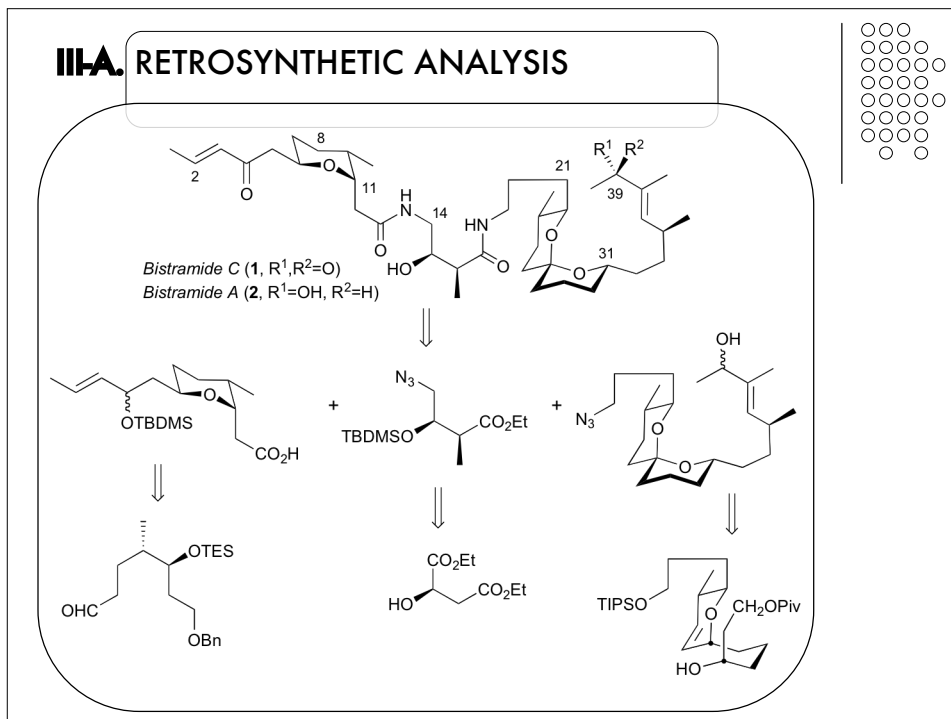


### III.A. RETROSYNTHETIC ANALYSIS



### Definitions

- **Disconnection:** An analytical operation, which breaks a bond and converts a molecule into a possible starting material. The reverse of a chemical reaction. Symbol  $\Rightarrow$  and a curved line drawn through the bond being broken.
- **FGI:** Functional Group Interconversion: The operation of writing one functional group for another so that disconnection becomes possible. The reverse of a chemical reaction. Symbol  $\Rightarrow$  with FGI written over it.
- **Reagent:** A compound which reacts to give an intermediate in the planned synthesis or to give the target molecule itself. The synthetic equivalent of a synthon.
- **Synthetic equivalent:** A reagent carrying out the function of a synthon which cannot itself be used, often because it is too unstable.
- **Synthon:** A generalized fragment, usually an ion, produced by a disconnection. (some people also use synthon for a synthetic equivalent).
- **Target Molecule:** The molecule whose synthesis is being planned.

## Retrosynthesis

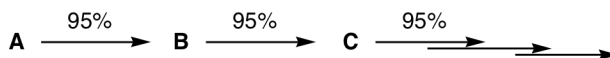
*Retrosynthesis* is the process of “deconstructing” a target molecule into readily available starting materials by means of

- imaginary breaking of bonds (*disconnections*) and by the conversion of one functional group into another (*functional group interconversions*).

The following factors need be taken into consideration:

Efficiency, e.g. the “arithmetic demon”:

“The arithmetic demon dictates one of the major axioms of synthesis: Get the most done in the fewest steps and in the highest yield.” (R. E. Ireland)



