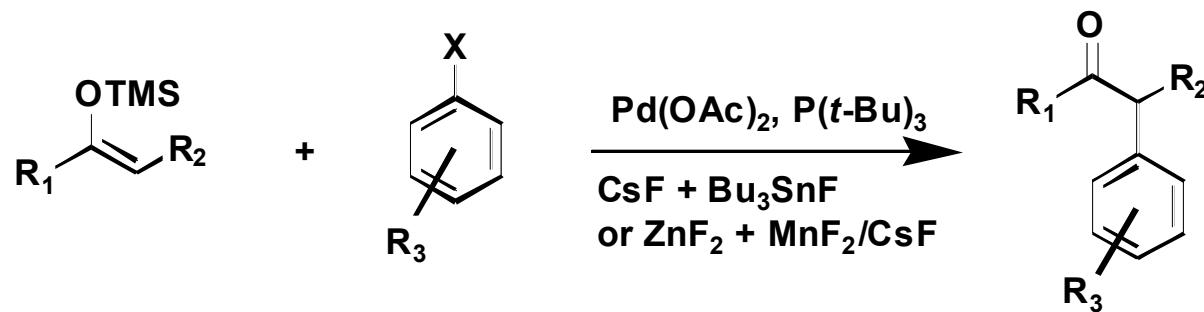


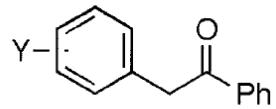
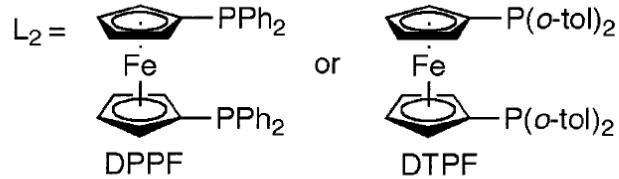
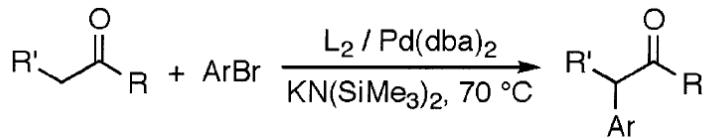
Pd-Catalyzed α -Arylation of Trimethylsilyl Enol Ethers with Aryl Bromides and Chlorides



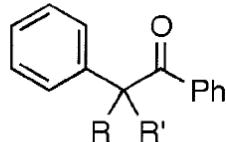
Weiping Su, Steven Raders, John G. Verkade,
Xuebin Liao, and John F. Hartwig
Angew. Chem. Int. Ed. ASAP

Current Literature
Chenbo Wang @ Wipf Group
August 12th, 2006

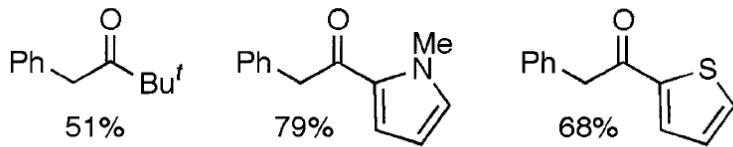
Pd-catalyzed Arylation of Enolates: Ketone Initial Works



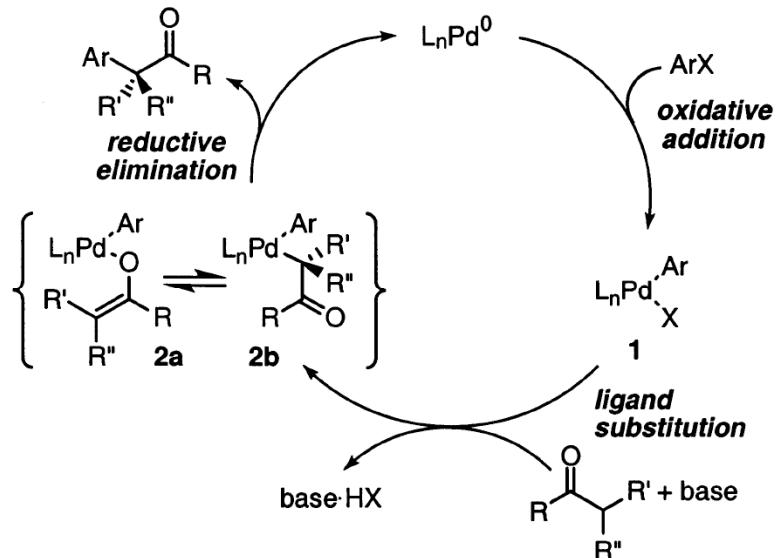
Y=4-t-Bu 85%
 Y=4-OMe 69%
 Y=3-CN 73%
 Y=2-Me 94%



