# An Improved Asymmetric Reformatsky Reaction Mediated by (-)-N,N-Dimethylaminoisoborneol

Organic Letters, **2006**, 8, 1125 Ralf J. Kloetzing, Tobias Thaler, and Paul Knochel

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Journal Club Presentation
4/10/2006

## **Outline**

- Background
- Features and Mechanism
- Extensions to the Reformatsky Reaction
- Improved Asymmetric Reformatsky Reaction
- Conclusions and Next Directions

# **Reformatsky Reaction**

- Discovered by S. Reformatsky in 1887
- The classical reformatsky involved a zinc-induced reaction between an  $\alpha$ -halo ester and an aldehyde or ketone.



$$XCH_{2}COOEt \xrightarrow{Zn} X-Zn-CH_{2}COOEt \xrightarrow{R_{2}} X-Zn-CH_{2}COOEt$$

$$X = X - Zn - O$$

$$R_{1} = C$$

$$CH_{2}COOEt \xrightarrow{R_{2}} CH_{2}COOEt$$

$$R_{2} = C$$

$$R_{2} = C$$

$$R_{3} = C$$

$$CH_{2}COOEt$$

"Reformatsky reactions are defined as those resulting from metal insertions into carbon-halogen bonds activated by carbonyl, carbonyl derived or carbonyl related groups in vicinal or vinylogous positions with practically all kinds of electrophiles."

## **Mechanism**

$$R^{2} \longrightarrow R^{2} \longrightarrow R^{3} \longrightarrow R^{4} \longrightarrow R^{4$$

## **Key Features**

#### **Features:**

- One Pot Synthesis
- Carried out in ether solvents or ether/polar solvent mixtures (CH<sub>3</sub>CN, DMF, DMSO, HMPA)
- Mild conditions
- high stereoselectivity
- •Use of activated metal reagents or low-valent metal salts
  - Activated Metals:
  - Zn, Li, Mg, Cd, Ba, In, Ge, Ni, Co, Ce
  - •Metal Salts:
    - CrCl<sub>2</sub>, SmCl<sub>2</sub>, TiCl<sub>2</sub>

## **Synthetic Utility**:

β-hydroxyesters are valuable precursors in natural products synthesis and pharmaceuticals

#### **Advantages over the Classic Aldol:**

- •Mild/Neutral Conditions
- •Works for sterically crowded ketones
- Ester enolate can be formed in presence of highly enolizable aldehyde and ketone functionalities
- Intra- and Inter- molecular reactions feasible
- •Variety of Electrophiles can be used

# The Reformatsky using Chiral Auxiliaries

• Diastereoselective formation of Fluroazetidones via a reformatsky-type reaction of Ethyl Bromodifluoroacetate with Chiral 1,3-Oxazolidines

### Why Fluorine?

- Small VDW radius and electronegativity has significant effect on physical and chemical properties of molecules
- gem-difluoro amino acids and derivatives are valuable tools for the inhibition of metabolic processes
  - •Difluoroazetidin-2-ones exhibit specific properties as inhibitors of Leucocyte Elastase

# The Reformatsky using Chiral Ligands

• Achieved highly enantioselective Reformatsky reaction of Ketones using cinchona alkaloids as chiral ligands

R 
$$\rightarrow$$
 BrZn  $\rightarrow$  Ot-Bu  $\rightarrow$  Ch<sub>2</sub>CO<sub>2</sub>t-Bu  $\rightarrow$  Pyridine, THF  $\rightarrow$  Tr up to 97 % ee

• Chelation with the sp<sup>2</sup>-nitrogen adjacent to the reactive carbonyl controlled the facial selectivity to give the chiral alcohol

## An Efficient Route to β-amino Esters

• Enantioselective one-pot imino-Reformatsky reaction of 4-chlorobenzaldehyde with 2-methoxy-aniline and bromoacetates

#### **Key Features**

- one-pot synthesis
- catalytic
- enantioselective
- highly efficient, practical
- ligand can be recovered
- Gave  $\beta$ -amino esters in good yield

### Development of an Improved Asymmetric Reformatsky

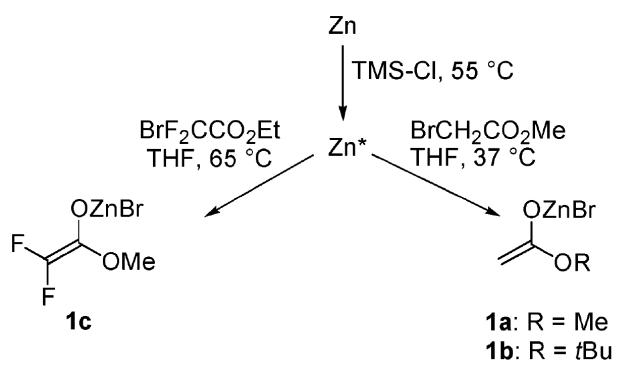
**Problem:** The direct Reformatsky reaction via insertion of activated zinc into  $\alpha$ -bromoacetates lack generality and have low enantioselectivities

**Solution:** Utilization of (-) DAIB as a ligand achieved good enantioselective additions of Reformatsky Reagents into aliphatic and aromatic aldehydes.

