

Iodomethylzinc Phosphates: Powerful Reagents for the Cyclopropanation of Alkenes

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Poulard

J. Am. Chem. Soc. ASAP

Natural Product Containing Cyclopropanes

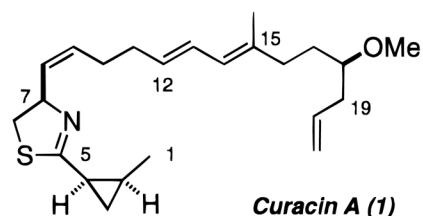
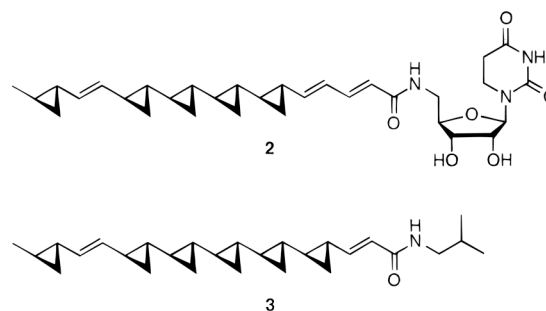
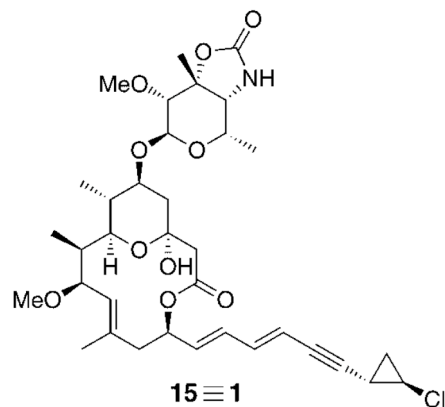


Figure 1. Curacin A, an antimetabolic metabolite of the blue-green alga *L. majuscula*.

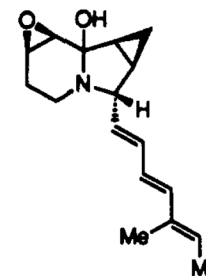


2-Anti-Fungal Agent FR-900848

3-Cholesteryl Ester Transfer Protein Inhibitor
U-1006305



Anticancer Compound Callipeltoside



Antibiotic Compound Indolizomycin

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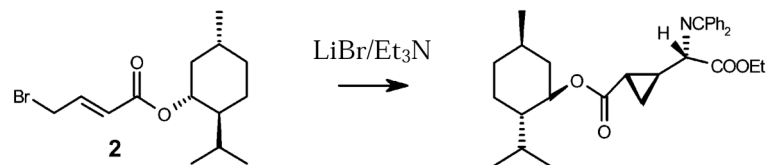
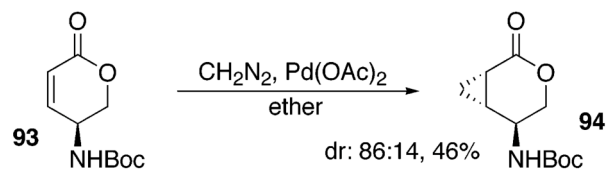
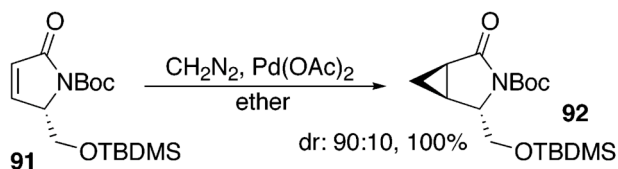
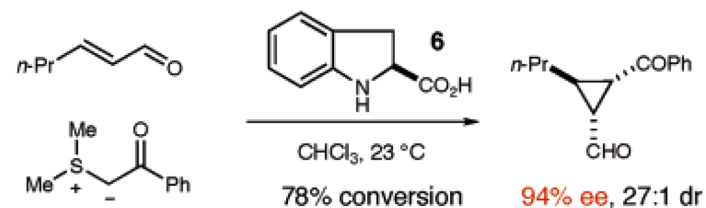
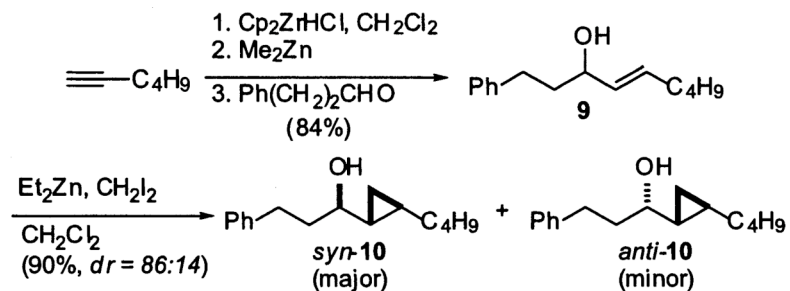
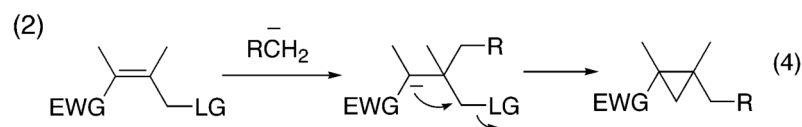
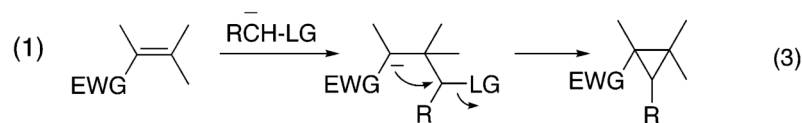
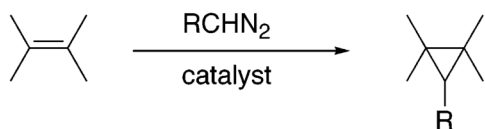
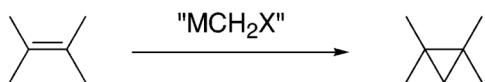
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General Types of Cyclopropanation