R=H, R'=Me 71%
 R,R'=Me 55%

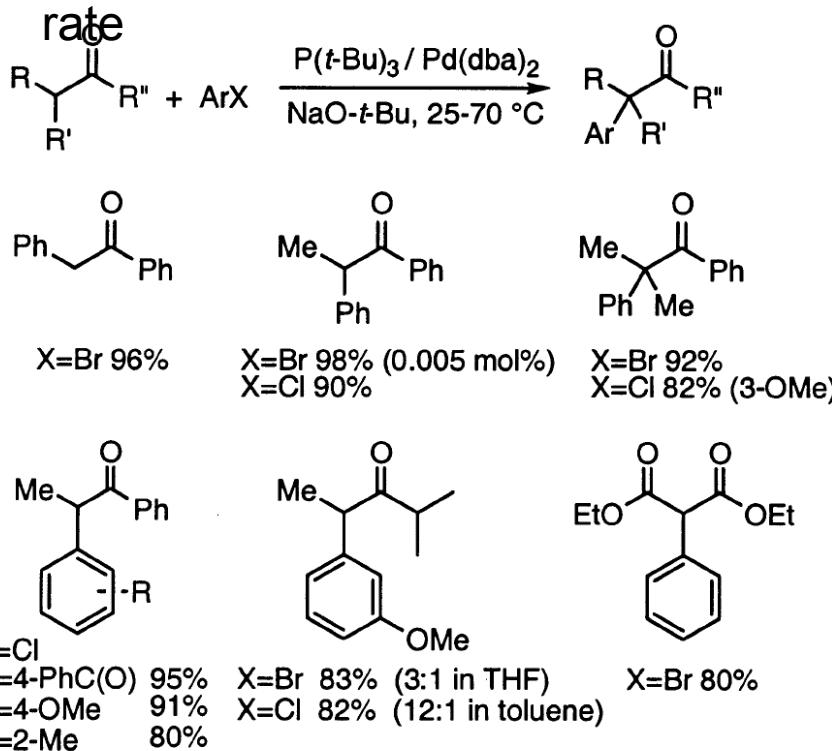


Mechanism:

