

“Direct Asymmetric anti-Mannich Type Reactions Catalyzed by a Designed Amino Acid”

S. Mitsumori, H. Zhang, P. Cheong, K. Houk, F. Tanaka, and C. Barbas
J. Am. Chem. Soc. **2006**, *128*, 1040

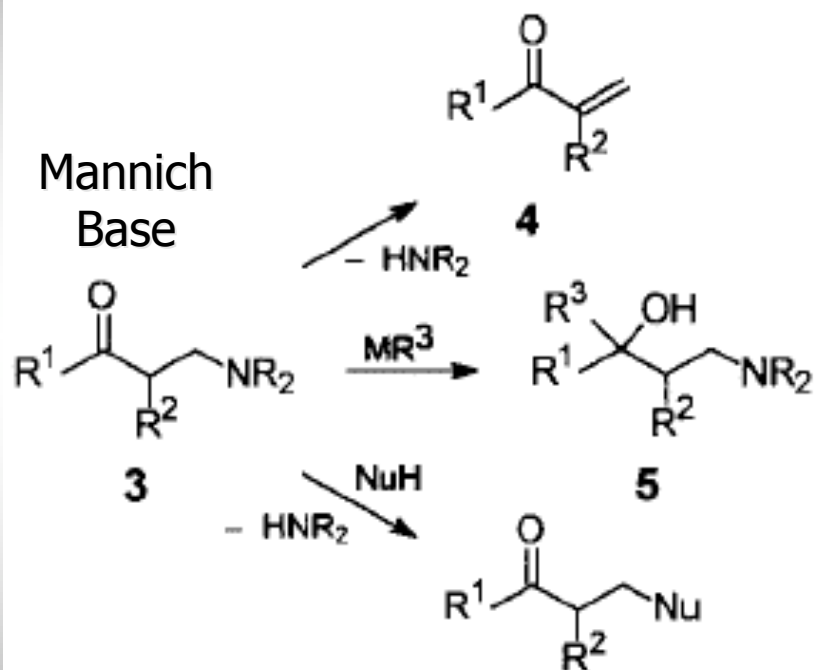
Presented by: Zack Brown
Chem 2320, Spring 2006

Mannich Reaction - Outline

- I. Current Implementations
- II. Classic Mannich Mechanism
- III. New Developments (Asymmetric)
 - Direct vs. Indirect
 - Proline Catalysis
- IV. Current Work
 - Designed Amino Acid
- V. Further Extensions

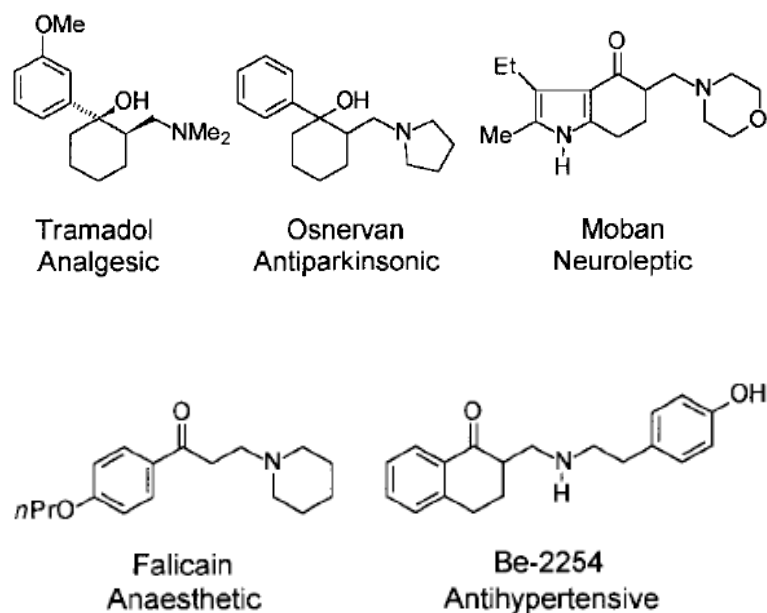
Implementations of the Mannich

Derivatization



- 3- \rightarrow 4: Formation of a Michael Acceptor
- 3- \rightarrow 5: Addition of Organometallic
- 3- \rightarrow 6: Sub. by a Nucleophile