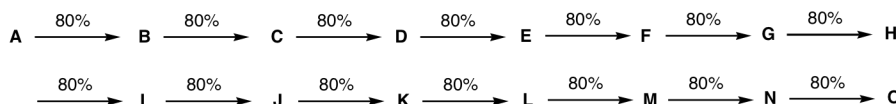
$$5 \text{ steps: } (0.95)^5 = 77\%$$

$$25 \text{ steps: } (0.95)^{25} = 28\%$$

$$100 \text{ steps: } (0.95)^{100} = 0.6\%$$

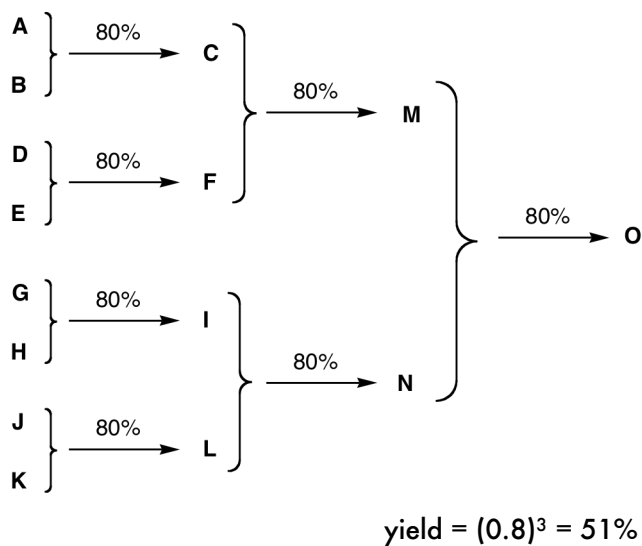


A linear synthesis with 14 starting materials and intermediates:

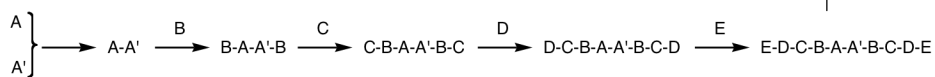


$$\text{yield} = (0.8)^{14} = 4\%$$

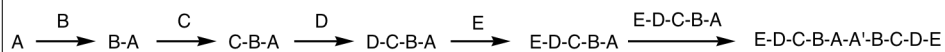
A convergent synthesis with 14 starting materials and intermediates:



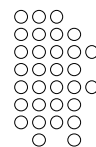
Question: If every reaction goes with a 60% yield, what is the more efficient strategy:



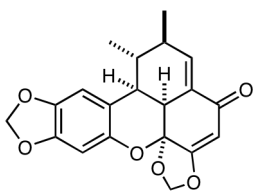
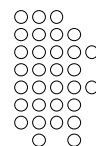
or



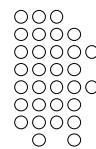
????????????????



Carpanone: Chapman, O. et al. *J. Am. Chem. Soc.* **1971**, 93, 6696.



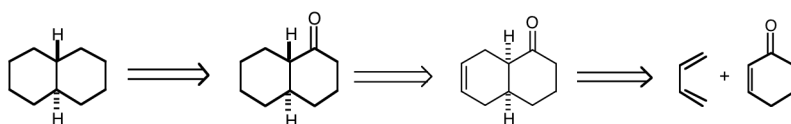
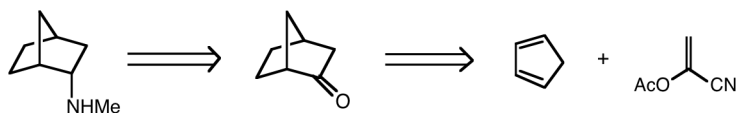
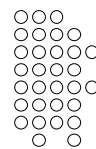
Mechanism:



Retrosynthetic analysis

General strategy:

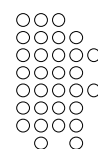
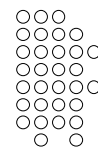
- Remove, add or change functional groups; FGI