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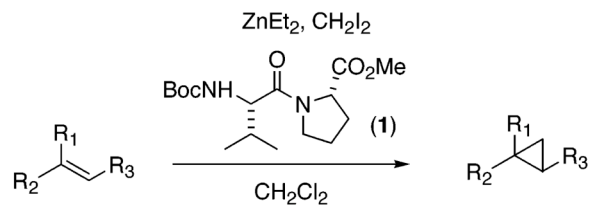
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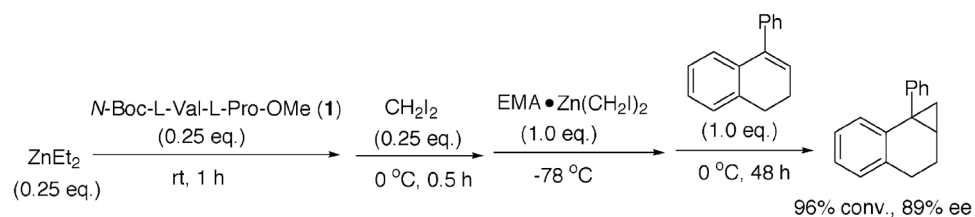
Shi Cyclopropanation

Stoichiometric Ligand



entry	substrate	yield (%) ^d	ee (%)
1		71	72 ^f
2 ^b		83	75 ^g
3 ^c		43	89 ^g
4		71	75 ^g
5 ^c		78	90 ^f
6		84	78 ^f (98 ^h)
7		83	90 ^f (99 ^h)

Catalytic Ligand



Entry	Substrate	Yield ^b (%)	ee ^c (%)
1		87	89
2		60	89
3		85	77
4		52	78
5		96	87

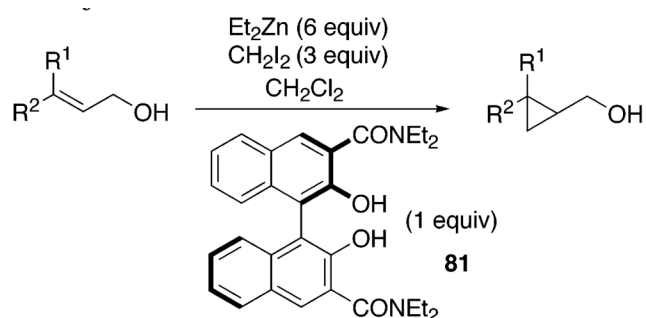
^a The cyclopropanation was carried out with olefin (1.0 equiv), ZnEt₂ (1.25 equiv), CH₂I₂ (2.25 equiv), ethyl methoxyacetate (1.0 equiv), dipeptide **1** (0.25 equiv), and ZnI₂ (0.25 equiv) in CH₂Cl₂ at 0 °C for 48 h except for entry 5, where the reaction was carried out at -40 °C for 72 h.

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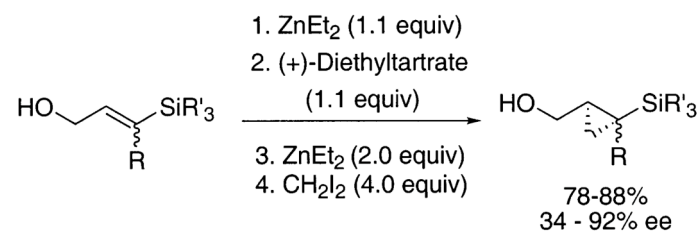
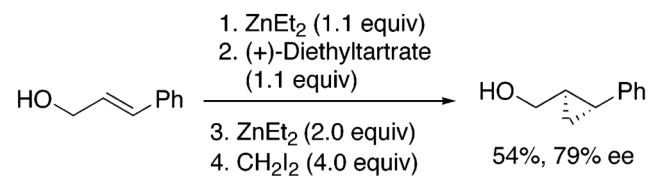
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9/6/2005