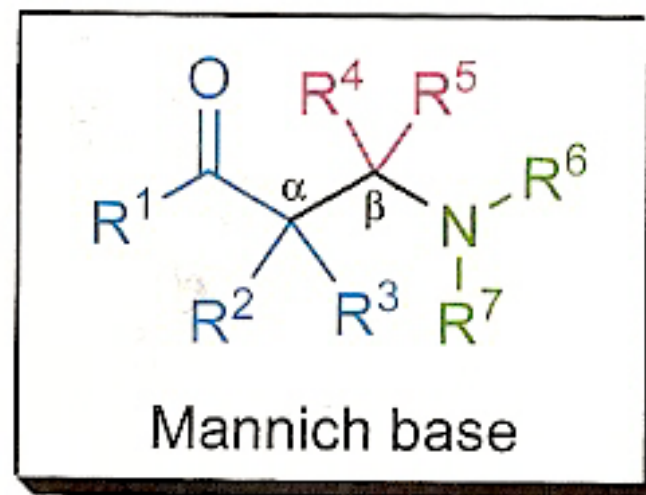
Pharmaceutical Examples



Classic Mannich Reaction

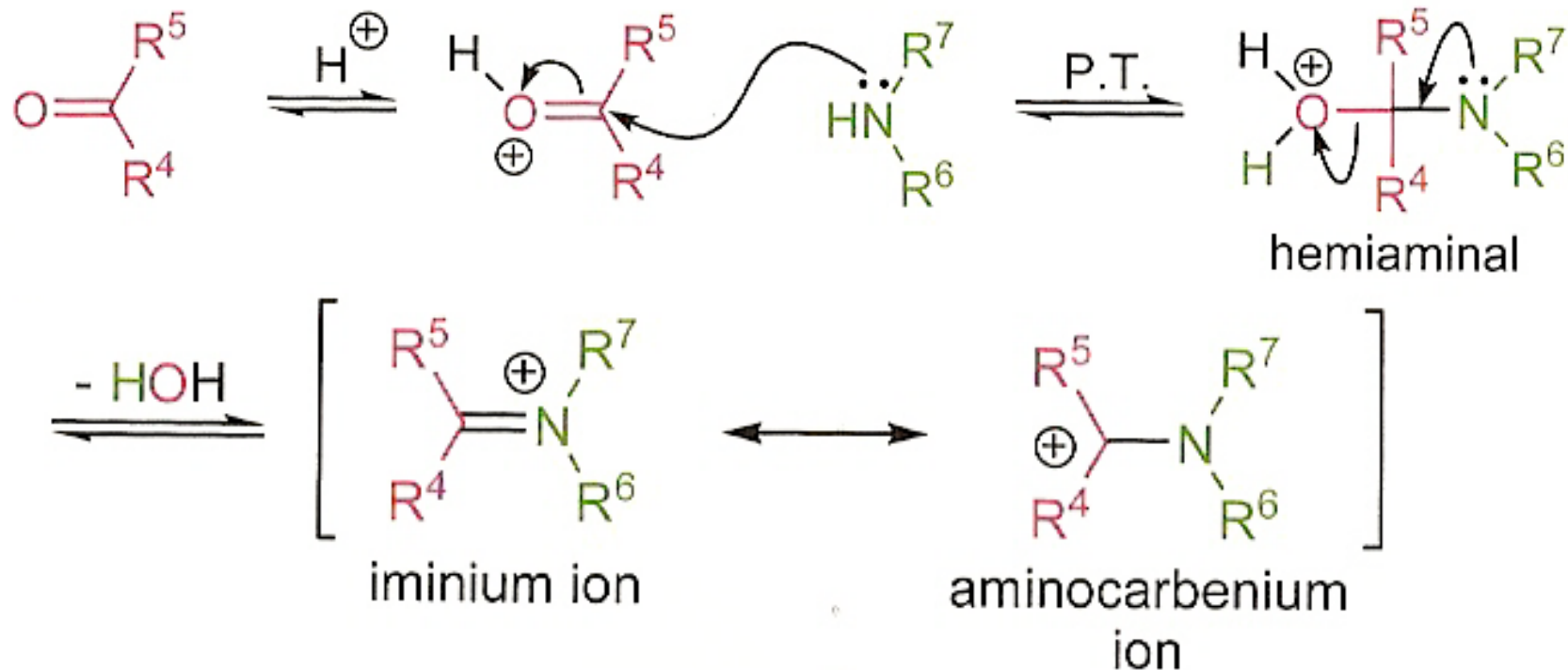
Multi-Component Condensation

- Nonenolizable Carbonyl (usually Aldehyde)
- Amine (1° or 2°)
- Enolizable carbonyl
- Acid (Usual) or Base Catalyzed
- Product: β -Amino Carbonyl Derivatives



