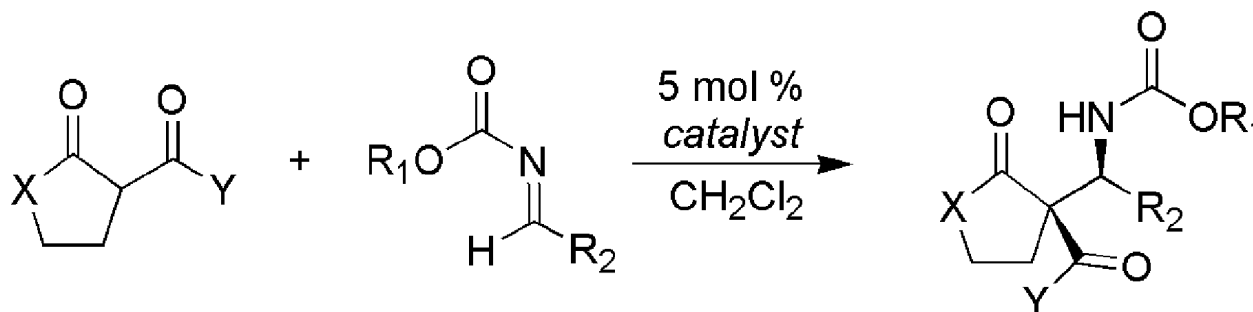


Poulsen *Angew. Chem., Int. Ed.* **2005**, *44*, 2896 - 2899

Used modified cinchona alkaloid as organocatalyst in Mannich reaction

Mannich Reactions Using Cinchona Alkaloids

Improved selectivity in cyclic systems



X = CH₂, Y = OCH₃

X = CH₂, Y = OCH₂CH₃

X = CH₂, Y = CH₃

X = O, Y = CH₃

R₁ = CH₃, CH₂CH=CH₂

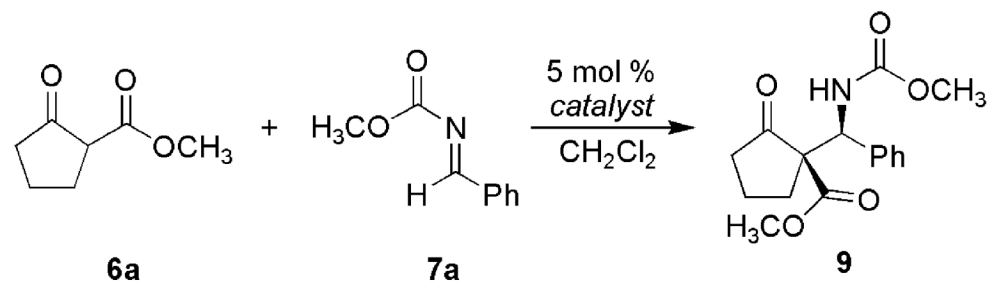
R₂ = Ar, (*E*)-CH=CH₂Ar

90-98% yield
90-99% de
90-99% ee

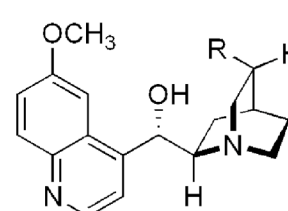
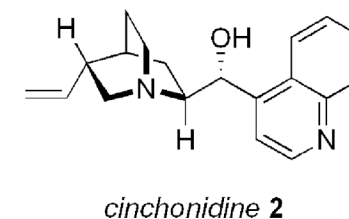
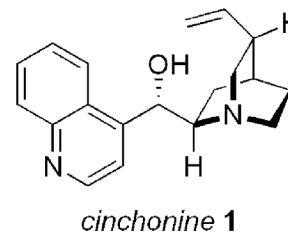
Expanded scope to cyclic 1,3-dicarbonyl compounds

- Improved isolated yields
- Observed improved diastereo- and enantioselectivity
- Demonstrates potential for applications in organic synthesis

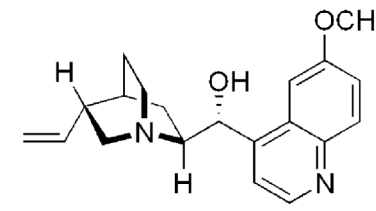
Screening of Chiral Lewis Bases



entry	catalyst	yield (%) ^b	de (%) ^c	ee (%) ^c
1	cinchonine 1	96	93	90
2	cinchonidine 2	96	94	-88
3	quinidine 3	96	95	18
4	dihydroquinidine 4	94	94	88
5	quinine 5	95	94	-10



