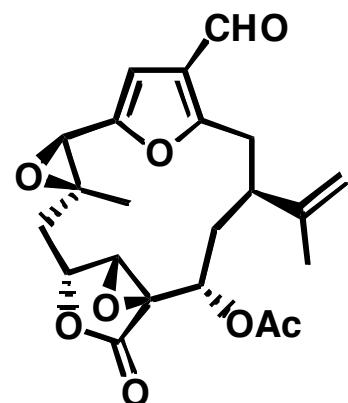


# *Progress towards the Total Synthesis of Lophotoxin*



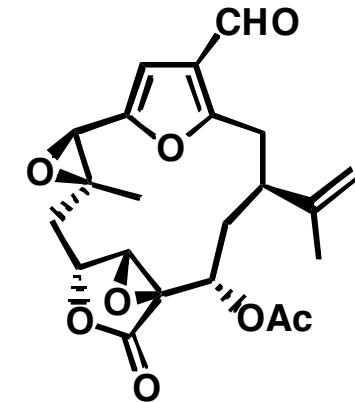
*Michel Grenon  
July 10<sup>th</sup>, 2004*

# *Presentation Outline*

## 👉 Cell signaling between neurons

*Mechanism*

*Neurotransmitters and their protein receptors (ionotropic)*



## 👉 Lophotoxin

*Isolation and structural features*

*Other members of the furanocembranolides*

*Bioactivity*

## 👉 Lophotoxin (Synthetic work)

*Synthetic work from other groups and previous work done by Pr. Wipf's research group*

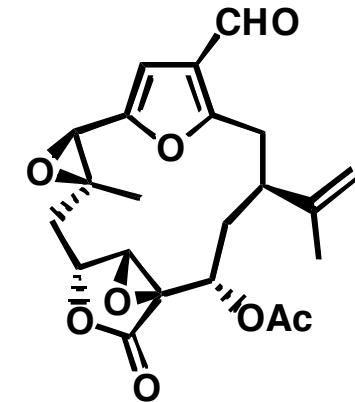
*Current work done in the group*

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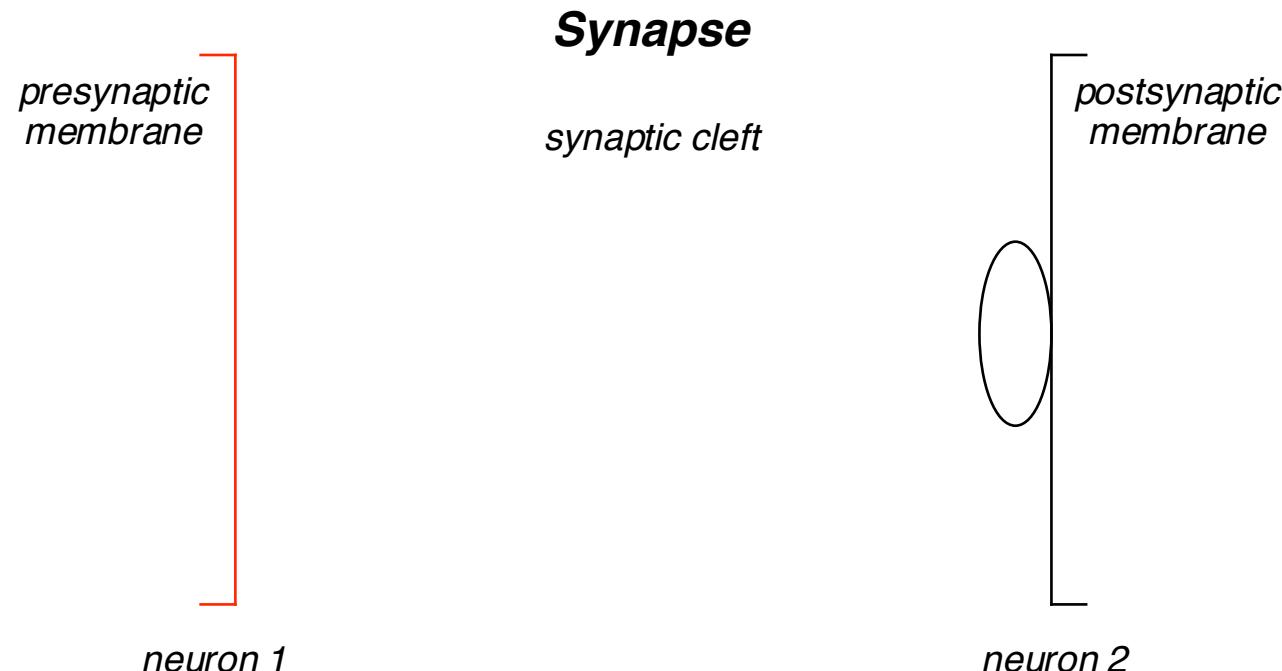
*Current work done in the group*