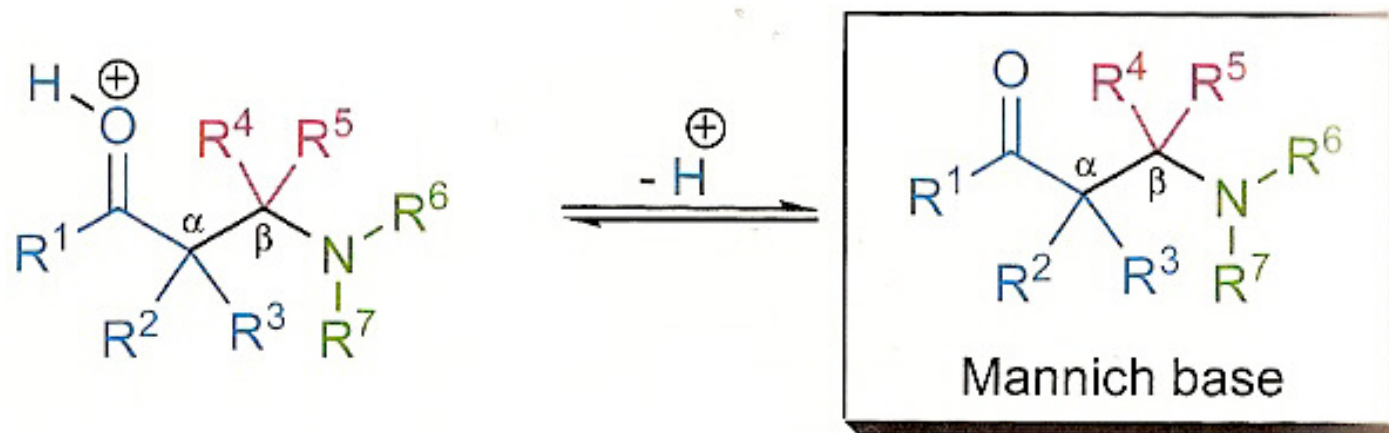
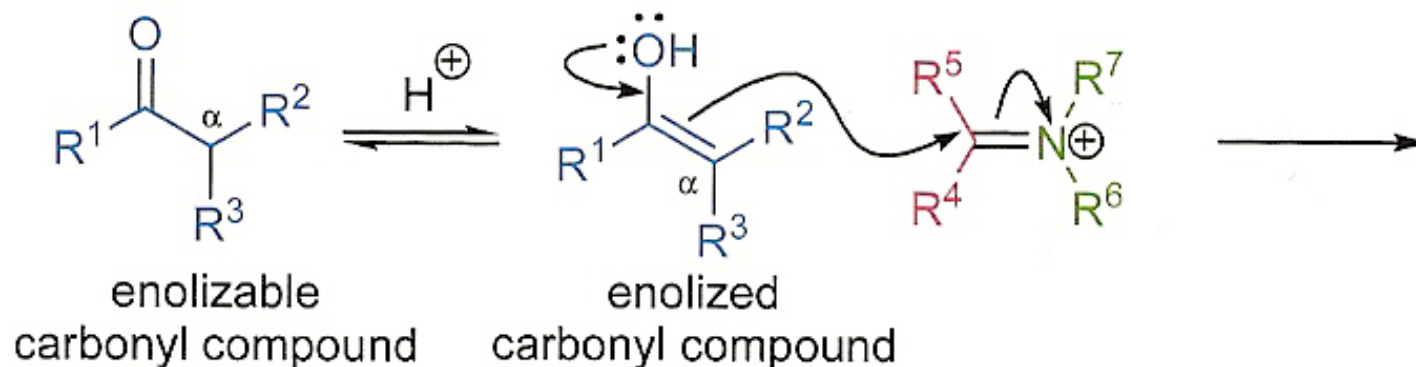
Mannich Reaction: Mechanism

Formation of the reactive iminium ion under acidic conditions:



Mannich Reaction: Mechanism

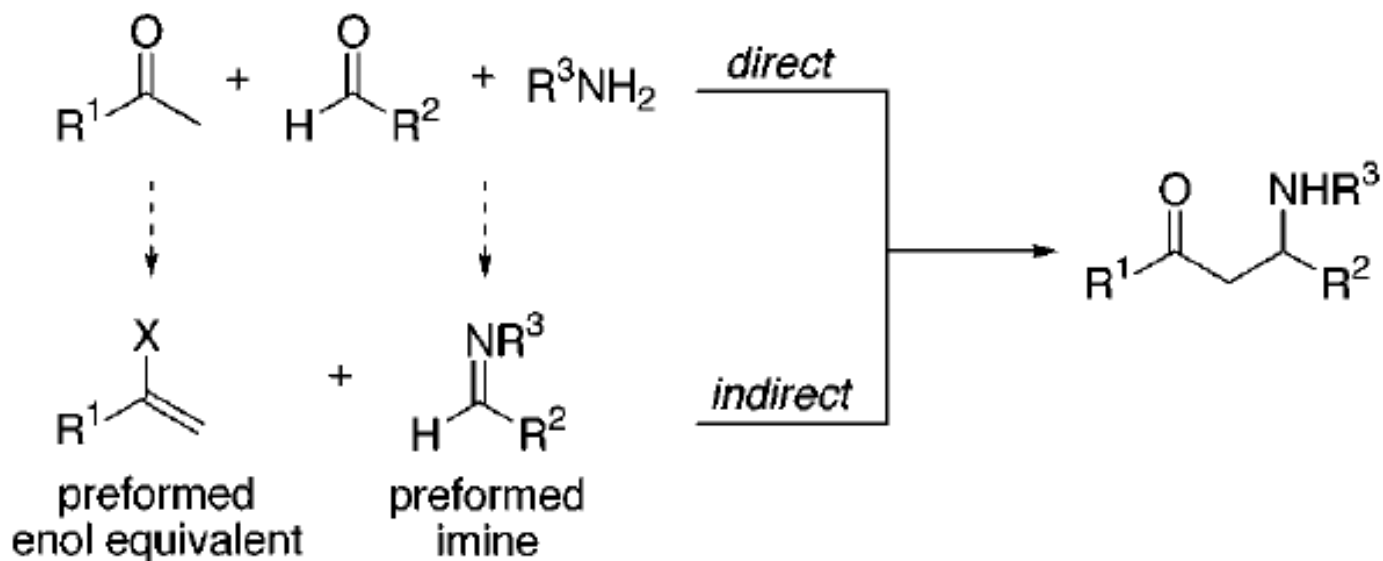
Alkylation of the enolized carbonyl compound:



Asymmetric Mannich Reactions

6

Mannich Reaction: Direct vs. Indirect



Direct:

- Components in Equilibrium
- Limitations of Substrates

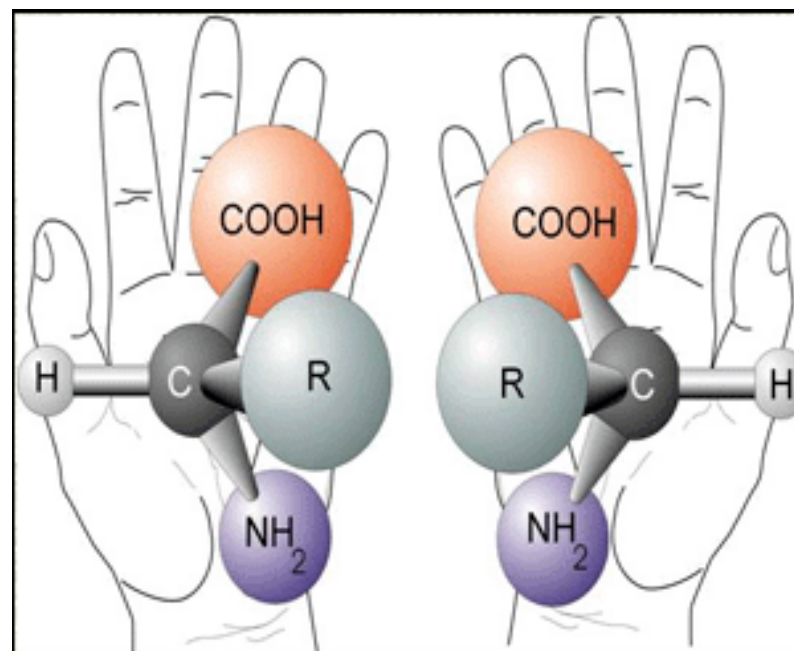
Indirect:

- Assign Chemical Roles
- Additional Steps

Mannich Reaction: New Twists

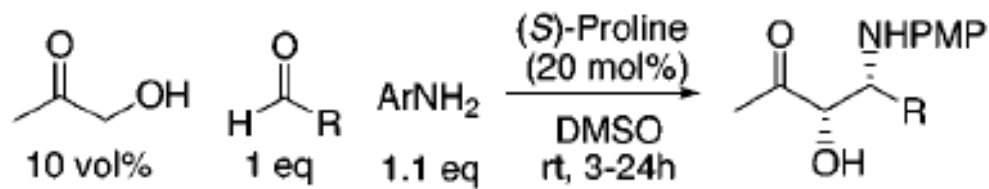
Enantioselective Catalysis

- Preformed Imine and Enols with Metals
- Transition Metals
- Proline



Source: <http://web99.arc.nasa.gov/~astrochm/aachiral.html>

Mannich Reaction: Proline

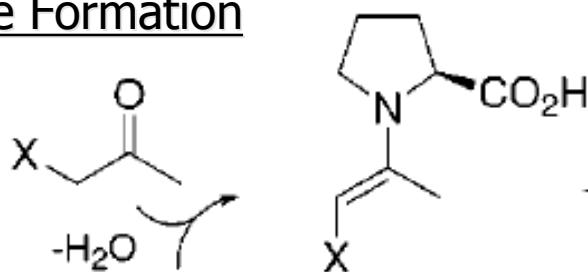


Entry	Product	Yield %	dr	ee %
1		92	20:1	>99
2		88	15:1	99
3		90	15:1	98

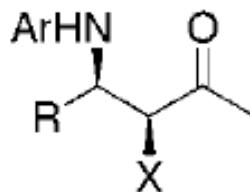
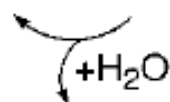
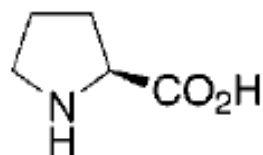
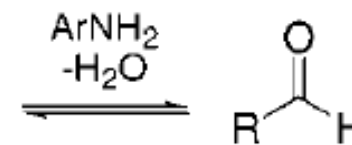
- Excellent dr's and ee's
- Exclusive syn diastereomer
- Other substrates tested