Stoichiometric Chiral Ligands

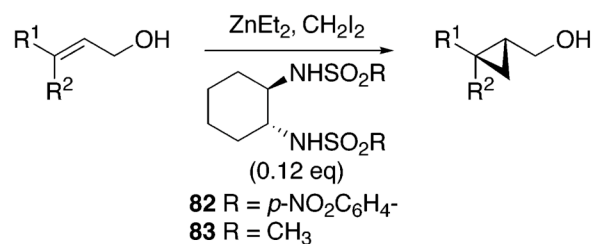


Entry	Substrate	Yield (%)	% ee
1		78	94
2		50	90
3		65	89
4		64	88
5		59	87
6		34	65



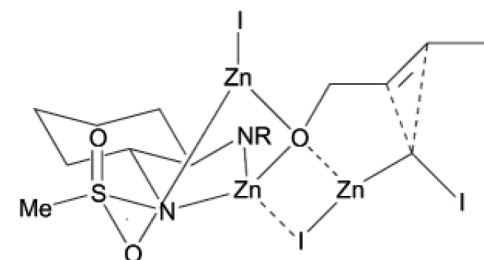
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Catalytic Chiral Ligands



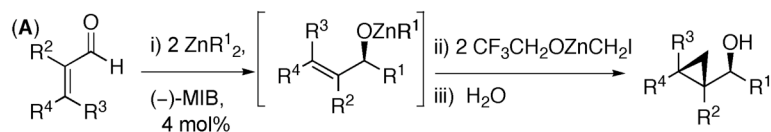
entry	R ¹	R ²	ee (%); yield (%)	
			catalyst 82	catalyst 83
1	Ph	H	76; 82	89; 92 ^a
2	H	Ph	75; 71	81; 81
3	PhCH ₂ CH ₂	H	82; 100	89; 88
4	BnOCH ₂	H	36; 70	
5	TrOCH ₂	H	80; 86	
6	H	BnOCH ₂	13; 36	
7	H	TrOCH ₂	65; 77	
8	Bu ₃ Sn	H	86; 94	
9	Me ₂ PhSi	H	81; 83	
10	H	Bu ₃ Sn	66; 75	
11	H	Me ₂ PhSi	59; 67	
12	H	PhCH ₂ CH ₂	72; 93	72; 93

^a In situ generation of ZnI₂.



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 Lebel, H.; Marcoux, J. F.; Molinaro, C.; Charette, A. B., *Chem. Rev.* **2003**, 977

Tandem Cyclopropanation Reactions



entry	ZnR ₂	cyclopropyl alcohol	ee (%)	dr ^a	yield (%)
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1	Et ^b		99	>20:1	90
2	-(CH ₂) ₄ iPr ^c		97	>20:1	66
3	-(CH ₂) ₅ OTBDPS ^c		98	>20:1	75

4	Me ^c		99	>20:1	76
5	Et		95	>20:1	78
6	-(CH ₂) ₄ iPr ^c		96	>20:1	64

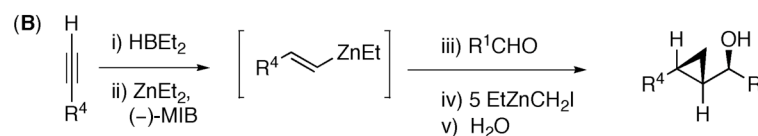
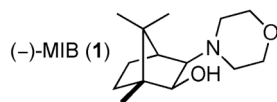
7	Me ^c		95	>20:1	85
8	Et		96	>20:1	90

9	Et		89	>20:1	87
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10	Et		98	>20:1	80
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11	Et		91	>20:1	91
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^a Determined by crude ¹H NMR analysis. ^b Stereochemistry assigned by X-ray analysis. See the Supporting Information. ^c With 5 equiv of ZnEt₂, 5 equiv of CF₃CH₂OH, and 5 equiv of CH₂I₂.



entry	cyclopropyl alcohol	R ⁴	ee (%)	dr ^a	yield (%)
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1		Ph	99	>20:1	75
2		nBu	92	>20:1	71
3		tBu	87	>20:1	78

4		Ph	99	>20:1	78
5		nBu	93	>20:1	84
6		tBu	96	>20:1	74
7		(CH ₂) ₄ Cl ^b	94	>20:1	80
8		CH ₂ CH ₂ OTr	93	>20:1	73

entry	ZnR ₁ ₂	cyclopropyl alcohol	ee (%)	dr ^a	yield (%)
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1	Et ^b		99	>20:1	68
2	-(CH ₂) ₅ OTBDPS		98	>20:1	70