# **Synthesis of Reformatsky Reagents**

•Reformatsky reagents prepared via direct zinc insertion into bromoacetic esters

Scheme 1. Preparation of Reformatsky Reagents 1a-c



## **Tested Various Chiral Ligands**

# • Examined amino alcohols that give high asymmetric inductions as chiral ligands for the addition of diethylzinc into aldehydes

**Table 1.** Reformatsky Reaction with Various Chiral Amino Alcohols

entry	amino alcohol	R	yield (%)	$ee^d$ (%)
1	2	Me	$99^b$	61
2	3	Me	$99^b$	9
3	4	Me	$99^b$	26
4	5	Me	$75^c$	86
5	5	<i>t</i> Bu	$84^b$	78
6	6	Me	$93^b$	72

<sup>a</sup> All reactions were performed on a 0.5 mmol scale using amino alcohol (1.2 equiv, 0.6 equiv in case of **2**), Et<sub>2</sub>Zn (0.7 equiv), and Reformatsky reagent (1.1 equiv). <sup>b</sup> Conversion determined by GC analysis with tetradecane as internal standard. <sup>c</sup> Isolated yield. <sup>d</sup> Determined by GC analysis (Chiraldex CB).

\*\* BuLi and MeMgCl used as deprotonating agents led to racemic mixtures or very poor ee

# **Examined Substrate Scope**

**Table 2.** Reformatsky Reaction with (–)-DAIB and Various Aldehydes<sup>a</sup>

entry	R	product	conv (%)	$yield^b$ $(\%)$	ee <sup>c,d</sup> (%)
1	4-bromophenyl	<b>7</b> b	79	87	81
2	2-bromophenyl	7c	71	91	78
3	4-chlorophenyl	7d	85	65	80
4	4-cyanophenyl	<b>7e</b>	89	88	72
5	4-isopropylphenyl	<b>7</b> f	63	90	88
6	$4 ext{-methylthiophenyl}^e$	7g	57	93	93
7	3-benzothienyl	<b>7h</b>	62	94	90
8	2-thienyl	7i	86	85	92
9	3-thienyl	<b>7</b> j	88	87	93
10	2-furyl	7k	99	86	84
11	neopentyl	71	77	83	92
12	n-pentyl	<b>7m</b>	64	88	78
13	2,2-dimethyl-2-methoxyethyl	<b>7</b> n	61	85	74
14	benzyloxymethyl	<b>7o</b>	79	87	66
15	2-phenylvinyl	7p	48	92	71

<sup>a</sup> See Table 1. <sup>b</sup> Isolated yield based on conversion. <sup>c</sup> Determined by chiral GC (Chiraldex CB) or chiral HPLC (Chiralcel OD-H). <sup>d</sup> The absolute stereochemistry (*S*) was assigned by comparison of the optical rotation with literature data. <sup>4e</sup>, For a procedure, see ref 17.

**4-methylthiophenyl Derivative** = intermediate in Synthesis of Duloxetine, Inhibitor of Serotonin and Norephedrine uptake carriers

#### **Electron Rich vs Poor**

- e- rich increase ee, lower % conversion
- Sulfur containing
  - ee enhanced regardless of position on substrate
- Aliphatic aldehydes
- more hindered substrates, greater increase in ee

# **Substrate Scope for Fluorinated Reagent**

- Observed that electron poor aldehydes had an increase in ee with the fluorinated reagent
- Position of the sulfur atom has significant effect on ee

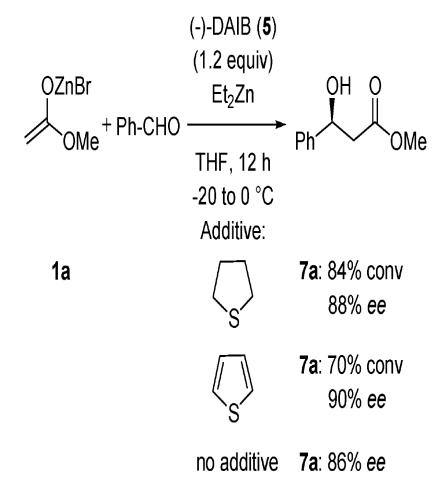
• Also selective for Sterically hindered, aliphatic aldehydes **Table 3.** Reformatsky Reaction with the Difluoro Reagent **1c**, (–)-DAIB and Various Aldehydes

entry	R	product	conv (%)	$\begin{array}{c} {\rm yield}^b \\ (\%) \end{array}$	ee <sup>c,d</sup> (%)
1	phenyl	8a	92	82	88
2	4-bromophenyl	<b>8b</b>	96	83	87
3	4-cyanophenyl	8 <b>c</b>	91	88	84
4	2-thienyl	<b>8d</b>	66	90	90
5	3-thienyl	<b>8e</b>	69	84	87
6	neopentyl	<b>8f</b>	91	81	80

 $^{a-c}$ See Table 2.  $^d$  The absolute stereochemistry (R) was assigned by comparison of the optical rotation with literature data.  $^{14a}$ 

## **Effects of Sulfur Additives**

**Scheme 2.** Reformatsky Reaction Using Sulfur Additives



· Sulfur containing compounds used as additives showed an increase in the enantioselectivity

### **Conclusions**

- (-) DAIB is an excellent ligand for the enantioselective addition of the Reformatsky Reagent in to various aldehydes
- High enantioselectivities were obtained for thiophene aldehydes or sterically hindered aliphatic aldehydes
- Addition of the difluoro zinc reagent also proceeded with high enantioselectivity
- The presence of sulfur additives had a significant impact on the selectivity of these reactions

### **Next Directions**

- Investigate the role of sulfur on the selectivity of these reactions and its involvement with regards to the mechanism for the reaction
- •Explore the effects of sulfur additives in fluoroderived substrates
- •Can this ligand be used in catalytic application?
- •Explore the benefit of using different metals or metal salts other than Zn

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