# ***Mechanism of Cell Signaling in the Central Nervous System***

**The Central Nervous System (CNS) is composed of billions of neurons**

The correct functioning of the CNS is based on the generation, propagation and coordinated integration of signals between different neurons

**The communication between nerve cells is performed at a highly specialized region called the *Synapse***

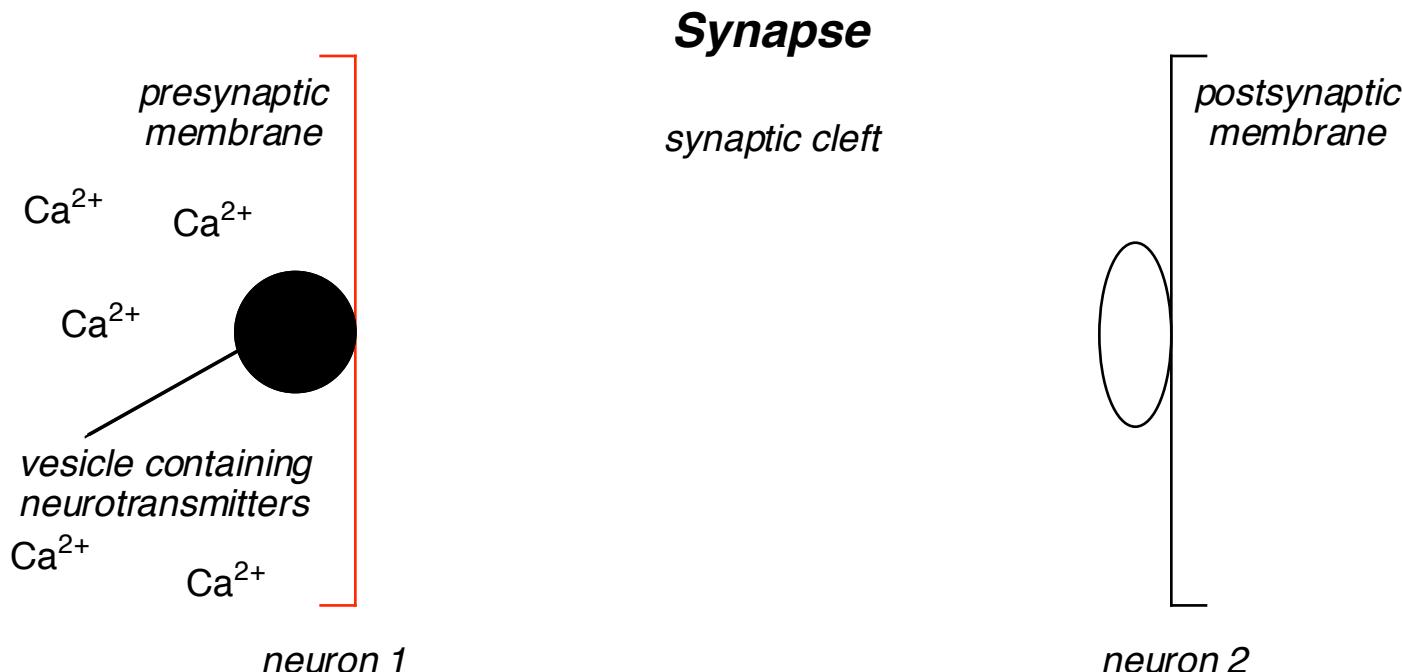


# ***Mechanism of Cell Signaling in the Central Nervous System***

**The transfer of information is achieved chemically by molecules called neurotransmitters**

Membrane depolarization results in enhancement of  $\text{Ca}^{2+}$  permeation

At raised intracellular  $\text{Ca}^{2+}$  concentrations, vesicles containing neurotransmitters fuse with the presynaptic membrane