Mechanism of the Proline Reaction

Enamine Formation



Imine Formation



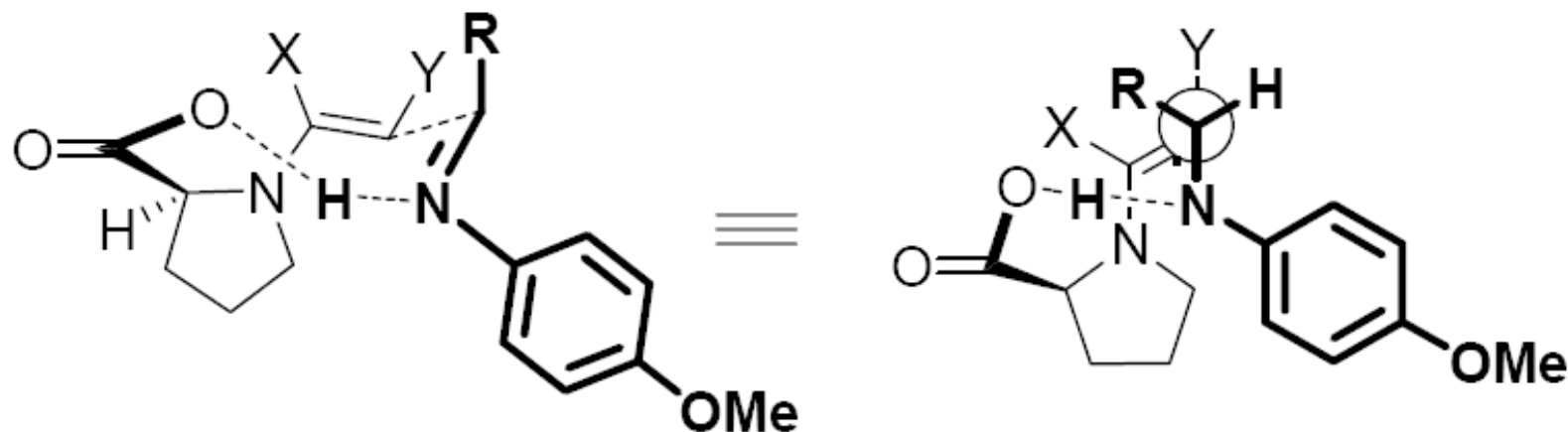
Mannich Base

Asymmetric Mannich Reactions

10

TS of the Proline Reaction

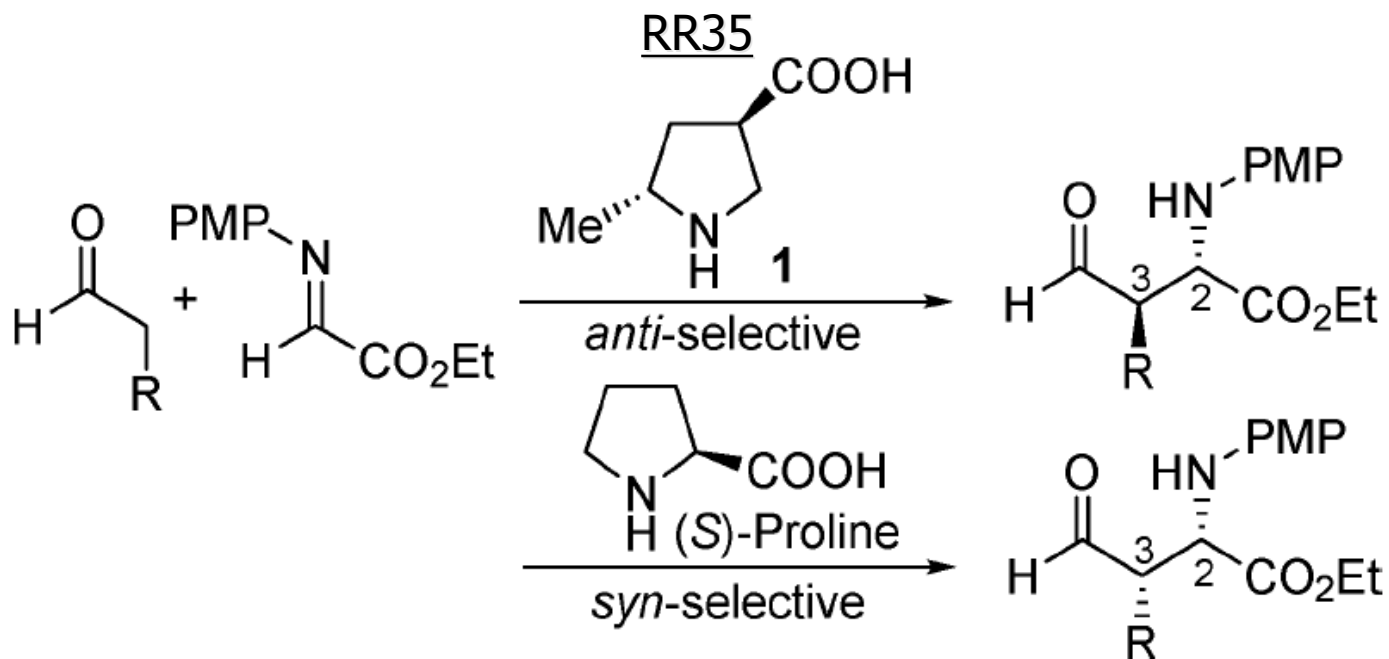
- Formation of the (E)-Aldimine
- Cyclic TS creates syn stereoselectivity through addition to Si face of Aldimine
- (R)-Proline give syn enantiomer