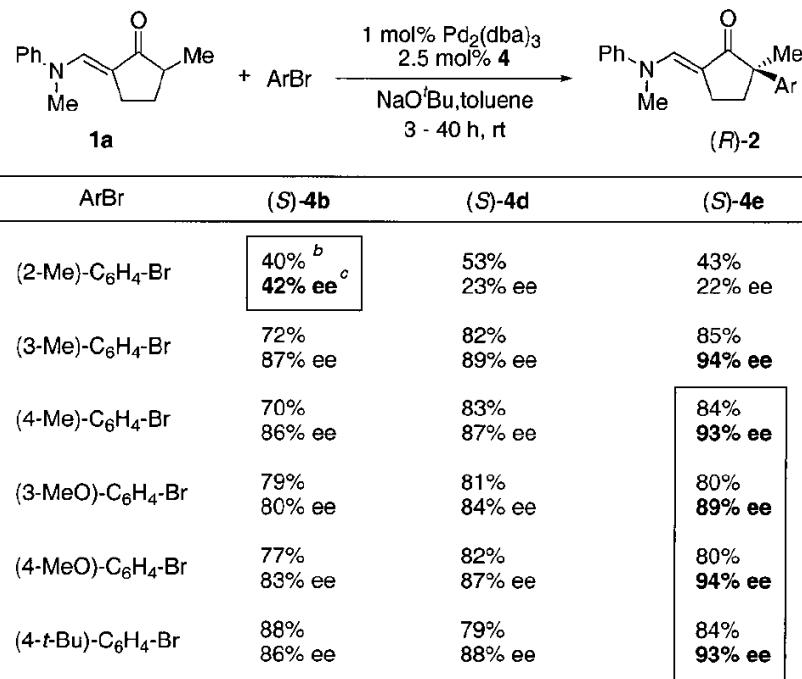


Pd-catalyzed Arylation of Enolates: Ketone Improvements

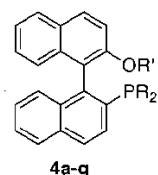
$\text{P}(t\text{-Bu})_3$: faster reaction



Asymmetric arylation:



^a Reaction conditions: 1.0 equiv ketone, 2.0 equiv ArBr, 2.0 equiv NaO'Bu, 0.25 M toluene, 3/Pd = 1.25/1. ^b Isolated yield. ^c % ee determined by HPLC (Daicel Chiralcel OD).



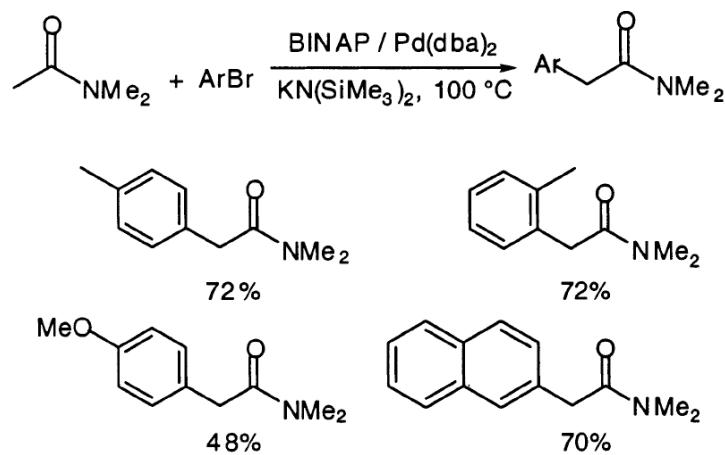
- 4a, R=Cy, R'=Me
- 4b, R='Pr, R'=Me
- 4c, R='Pr, R'=iPr
- 4d, R='Pr, R'=Bn
- 4e, R='Pr, R'=CH₂(1-naphth)
- 4f, R='Pr, R'=CH₂(2-naphth)
- 4g, R='Pr, R'=CH₂(9-phen)