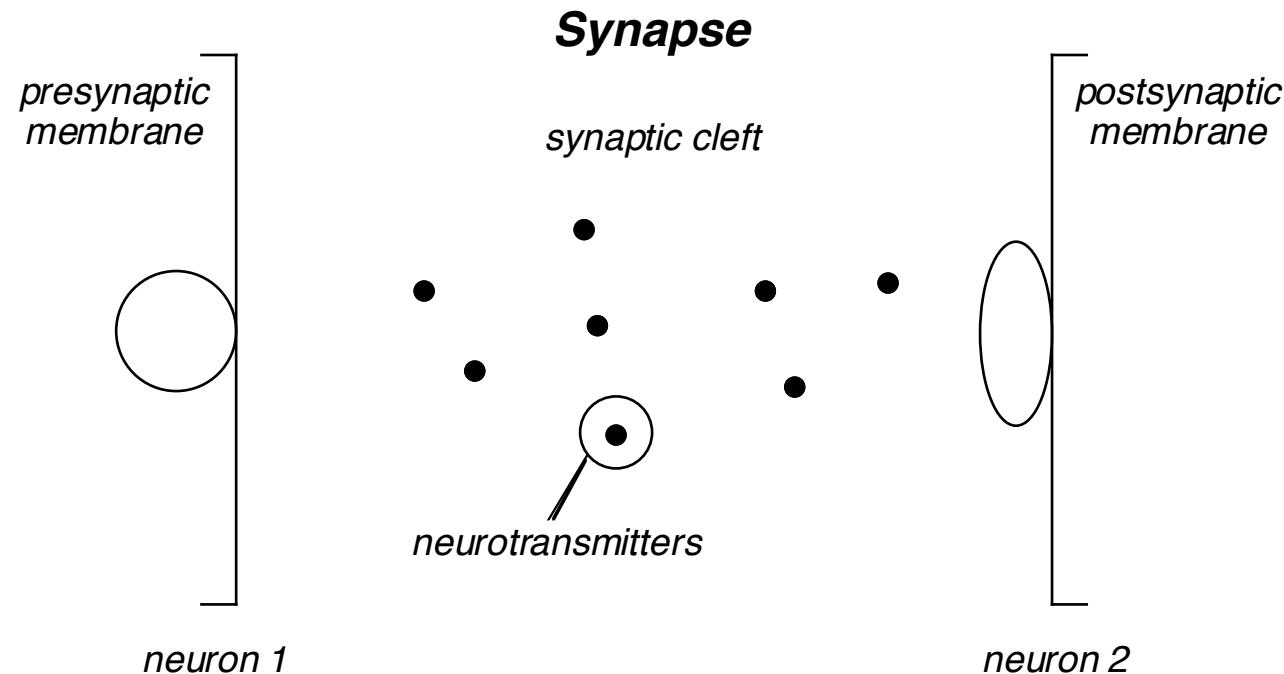


# ***Mechanism of Cell Signaling in the Central Nervous System***

**The vesicle content is spilled into the synaptic cleft**

The time course for neurotransmitter clearance is between 0.1 and 2.0 ms

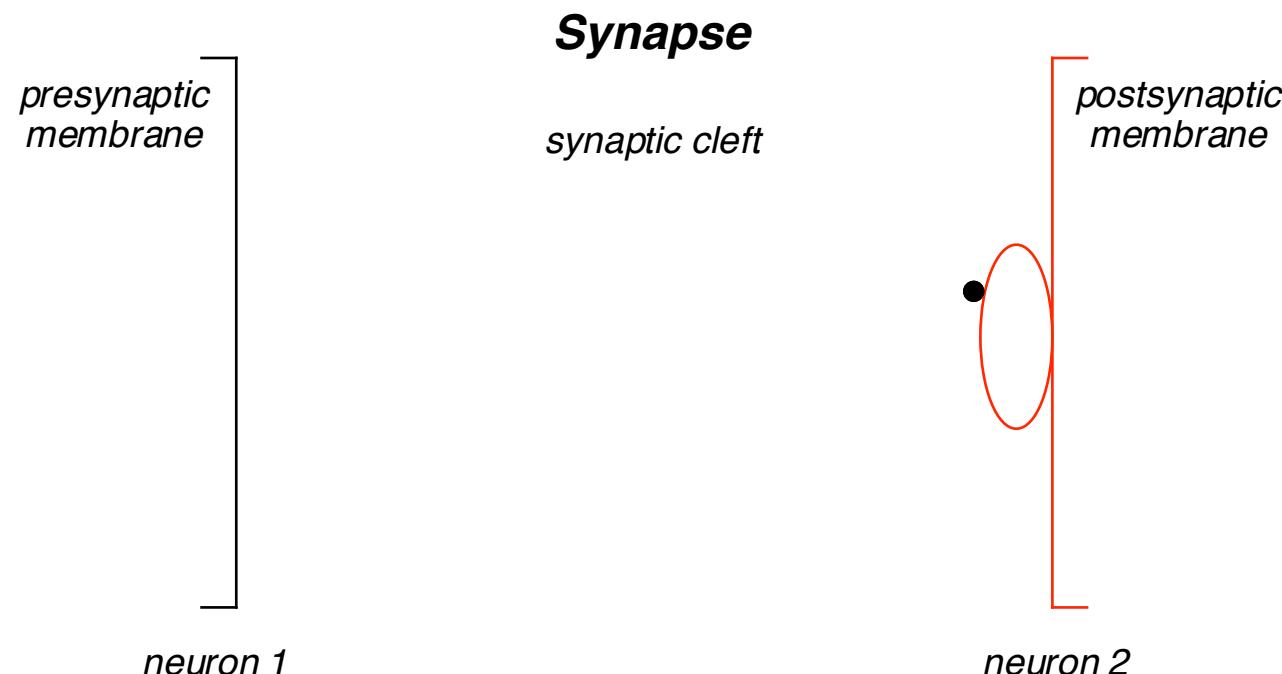
The neurotransmitter molecules diffuse through the synaptic space in less than 0.2 ms, reaching concentrations of 1-5 mM



# ***Mechanism of Cell Signaling in the Central Nervous System***

**The chemical information is converted into electrical currents on the postsynaptic membrane**

This latter membrane is highly specialized in the recognition and binding of neurotransmitters by means of protein receptors

