CRTH2 Antagonist MK-7246: A Synthetic Evolution from Discovery through Development

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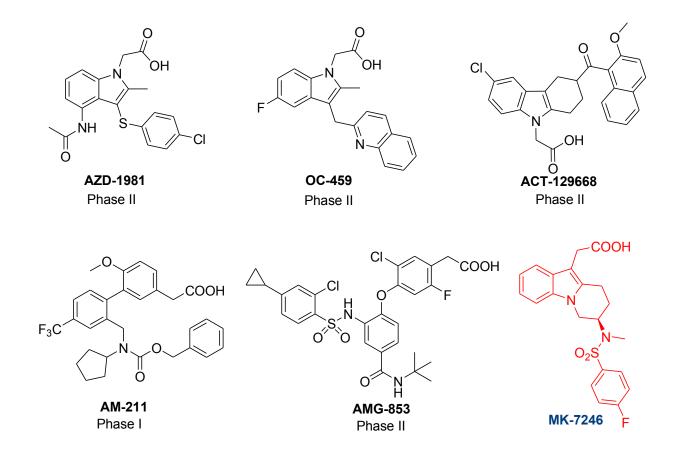
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Feng Zhang
Current Literature
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Introduction and Background:

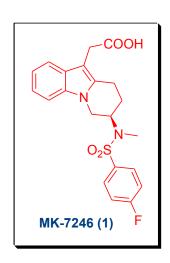
- CRTH2 (DP2) is one of the two high-affinity transmembrane receptors for prostaglandin d2 (PGD2).
- PGD2 has been implicated as a mediator of allergic inflammation and diseases including asthma, allergic rhinitis, and atopic dermatitis.
- The first PGD2 receptor characterized, DP, is known mainly for its role vasodilatation.
- By contrast the more recently identified CRTH2 has been found to play a role in leukocyte activation.
- Biological and genetic data suggest that the downstream events triggered by activation of CRTH2 through binding to PGD2 play a key role in stimulating latephase allergic inflammation.
- The hypothesis that blockade of the CRTH2 receptor could provide a novel mechanism for treatment of chronic allergic disease.
- Many companies have engaged in CRTH2 antagonist programs which have reached various stages of preclinical and clinical development.

Some of DP₂ receptor antagonists in development



Norman, P. Expert Opin. Invest. Drugs **2010**, 19, 947–961.

Abstract



·CO₂R **1st Synthesis Evolution** OTBS 1) aziridine ring-opening 2) Friedel-Crafts cycliation 16 O_2S_2 Ar=4F-C₆H₄ 15

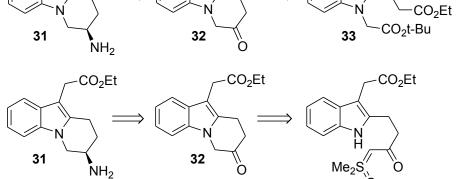
CO₂Et

2nd Synthesis Evolution 1) Transamination

2) Dieckmann Cyclization

Manufacturing Route

- 1) Transamination
 2) Ir catalyzed N-H Insertion



-CO₂Et

-CO₂Et

Original Synthesis of MK-7246 (1) and Analogues (13)

2 3 4:
$$R \neq H$$
[5: $R = H$]

O₂S

Ar

 R_1
 R_2
 R_1
 R_2
 R_1
 R_3
 R_4
 R_5
 R_1
 R_4
 R_5
 R_1
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 $R_$

Reagents: (a) 1. LiAlH₄; 2. MnO₂; (b) 1. Ph₃PCHCO₂Et; 2. BrCH₂CO₂t-Bu, Cs₂CO₃; 3. H₂, Pd/C; (c) KOt-Bu; (d) 1. silica gel, toluene, reflux; 2. NaBH₄; (e) 1. MsCl, Et₃N; 2. NaN₃; (f) 1. H₂, Pd/C; 2. ArSO₂Cl, Et₃N; 3. NaH, Mel; (g) 1. (COCl)₂; 2. MeOH; (h) 1. NaBH₄; 2. Et₃SiH, TFA; (i) 1. chiral HPLC; 2. LiOH, THF

Drawbacks

- the synthesis was linear with the indole core being incorporated in the first step, limiting the possibilities for subsequent diversification.
- synthetic routes suffered from tedious separation and lower overall yield. (at 18 steps, ~10% overall yield, including a late-stage chiral separation)
- long time, the synthesis time per analogue was about 3–4 weeks.

Retrosynthetic Analysis: Aziridine Ring-Opening Reaction as the Key Coupling Strategy

Preparation of Aziridine 16

Reagents: (a) SOCl₂, MeOH, 99%; (b) ArSO₂Cl, Et₃N, THF, 93%; (c) NaBH₄ (5 eq), EtOH, 60%; (d) ADDP, n-Bu₃P, THF, 77%; (e) TBSCl, Et₃N, THF, 84%.