Source: www.scs.uiuc.edu/chem/gradprogram/chem435/fall05/Hoyt_Mirth.pdf

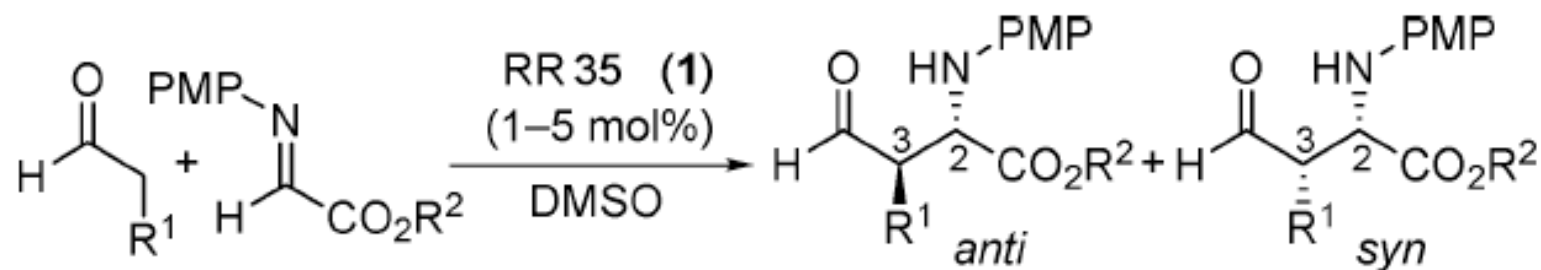
A Designed Catalyst

- Derivative of Proline offers Anti Diastereomers



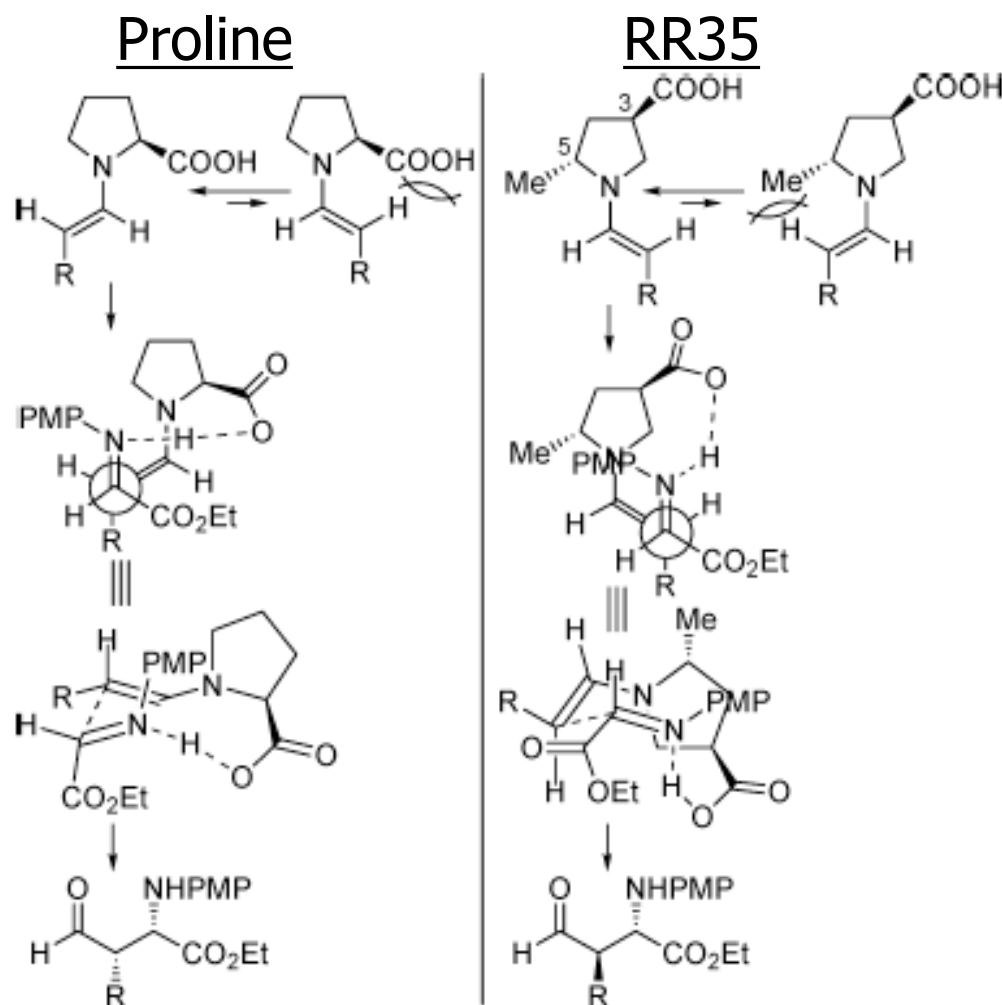
RR35 = (3R,5R)-5-methyl-3-pyrrolidinecarboxylic acid