3	Me		99	>20:1	78
4	Et ^b		95	>20:1	62
5	-(CH ₂) ₄ iPr		96	>20:1	60

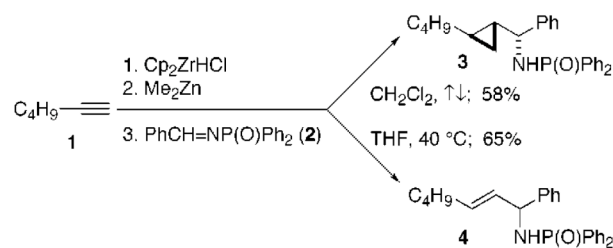
6	Et		89	>20:1	70
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7	Et		96	>20:1	56
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8	Et		98	>20:1	74
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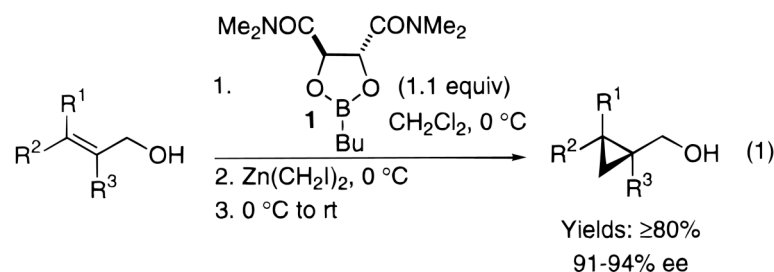
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Tandem Cyclopropanation Reactions con'd



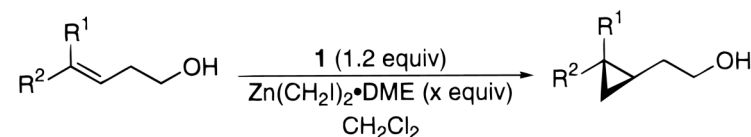
entry	alkyne	imine	amino cyclopropane	yield ^d
1	$\text{C}_4\text{H}_9\text{—}\equiv\text{C—C}$ 1	Ph—CH=NP(O)Ph_2 2	$\text{C}_4\text{H}_9\text{—}\triangle\text{—Ph}$ 3	74%
2	$\text{C}_2\text{H}_5\text{—}\equiv\text{C—C}_2\text{H}_5$ 8	2	$\text{C}_2\text{H}_5\text{—}\triangle\text{—Ph}$ 9	46%
3	$\text{BDPSO}(\text{CH}_2)_2\text{—}\equiv\text{C—C}$ 10	2	$\text{BDPSO}(\text{CH}_2)_2\text{—}\triangle\text{—Ph}$ 11	68%
4	$\text{TIPSO}_2\text{C}(\text{CH}_2)_3\text{—}\equiv\text{C—C}$ 12	2	$\text{TIPSO}_2\text{C}(\text{CH}_2)_3\text{—}\triangle\text{—Ph}$ 13	73%
5	$\text{EtO}_2\text{CN}(\text{Ts})(\text{CH}_2)_2\text{—}\equiv\text{C—C}$ 14	2	$\text{EtO}_2\text{CN}(\text{Ts})(\text{CH}_2)_2\text{—}\triangle\text{—Ph}$ 15	45%
6	1	$(p\text{-MeO}_2\text{C})\text{Ph—CH=NP(O)Ph}_2$ 16	$\text{C}_4\text{H}_9\text{—}\triangle\text{—Ph—CO}_2\text{Me}$ 17	69%
7	10	16	$\text{OBdPS—}\triangle\text{—Ph—CO}_2\text{Me}$ 18	84%
8	1	$(p\text{-Cl})\text{Ph—CH=NP(O)Ph}_2$ 19	$\text{C}_4\text{H}_9\text{—}\triangle\text{—Ph—Cl}$ 20	65%
9	1	$(m\text{-MeO})\text{Ph—CH=NP(O)Ph}_2$ 21	$\text{C}_4\text{H}_9\text{—}\triangle\text{—Ph—OMe}$ 22	51%
10	1	$\text{Ph—}\equiv\text{C—CH=NP(O)Ph}_2$ 23	$\text{C}_4\text{H}_9\text{—}\triangle\text{—Ph—C}\equiv\text{C—Ph}$ 24	44%