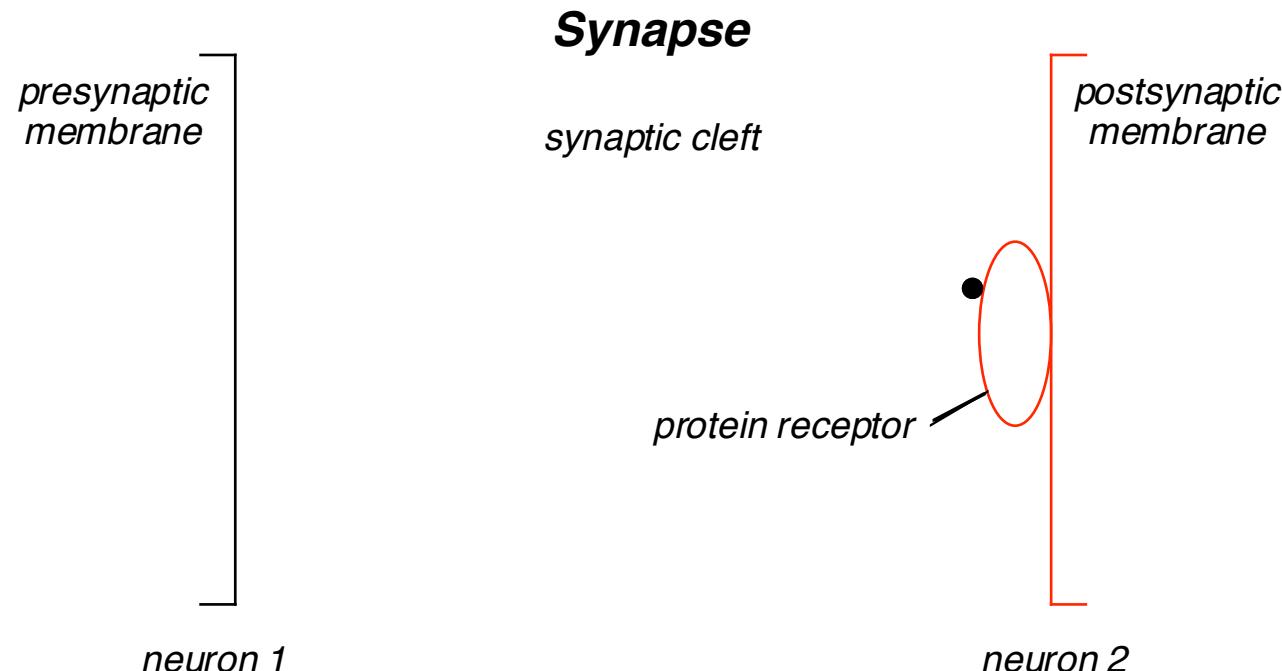


# ***Mechanism of Cell Signaling in the Central Nervous System***

**The chemical information is converted into electrical currents on the postsynaptic membrane**

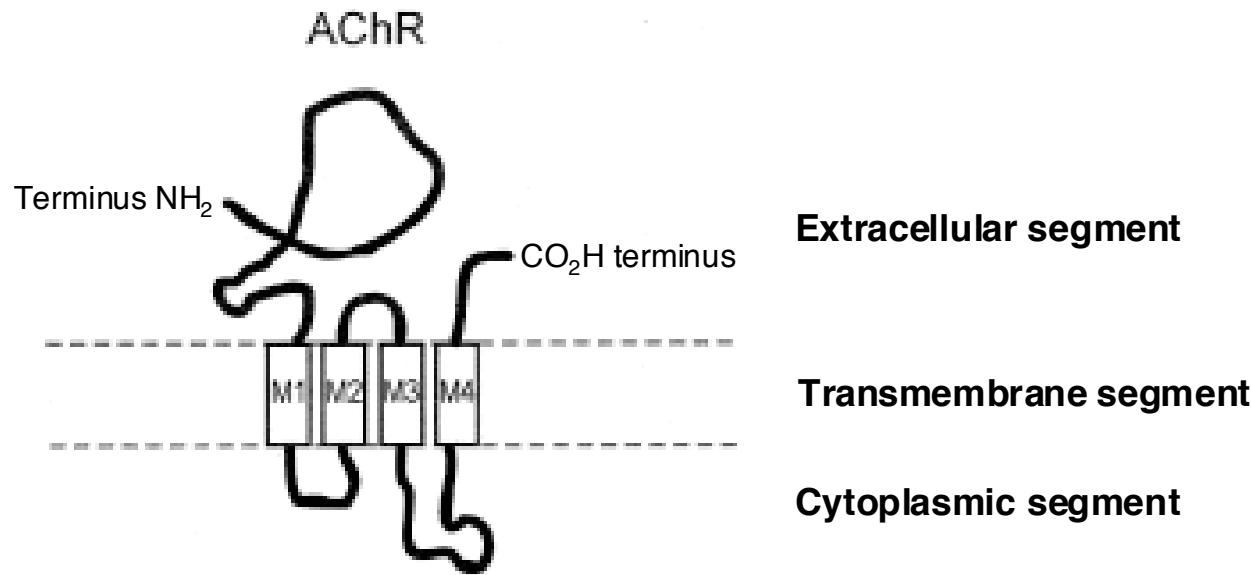
There are two main receptor classes; *ionotropic* and *metabotropic*