Performance of the Catalyst



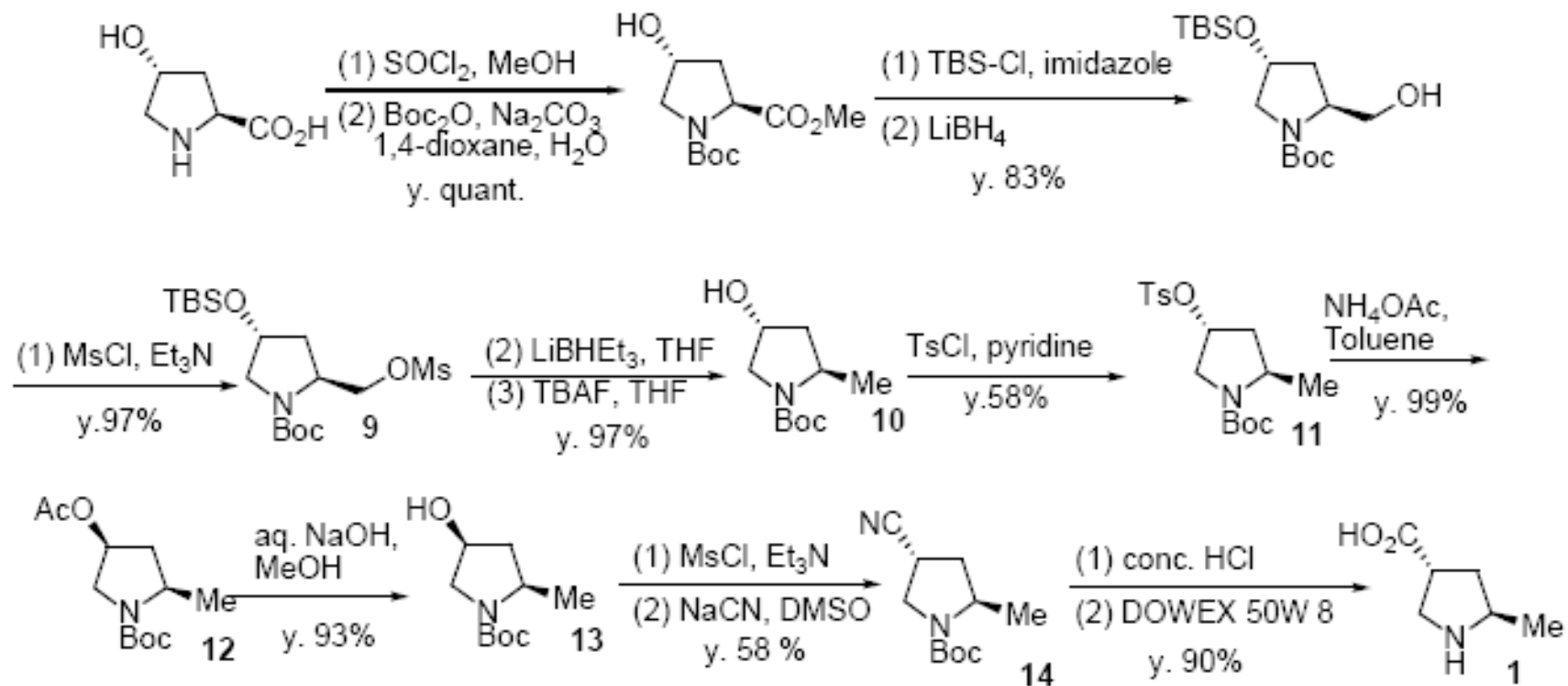
entry	R ¹	R ²	time (h)	product	yield (%)	dr ^b anti:syn	ee ^c (%)
1 ^d	Me	Me	-	—	—	95:5	98
2	Me	Et	1	2	70	94:6	>99 ^e
3	<i>i</i> -Pr	Et	3	3	85	98:2	99
4	<i>n</i> -Bu	Et	0.5	4	54	97:3	99
5 ^{f,g}	<i>n</i> -Bu	Et	1	4	71	97:3	99
6 ^{f,h}	<i>n</i> -Bu	Et	2	4	57	97:3	>99

Anti-Selective Mechanism



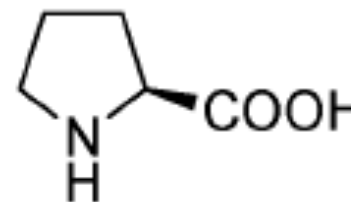
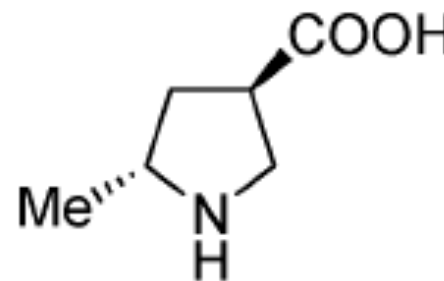
- s-trans conformation of enamine
- Placement of methyl group

Synthesis of the Catalyst



Future of the Asymmetric Mannich

- Applications and Extensions of this Paper
- Rational Design of Organocatalysts



Conclusion

- I. Classic Mannich
- II. State of the Art
- III. Current Work
- IV. Further Extensions



References

1. S Mitsumori, H Zhang, P Cheong, K. N. Houk, F Tanaka, and C Barbas, III. *J. Am. Chem. Soc.* **2006**, *128*, 1041
2. B List, *Tetrahedron*, **2002**, *58*, 5573
3. B List, P Pojarliev, W Biller, and H Martin, *J. Am. Chem. Soc.* **2002**, *124*, 827
4. M Arend, B Westermann, and N Risch, *Angew. Chem Int. Ed.*, **1998**, *37*, 1044
5. B List, *J. Am. Chem. Soc.*, **2000**, *122*, 9336
6. "Strategic Applications of Named Reactions in Organic Synthesis", L Kurti, B. Czako, Elsevier, St. Louis, MO, **2005**