Previous Charette Group Work



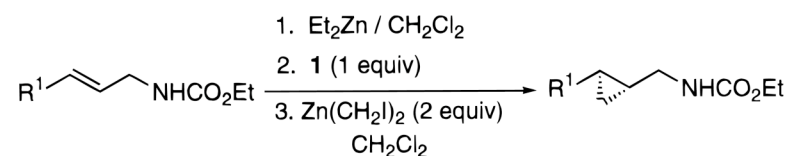
entry	R ₁	R ₂	R ₃	proc ^a	yield, ^b % (cpd)	ee, ^c %
1	H	Ph	H	A	>98 (5)	93 ^d
				B	95 (5)	94 ^d
2	H	3-MeOPh	H	B	>98 (6)	93 ^d
3	H	Pr	H	A	80 (7)	93 ^e
4	H	PhCH ₂ CH ₂	H	B	90 (8)	94 ^f
5	H	BnOCH ₂	H	B	87 (9)	94 ^f
6	H	Bu ₃ Sn	H	A	82 (10)	ca. 90 ^g
				B	88 (10)	ca. 90 ^g
7	H	I	H	C	83 (11)	90 ^h
8	Bu ₃ Sn	H	H	B	73 (12)	ca. 90 ^g
9	I	H	H	C	71 (13)	83 ^e
10	CH ₃ CH ₂	H	H	A	90 (14)	93 ^h
				D	>98 ⁱ (14)	ca. 87 ^g
11	TBDPSOCH ₂	H	H	A	80 (15)	91 ^h
12	BnOCH ₂	H	H	A	93 (16)	91 ^f
				B	92 (16)	91 ^f
13	CH ₃	CH ₃	H	A	85 (17)	94 ^h
14	H	Ph	CH ₃	A	96 (18)	85 ^d
				B	80 (18)	82 ^d
15	H	CH ₃ CH ₂	CH ₂ OTIPS	B	>98 (19)	89 ^e
16	CH ₃	CH ₃	CH ₂ OTIPS	B	85 (20)	88 ^h
17	H	-CH ₂ CH ₂ CH ₂ CH ₂ -		B	84 (21)	60 ^h

^a Procedure A (with <1.0 mmol): 2 equiv of $\text{Zn}(\text{CH}_2\text{I}_2)$ at 0 °C, then rt for 2 h. Procedure B (with <1.0 mmol): 2 equiv of $\text{Zn}(\text{CH}_2\text{I})_2 \cdot \text{DME}$ at -10 °C, then rt for 8 h. Procedure C (with <1.0 mmol): 5 × 2 equiv of $\text{Zn}(\text{CH}_2\text{I})_2 \cdot \text{DME}$ at -10 °C. Procedure D: dioxaborolane **22** was used instead of **1**. ^b Unless otherwise noted, these



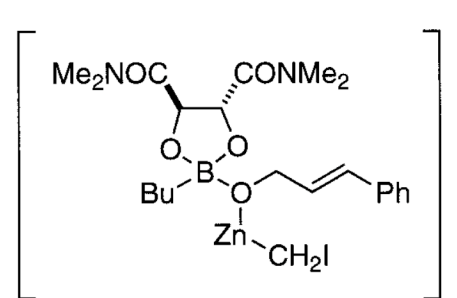
entry	R ₁	R ₂	x equiv	yield, ^a % (cpd)	ee, %
1	CH ₃ CH ₂	H	2.0	90 (43)	82 ^b
2	H	Ph	3.0	90 (44)	82 ^c
3	Ph	H	4.0	86 (45)	81 ^c

^a Isolated yields. ^b Determined by ¹³C NMR of the corresponding Mosher ester. ^c Determined by chiral HPLC.

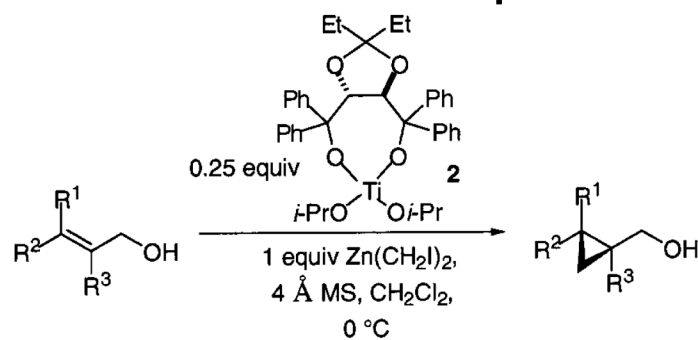


entry	R ₁	yield, ^a % (cpd)	ee, ^b %
1	Ph	21 (46)	53
2	PhCH ₂ CH ₂	25 (47)	~55

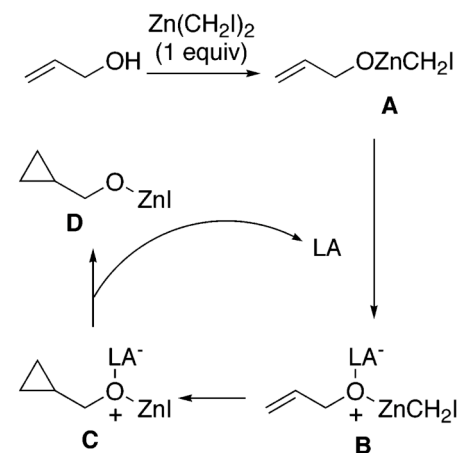
^a Isolated yields. ^b Determined by chiral HPLC.