R = CH_2CH_3 , dihydroquinidine **4**

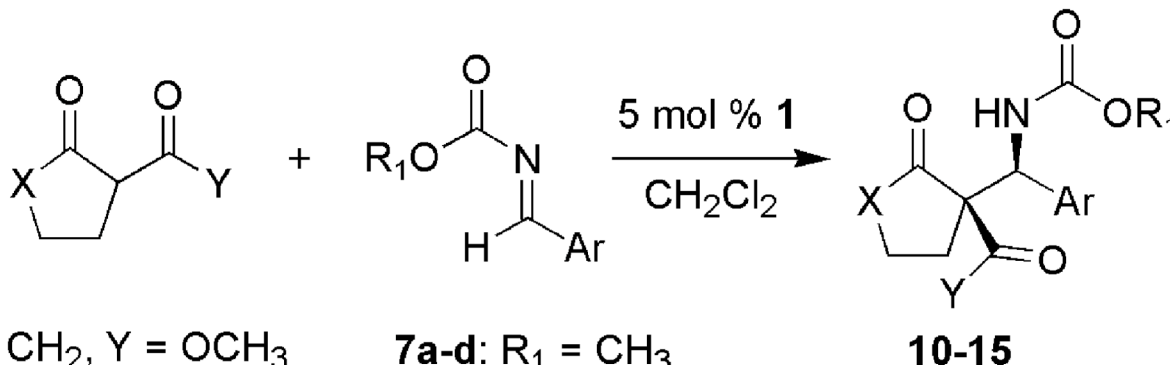


Cinchona alkaloids were screened in asymmetric Mannich reaction

- Afford products in high yield and in high diastereo- and enantiomeric excess

Substrate Scope Using Cinchonine (1)

Addition of β -keto esters, β -keto lactones, and β -diketones to benzylidene carbamates



6a: X = CH_2 , Y = OCH_3 **7a-d:** R_1 = CH_3
6b: X = CH_2 , Y = OCH_2CH_3 **8a-b:** R_1 = allyl
6c: X = CH_2 , Y = CH_3
6d: X = O, Y = CH_3

Cinchonine was established to be an effective catalyst for asymmetric synthesis

Yield (%) range

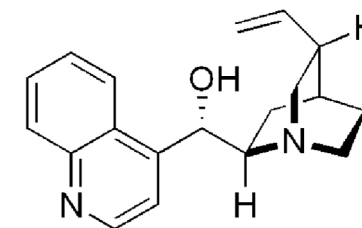
78 - 98

de (%) range

38 - 99

ee (%) range

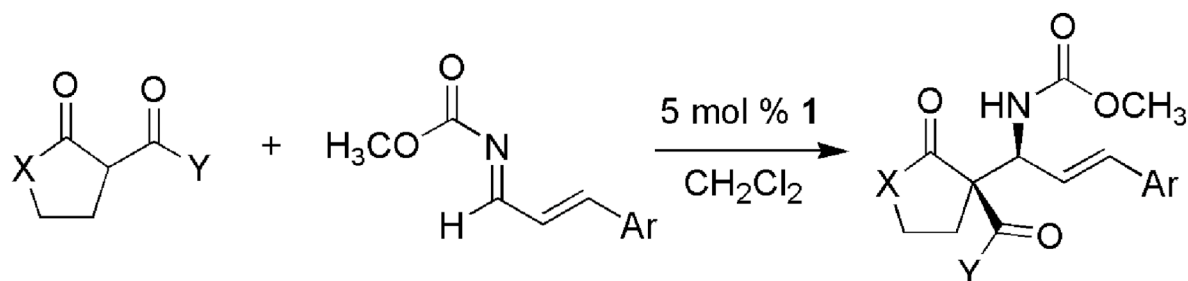
90 - 99



cinchonine **1**

Substrate Scope Using Cinchonine (1)

Addition of β -keto esters, β -keto lactones, and β -diketones to aryl-propenyl acyl imines