Kawatsura, M. and Hartwig, J. F. *J. Am. Chem. Soc.* **1999**, 121, 1473

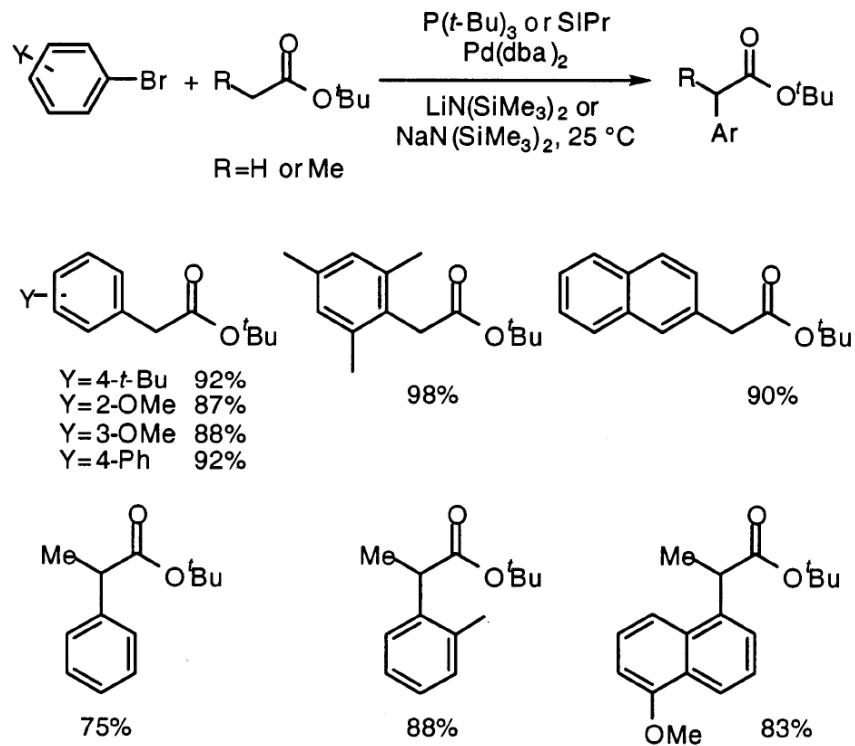
Hamada, T.; Chieffi, A.; Ahman, J. and Buchwald, S. L. *J. Am. Chem. Soc.* **2002**, 124, 1261

Pd-catalyzed Arylation of Enolates: Amides and Esters

Amides:

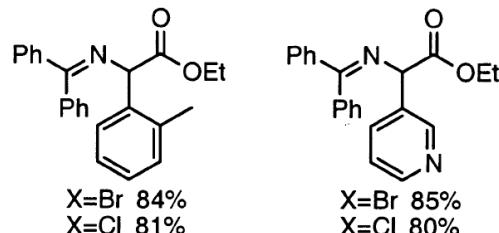
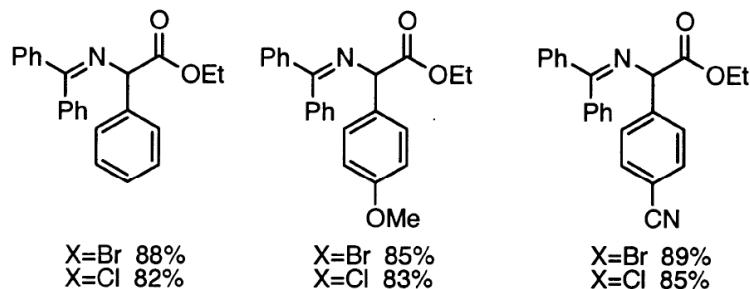
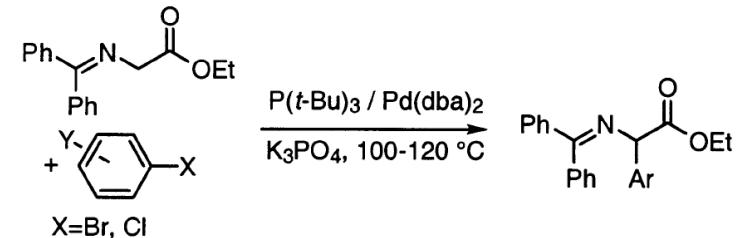


Esters:

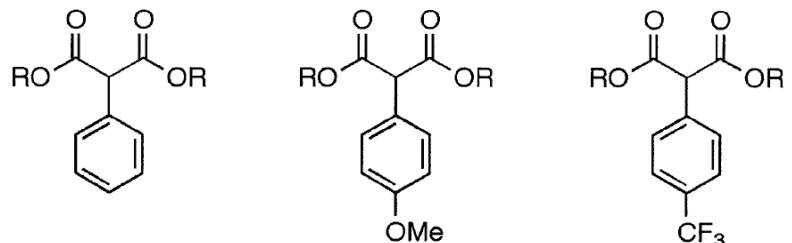
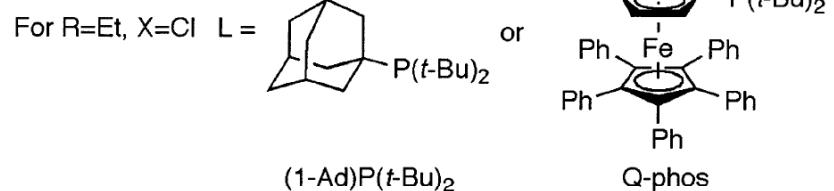
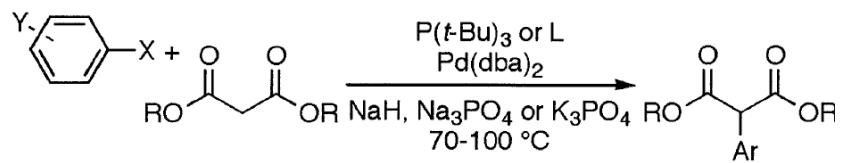