Previous Charette Group Work con'd



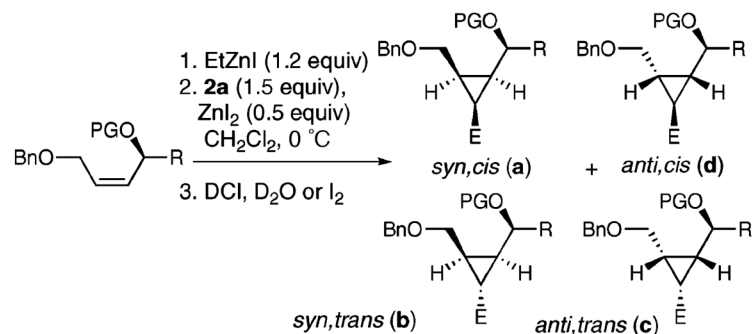
Entry	R ¹	R ²	R ³	Yield ^a (%)	er	er K-D ^h
1 ^g	H	Ph	H	85	96 : 4 to 97 : 3 ^b	---
2 ^g	H	Ph	H	83	4 : 96 ^{b,f}	5 : 95
3 ^h	Ph	H	H	62	86 : 14 ^b	9 : 91
4 ^g	Me	Ph	H	80	94 : 6 ^b	13 : 87
5 ^h	H	Ph	Me	80	75 : 25 ^b	47 : 53
6 ^g	H	3,5-Me ₂ -Ph	H	86	96 : 4 ^b	---
7 ^g	H	2-napht	H	81	96 : 4 ^c	---
8 ^h	H	1-napht	H	80	92 : 8 ^c	---
9 ⁱ	H	<i>p</i> -Me-OPh	H	90	96 : 4 ^b	---
10 ^g	H	<i>p</i> -Cl-Ph	H	56	91 : 9 ^c	---
11 ^h	H	<i>p</i> -Cl-Ph	H	81	91 : 9 ^c	---
12 ^h	H	Pr	H	68	87 : 13 ^d	---
13 ^h	Pr	H	H	87	74 : 26 ^e	---
14 ^h	H	PhCH ₂ CH ₂	H	63	80 : 20 ^c	5 : 95
15 ^h	H	Cyclohexyl	H	60	83 : 17 ^d	---
16 ^h	Me	Me	H	89	86 : 14 ^e	---
17 ^g	H		H	73	94 : 6 ^b	---
18 ^g	H		H	86	92 : 8 ^c	---



Charette, A. B.; Molinaro, C.; Brochu, C., *J. Am. Chem. Soc.* **2001**, 12168

Previous Charette Group Work con'd

Table 1. Diastereoselective Zinco-Cyclopropanation of *cis*-Allylic Alcohols and Ethers



entry	R (PG)	dr (a:(b + c + d))	yield (%) (E = D)	yield (%) (E = I)
1	Me (H) (5)	>95:5	82 (6)	75 (7)
2	Et (H) (8)	>95:5	85 (9)	84 (10)
3	<i>i</i> -Pr (H) (11)	>95:5	86 (12)	87 (13)
4	<i>t</i> -Bu (H) (14)	>95:5	91 (15)	91 (16)
5	Ph (H) (17)	>95:5	62 (18)	58 (19)
6	Me (Bn) (20)	>95:5	86 (21)	77 (22)

Scheme 1

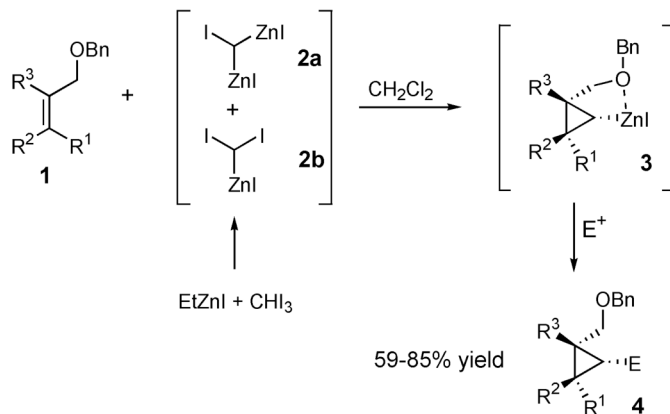
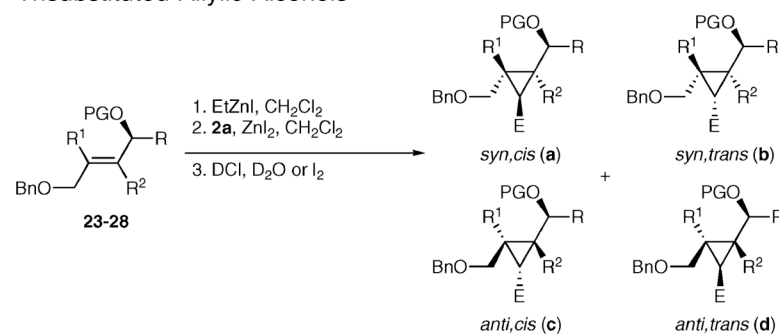