6a: X = CH_2 , Y = OCH_3

17a-b

20, 21

6b: X = CH_2 , Y = OCH_2CH_3

6c: X = CH_2 , Y = CH_3

6d: X = O, Y = CH_3

Yield (%) range

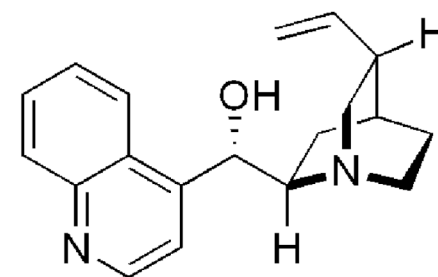
88 - 98

de (%) range

38 - 99

ee (%) range

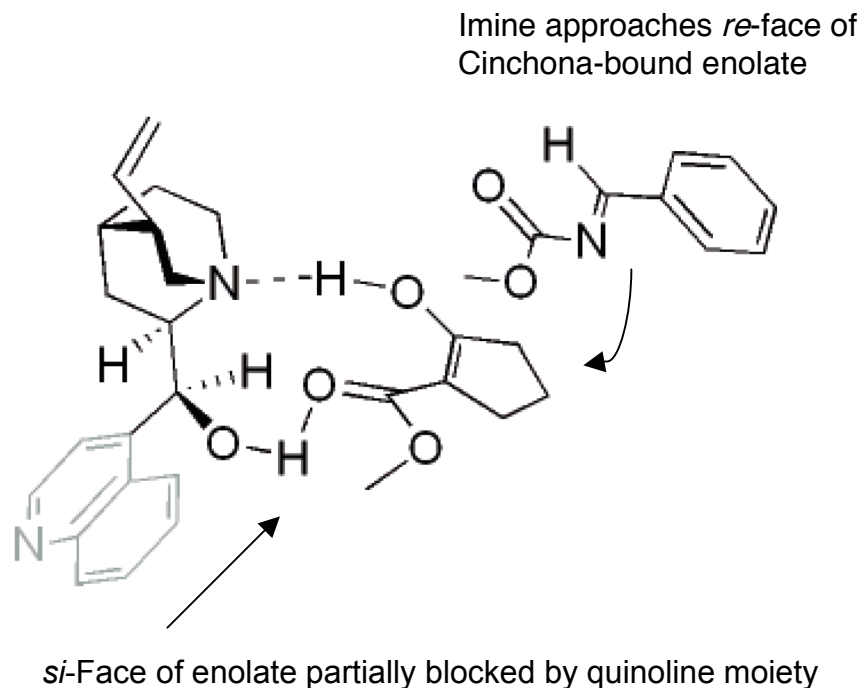
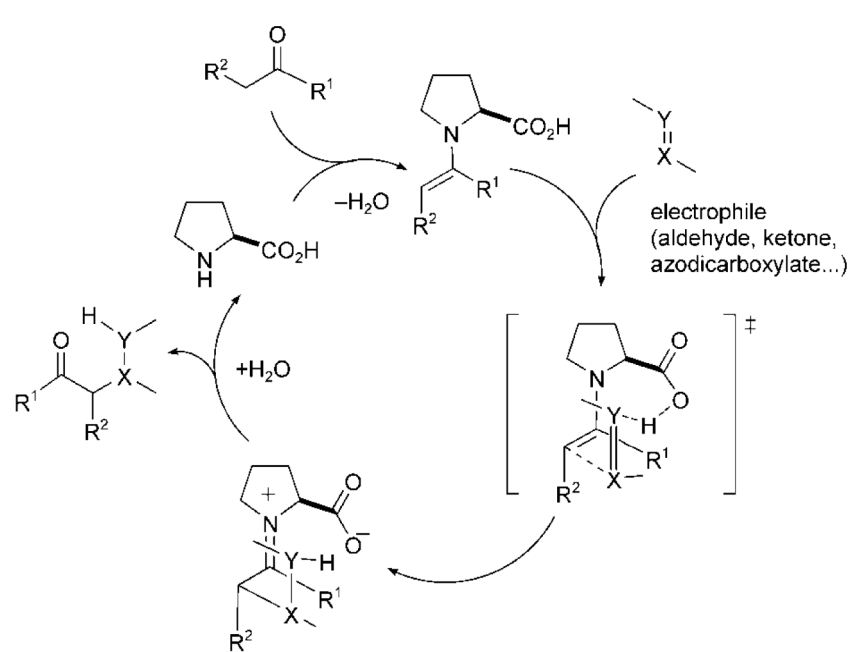
93 - 99



cinchonine 1

Transition States

Using molecular modeling* to probe origin of selectivity in Cinchona-catalyzed Mannich reactions



P. Dalko *Angew. Chem., Int. Ed.*, **2004**, 43, 5138 - 5175

*Calculations performed using MMFF (Merck Molecular Force Field) to determine lowest-energy conformer of enol (above).

Summary & Next Directions

- *Cinchona alkaloids catalyze Mannich reactions to form highly diastereo- and enantioenriched products*
- *Using organocatalysts to facilitate highly diastereo- and enantioselective addition will continue to make this versatile reaction attractive*
- *Apply this methodology towards six-membered ring-containing β -keto esters and β -diketones*