Pd-catalyzed Arylation of Enolates: Amino Acids and Malonates

Protected Amino Acids:



Malonates :

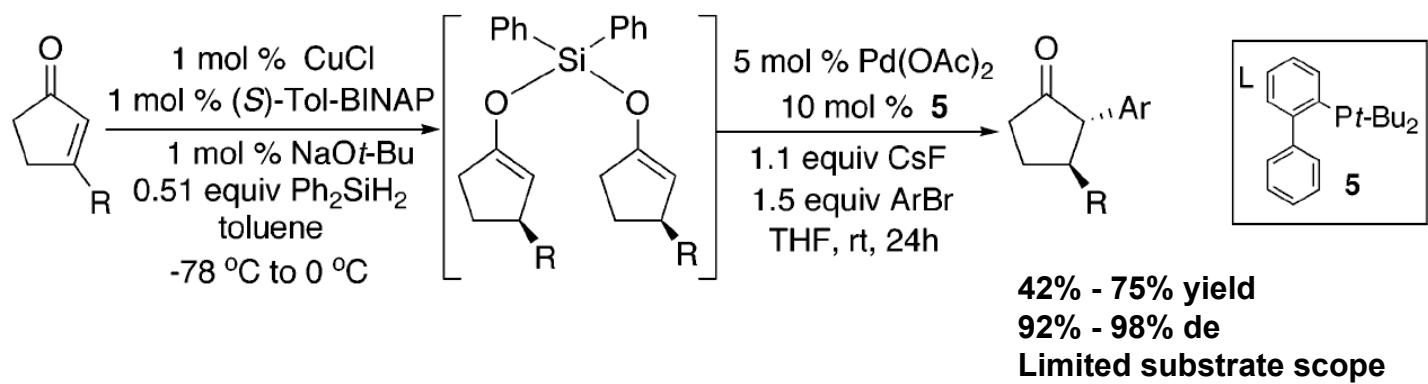
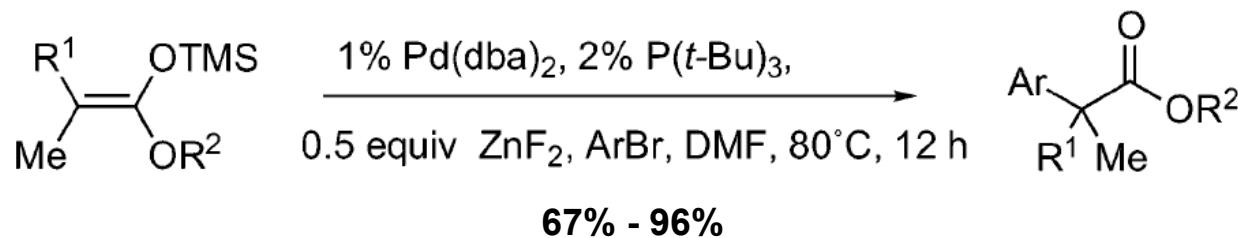
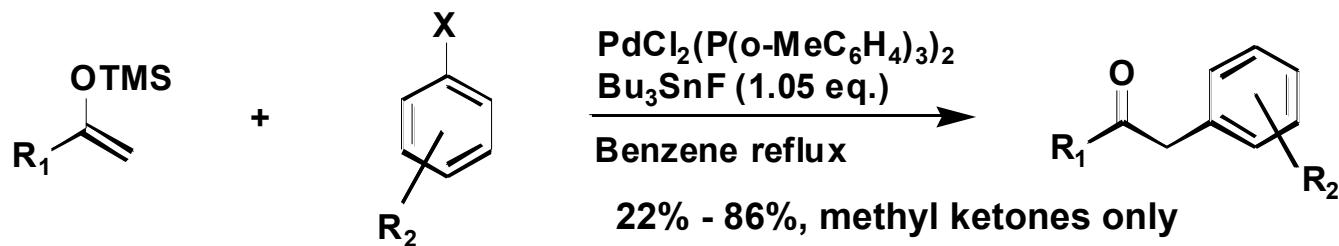