Table 2. Diastereoselective Zinco-Cyclopropanation of Trisubstituted Allylic Alcohols^a



entry	PG	R	R ¹	R ²	dr (<i>syn:anti</i>)	dr (<i>cis:trans</i>)	yield (%)
1	H	Me	H	H (23)	60:40		85 (29) ^b
2	H	<i>t</i> -Bu	H	H (24)	94:6	28:72	73 (30) ^b
3	TIPS	<i>t</i> -Bu	H	H (25)	>95:5	75:25	64 (31) ^b
4	H	Me	H	TMS (26)	<5:95	>95:5	68 (32c)
5	H	<i>t</i> -Bu	H	TMS (27)	<5:95	>95:5	77 (33c)
6	H	<i>t</i> -Bu	TMS	H (28)	>95:5	<5:95	84 (34b)
7	H	<i>t</i> -Bu	TMS	H (28)	>95:5	<5:95	81 (35b)

^a In entries 2–6, cyclopropylzinc was quenched with D₂O (E = D). I₂ was used in entries 2 and 7 (E = I). H₂O was used in entry 1. ^b Combined yield of the diastereomers.

Achiral Iodomethylzinc Phosphates

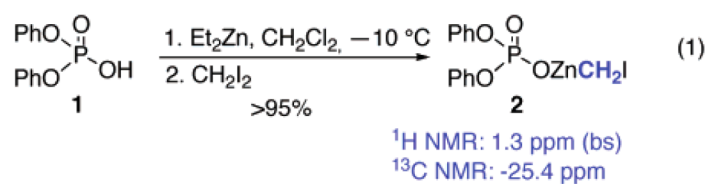


Table 1. Cyclopropanation Using Achiral Phosphoric Acid **2**

$$\begin{array}{c}
 \begin{array}{ccc}
 \text{R}^1 & & \text{R}^3 \\
 & \backslash & / \\
 & \text{C} = \text{C} & \\
 & / & \backslash \\
 \text{R}^2 & & \text{R}^4
 \end{array}
 \xrightarrow[\text{CH}_2\text{Cl}_2, -10\text{ }^\circ\text{C to rt, 15h}]{\begin{array}{c} (\text{PhO})_2\text{P}(\text{O})\text{OH} \text{ (n equiv)} \\ \text{Et}_2\text{Zn} \text{ (n equiv), CH}_2\text{I}_2 \text{ (n equiv)} \end{array}}
 \begin{array}{ccc}
 \text{R}^1 & & \text{R}^3 \\
 & \backslash & / \\
 & \text{C} - \text{C} & \\
 & / & \backslash \\
 \text{R}^2 & & \text{R}^4
 \end{array}
 \end{array}$$

entry	substrate	product	n	yield (%)
1		3	1.2	> 95 (98%) ^{a,b}
2		4	1.2	> 95 (98%) ^a
3		5	1.2	> 95 (98%) ^a
4		6	1.5	95 ^c
5		7	1.5	62 ^c

^a Determined by ¹H NMR using an internal standard. Isolated yield in parentheses. ^b Only the allylic alcohol double bond reacted. ^c Determined by GC analysis using an internal standard.

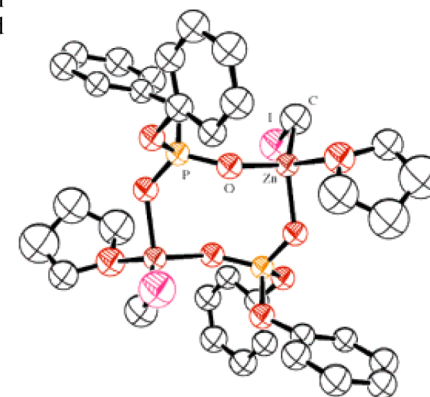
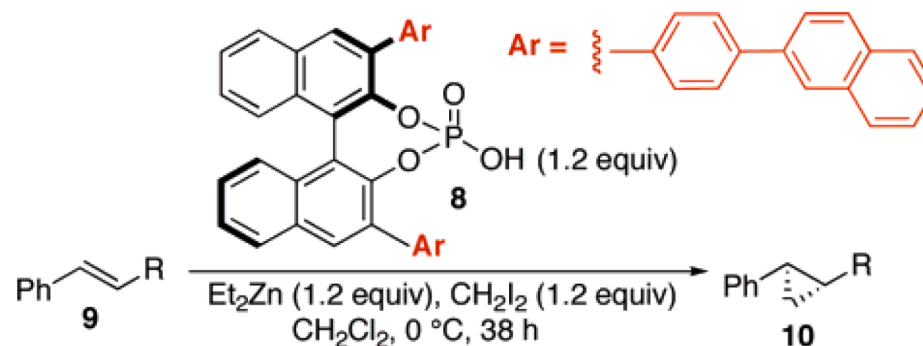


Figure 1. ORTEP drawing of **2**·THF dimer.