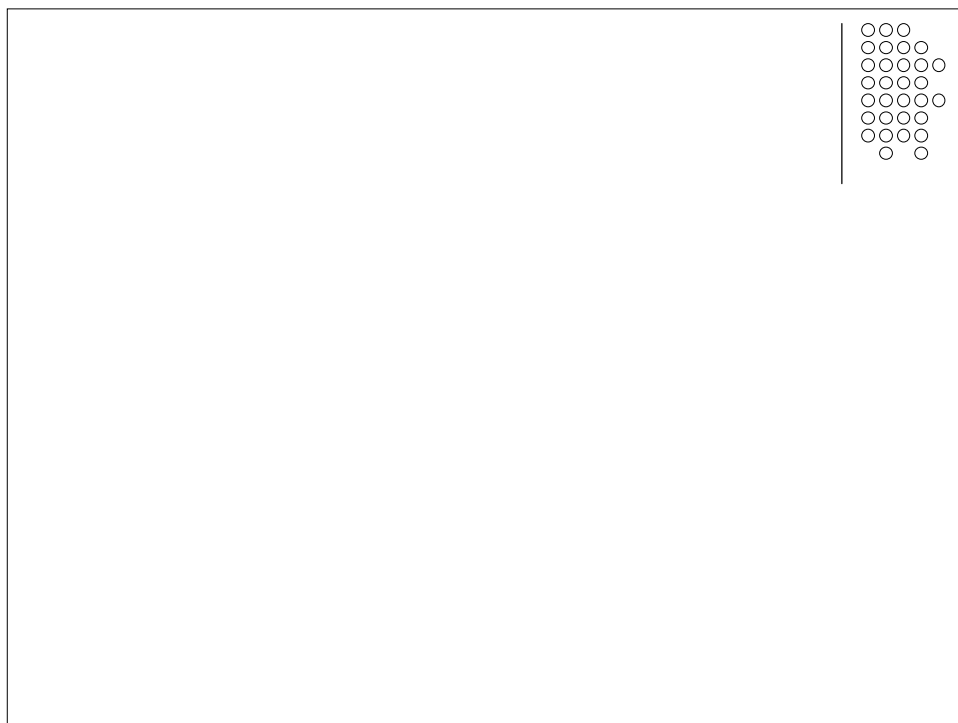
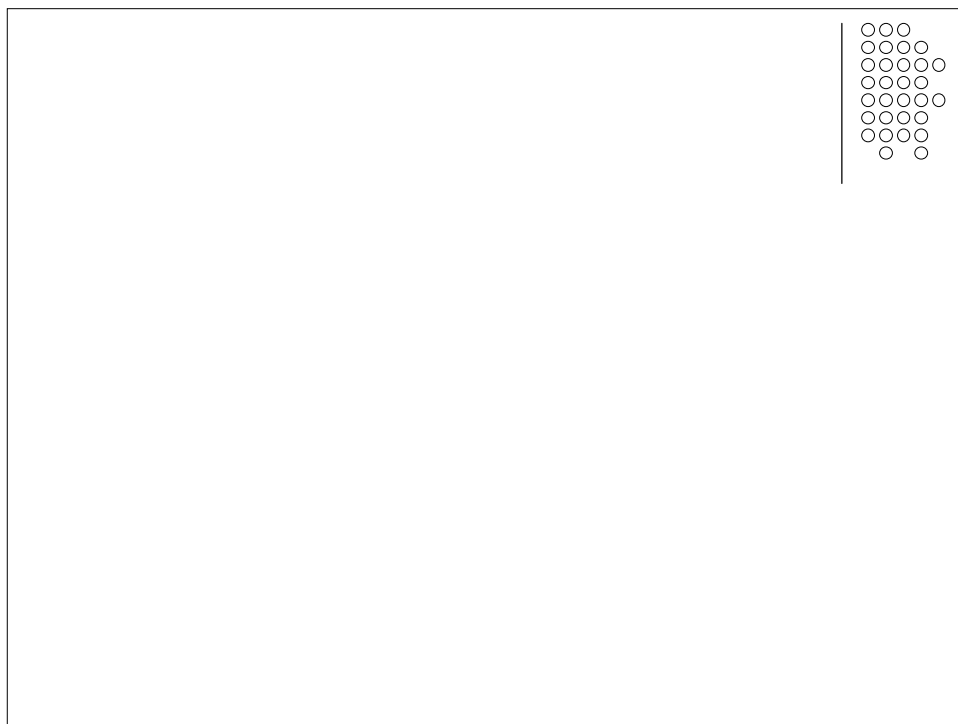