R=t-Bu X=Br 89% X=Cl 88%	R=t-Bu X=Br 90% X=Cl 84%	R=t-Bu X=Br 88% X=Cl 86%
R=Et X=Br 89% X=Cl 81%	R=Et X=Br 89% X=Cl 86%	R=Et X=Br 91% X=Cl 89%

Pd-catalyzed Arylation of Enolates: Problems

- Regioselectivity of ketone enolates: less hindered side is favored
- Strong basicity:
 - Functional group compatibility
 - Product's α -proton is more acidic than starting material:
 - Double arylation
 - Product may racemize under the reaction conditions

Pd-catalyzed Arylation of Silyl Enol Ethers: Previous Works



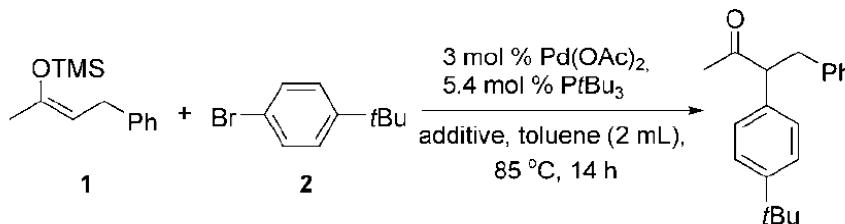
Kuwajima, I. and Urabe, H. *J. Am. Chem. Soc.* **1982**, *104*, 6831

Liu, X. and Hartwig, J. F. *J. Am. Chem. Soc.* **2004**, *126*, 5182

Chae, J.; Yun, J. and Buchwald, S. L. *Org. Lett.* **2004**, *6*, 4809

Pd-catalyzed Arylation of Silyl Enol Ethers with Two Metal Fluorides as Additives: Optimizing Reaction Conditions

Table 1: Coupling of **1** with **2**.^[a]



Entry	Additive (equiv)	Ratio of 1/2	Yield [%] ^[b]
1		1.2:1	0
2	Bu ₃ SnF (1.2)	1.2:1	34
3	CsF (1.2)	1.2:1	18
4	Bu ₃ SnF (1.2) CsF (1.2)	1.2:1	81
5	ZnF ₂ (1.2)	1.2:1	38 ^[c]
6	Me ₄ NF (1.2)	1.2:1	0
7	Bu ₃ SnF (1.2) CsF (1.2)	1:1.2	65
8	Bu ₃ SnF (1.4) CsF (1.4)	1.4:1	98
9	Bu ₃ SnF (0.14) CsF (1.4)	1.4:1	67
10	CsF (1.4)	1.4:1	81 ^[d]
11	Bu ₃ SnF (1.4) CsF (1.4)	1.4:1	93 ^[d]

[a] Reactions were run at 85 °C with 0.5 mmol **2** (0.25 M). [b] Yield of the isolated product (average of two runs). [c] DMF was used as the solvent.

[d] THF was used as the solvent.