Synthesis of MK-7246 (1) Using the Aziridine Ring-Opening Strategy

Reagents: (a) MeOH, cat H_2SO_4 , 81%; (b) 1. KHMDS, DMF; 2. **16**; (c) MeI; (d) HCI, 63% (69 LCAP) from **16**; (e) reslurry in EtOAc/MTBE/*n*-heptane, 96% (83 LCAP); (f) (COCl₂)₂, DMSO, Et₃N, CH₂Cl₂; PPTS, toluene, 60 °C, 80% from **26**; (h) H₂, Pd/C, EtOAc, 91%; (i) LiOH, THF, 97%.

Advantages

- set the stereocenter on the molecules eliminating chiral separations
- overall 11 steps, 15% yield from D-aspartic acid
- less than 1 week processing time on laboratory scale
- with the potential for structural diversification at many sites
- scalable to the extent that 1–10 kg quantities of MK-7246 (1) could be prepared, enough to supply the early phase I clinical trials and animal toxicity studies

Drawbacks

 moderate yields, processing difficulties for several steps, and instability of certain intermediates combined to render this route impractical for the production of the significantly larger quantities (>10 kg) of MK-7246 (1) needed for late-stage development into phase II and beyond.

Retrosynthetic Analysis: An Enzymatic Reaction Installs the Amine-Bearing Stereocenter in a Single Step

$$\begin{array}{c} \text{CO}_2\text{H} \\ \text{30} \\ \text{N}_{\text{R}_2} \\ \text{N}_{\text{R}_2} \\ \text{CO}_2\text{Et} \\ \text{Dieckmann} \\ \text{32} \\ \text{O} \\ \end{array}$$

Preparation of Transaminase Ketone Precursor 32: Dieckmann Cyclization Route

Reagents: (a) PhNHNH₂·HCl, toluene, reflux, 87%; (b) H₂SO₄, EtOH, reflux, 75%; (c) BrCH₂CO₂t-Bu, Cs₂CO₃, DMF, 93%; (d) KOt-Bu, THF, 64%; (e) silica gel, toluene, reflux, 88%; (e) CDX -017 (4.3 wt%), PLP, i-PrNH₂, PH 8.5 Et₃N buffer, aq DMSO, 45 °C, 80 - 81% assay yield of 31, 98 -99% ee.

[40: R = H]

Advantages

 This amine was highly valuable for our Medicinal Chemistry colleagues since it allowed subsequent SAR of positions R1 and R2 of compound 30 in only two steps.

Drawbacks

- the relatively high number of steps
- the paucity of crystalline intermediates and need for chromatographic purification after each step
- the lack of regioselectivity in the cyclization reaction.

Revised Retrosynthetic Analysis: N-H Carbenoid Insertion Approach to Tricyclic Ketone 32

Manufacturing Route to MK-7246 (1)

EtO₂C
$$\xrightarrow{34}$$
 $\xrightarrow{CO_2Et}$ \xrightarrow{B} $\xrightarrow{CO_2Et}$ \xrightarrow{B} $\xrightarrow{CO_2Et}$ $\xrightarrow{CO_2Et}$ $\xrightarrow{CO_2Et}$ $\xrightarrow{CO_2Et}$ \xrightarrow{Ar} $\xrightarrow{CO_2Et}$ \xrightarrow{Ar} $\xrightarrow{$

Reagents: (a) PhNHNH₂•HCl, ZnCL₂, toluene, 105°C; (b) Me₃SOl, KO*t*-Bu, THF, 77% from **34**; (c) [IrCl(COD)]₂ (1 mol%), toluene/DMF, 85°C, 83%; (d) CDX -017 (4.3 wt%), PLP, *i*-PrNH₂, PH 8.5 Et₃N buffer, aq DMSO, 45°C, 81% AY, 98 -99% ee; (e) HCl, IPA/CPME, 76% from **32**; (f) ArSO₂Cl, aq Na₂CO₃, IPAc, 89%; (g) Mel, K₂CO₃, DMF, 96%; (h) aq LiOH, THF/EtOH, 99%.

Conclusion

- the route proceeded in eight steps and 49% overall yield from commercially available starting materials
- required no chromatographic purifications
- amenable to pilot-plant scale production (>100 kg)

Highlights

- an Ir-catalyzed intramolecular N–H insertion of sulfoxonium ylide 41
- conversion of ketone **32** to amine **31** in a single step with excellent enantioselectivity through a transaminase process
- Reactions such as these illustrate the enabling impact and efficiency gains that innovative developments in chemo- and biocatalysis can have on the synthesis of pharmaceutically relevant target molecules