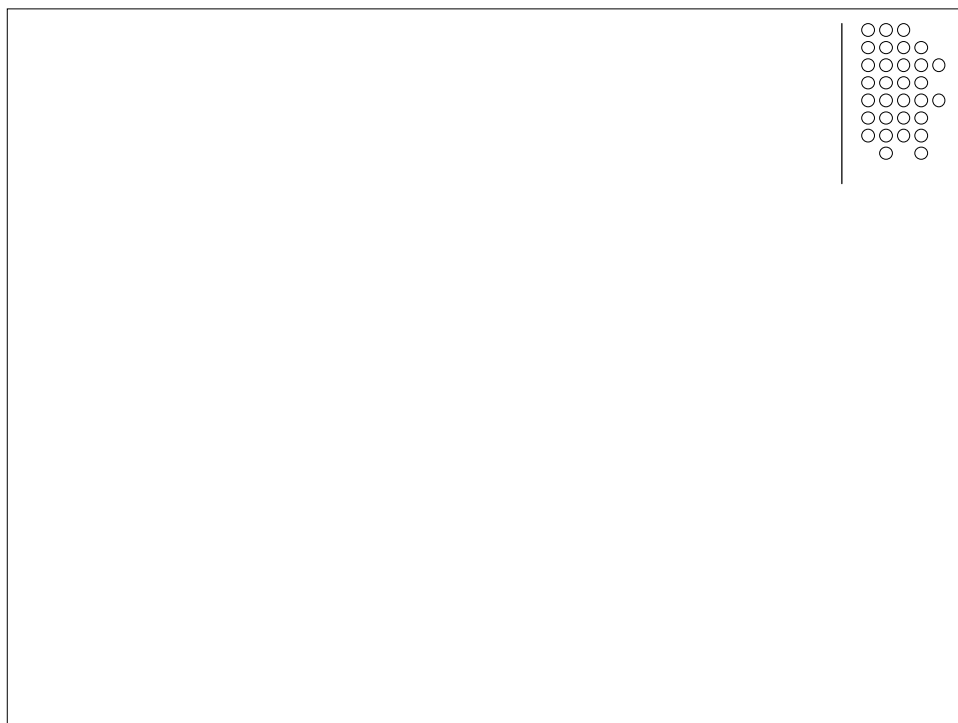
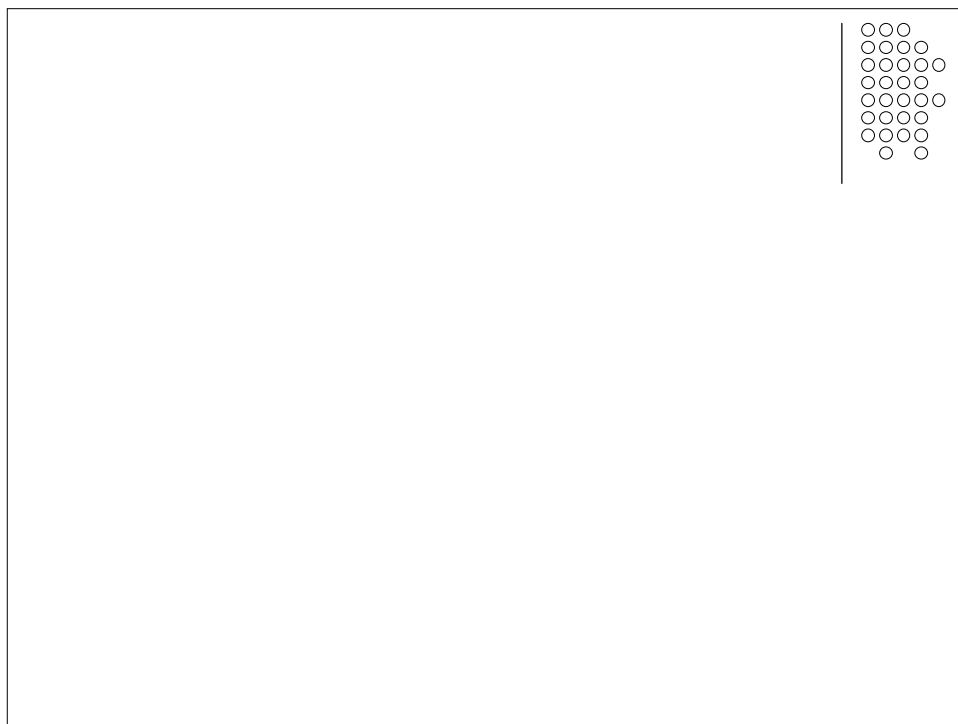
- Apply one- and two-group disconnections ("1,n-relationships), pericyclic reactions, etc.

Example:

Retrosynthesis 1:

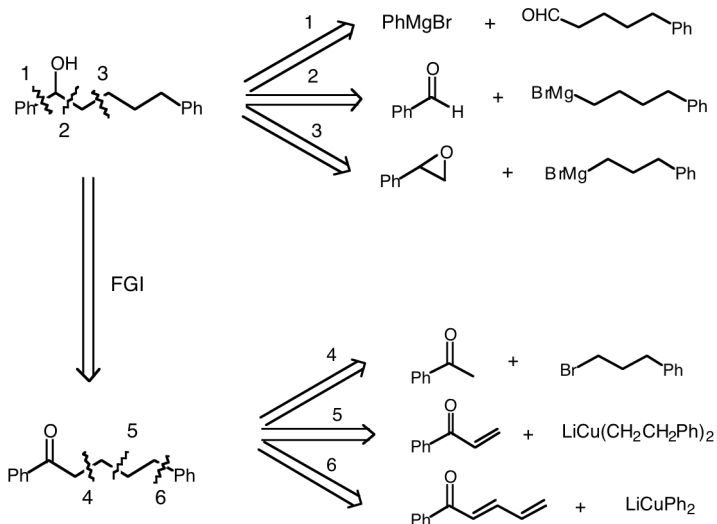








## Summary of retrosynthetic approaches:

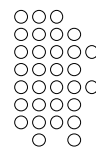


The "best" approach uses readily available starting materials in high yielding, reliable transformations. It may very often be based on personal choice.

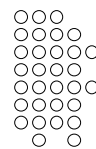


Warren, S. *Designing Organic Synthesis*, Wiley, 1978.

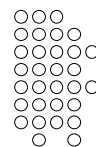
- One group disconnections:



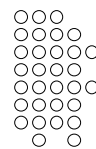
Two group disconnections:



Illogical two group disconnections:



Heteroatoms:



Corey, E. J. *J. Am. Chem. Soc.*  
**1975**, 97, 430