Chiral Stoichiometric Iodomethylzinc Phosphates

Table 2. Cyclopropanation with Chiral Phosphoric Acid **8**

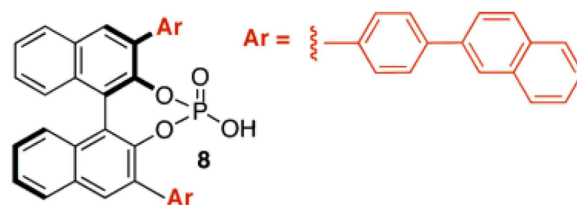
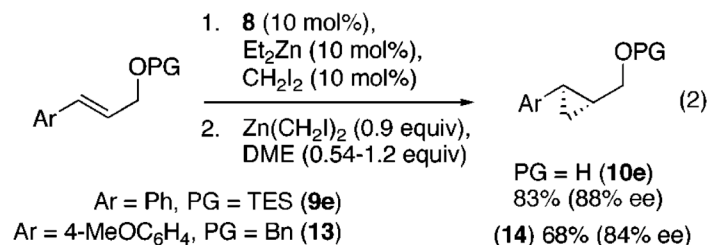
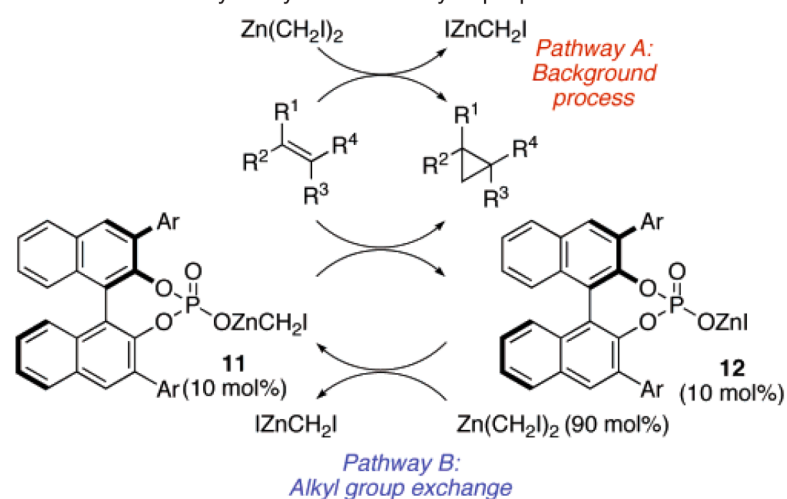


entry	R		yield (%) ^a	ee (%) ^b
1	CH_2OMe	(9a)	>95 (73)	91
2	CH_2OBn	(9b)	>95 (89)	90
3	CH_2OPMB	(9c)	95 (78)	85
4	CH_2OMOM	(9d)	75 (47)	90
5	CH_2OTES	(9e)	>95 (79) ^c	87
6 ^d	CH_2OTES	(9e)	>95 (84) ^c	92
7	$\text{CH}_2\text{CH}_2\text{OBn}$	(9f)	>95 (85)	93
8	CH_2OZnEt	(9g)	>95 (80)	39 ^e

^a Determined by ^1H NMR using an internal standard. Isolated yield in parentheses. ^b Determined by HPLC on chiral stationary phase. ^c The corresponding alcohol was obtained after deprotection with 1.0 M H_3PO_4 and NH_4F at $40\text{ }^\circ\text{C}$. ^d The reaction was performed at $-20\text{ }^\circ\text{C}$ for 48 h. ^e The opposite (*S,S*)-enantiomer was obtained.

Chiral Catalytic Iodomethylzinc Phosphates

Scheme 1. Catalytic Cycle for the Cyclopropanation Reaction



Conclusion

- The Charette group has reported a novel phosphate bound Zn reagent for the cyclopropanation of allylic alcohols. This reagent can be used in either stoichiometric or catalytic amounts to give moderate to good ee's and yields.
- The scope of both the stoichiometric and catalytic reaction needs to be expanded. The use of unfunctionalized olefins in this reaction would be an important improvement.
- Work is ongoing to develop a ligand which gives higher ee's in both reactions.