Pd-catalyzed Arylation of Silyl Enol Ethers with Two Metal Fluorides as Additives: Substrate Scope

Coupling to vinyl bromide:

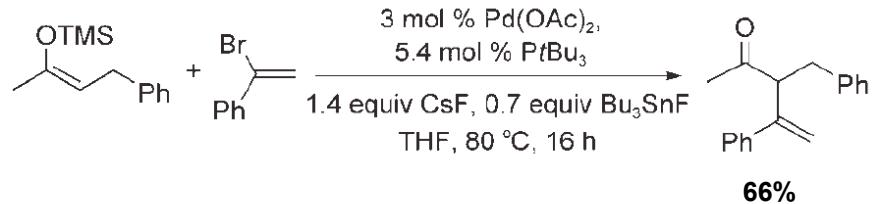


Table 2: Scope of the arylation of a trimethylsilyl enol ether.^[a]

Entry	Silyl enol ether	Halide	Products	Yield [%] ^[b]
1				89 ^[c]
2				84
3				96
4				97
5				97
6				93
7				91
8				55
9				80 ^[c]
10				80 ^[c]
11				70 ^[c]
12				84
13				77 ^[c]
14				85
15				78

[a] Reactions were run under the conditions of entry 8 in Table 1 for 12–20 h. [b] Yield of the isolated product (average of two runs). [c] Reaction run at 90 °C.

Pd-catalyzed Arylation of Silyl Enol Ethers with Two Metal Fluorides as Additives: Tin-free Conditions

Table 3: Scope of the palladium-catalyzed arylation of trimethylsilyl enol ethers in the presence of ZnF_2 and CsF or MnF_2 .

Entry	Silyl enol ether	Halide	Product	Yield [%]
1				89 ^[a]
2				87 ^[a]
3				90 ^[a]
4				53 ^[a]
5				78 ^[b]
6				71 ^[b,c]
7				64 ^[d,e]
8				68 ^[f,g]

[a] Reaction conditions: aryl halide (1.0 equiv), silyl enol ether (1.4 equiv), zinc fluoride (1.4 equiv), cesium fluoride (0.4 equiv), $[Pd(dba)_2]$ (3 mol%), and $PtBu_3$ (5.4 mol%), 85 °C; DMF (1 mL) was added per 0.2 mmol of aryl halide. [b] Reaction conditions: aryl halide (1.0 equiv), silyl enol ether (1.5 equiv), zinc fluoride (1.0 equiv), manganese fluoride (0.4 equiv), $[Pd(dba)_2]$ (2 mol%), and $PtBu_3$ (4 mol%), 70 °C; DMF (1 mL) was added per 0.2 mmol of aryl halide. [c] The ratio of mono/diarylation was 5.5:1. [d] Reaction conditions: aryl halide (1.0 equiv), silyl enol ether (1.5 equiv), zinc fluoride (1.4 equiv), manganese fluoride (1.4 equiv), $[Pd(dba)_2]$ (3 mol%), and $PtBu_3$ (5.4 mol%), 60 °C; DMF (1 mL) was added per 0.2 mmol of aryl halide. [e] The ratio of mono/diarylation was 4:1. [f] Reaction conditions: aryl halide (1.0 equiv), silyl enol ether (5.0 equiv), zinc fluoride (1.4 equiv), manganese fluoride (1.4 equiv), $[Pd(dba)_2]$ (3 mol%), and $PtBu_3$ (5.4 mol%), 70 °C; DMF (2 mL) was added per 0.2 mmol of aryl halide. [g] The ratio of mono/diarylation was 5.5:1.

Summary

- Pd-catalyzed coupling of silyl enol ethers with aryl bromides and chlorides was realized by the combination of two metal fluorides (“synergistic effect”):
 - $\text{CsF} + \text{Bu}_3\text{SnF}$: non-polar solvent
 - $\text{ZnF}_2/\text{MnF}_2$ or ZnF_2/CsF : Tin-free
- Future Work:
 - Origin of the synergistic effect
 - Milder conditions
 - Monoarylation
 - Asymmetric version