The binding of a neurotransmitter to its ionotropic receptor induces a fast opening of the ion channel ( $\text{Na}^+$ ,  $\text{K}^+$  or  $\text{Ca}^{2+}$ ) intrinsically coupled to the receptor



## *Example of ionotropic Receptor: Nicotinic Acetylcholine Receptor (AChR)*

*Diagram of the tertiary organization of AChR*

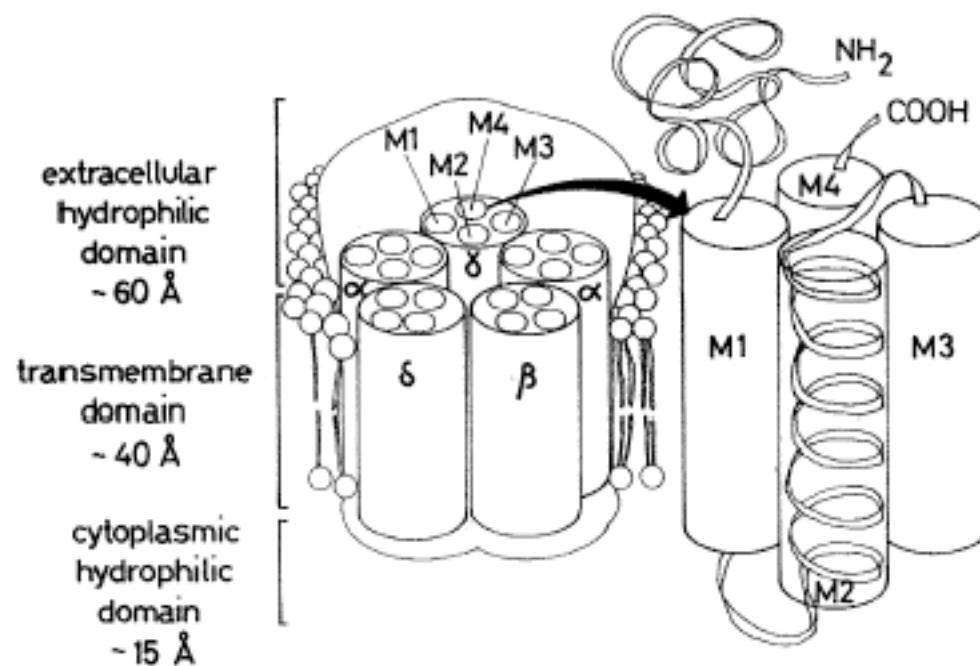


1. Long NH<sub>2</sub>-terminal hydrophilic extracellular region (210 A.A., 60 Å)
2. Four highly hydrophobic domains (**M1**, **M2**, **M3** and **M4**)
3. Major hydrophilic segment facing the cytoplasm in which the **M4** domain orients the CO<sub>2</sub>H-terminal towards the synaptic side of the membrane

Arias, H. R. *Neurochem. Int.* **2000**, *36*, 595

## *Example of Ionotropic Receptor: Nicotinic Acetylcholine Receptor (AChR)*

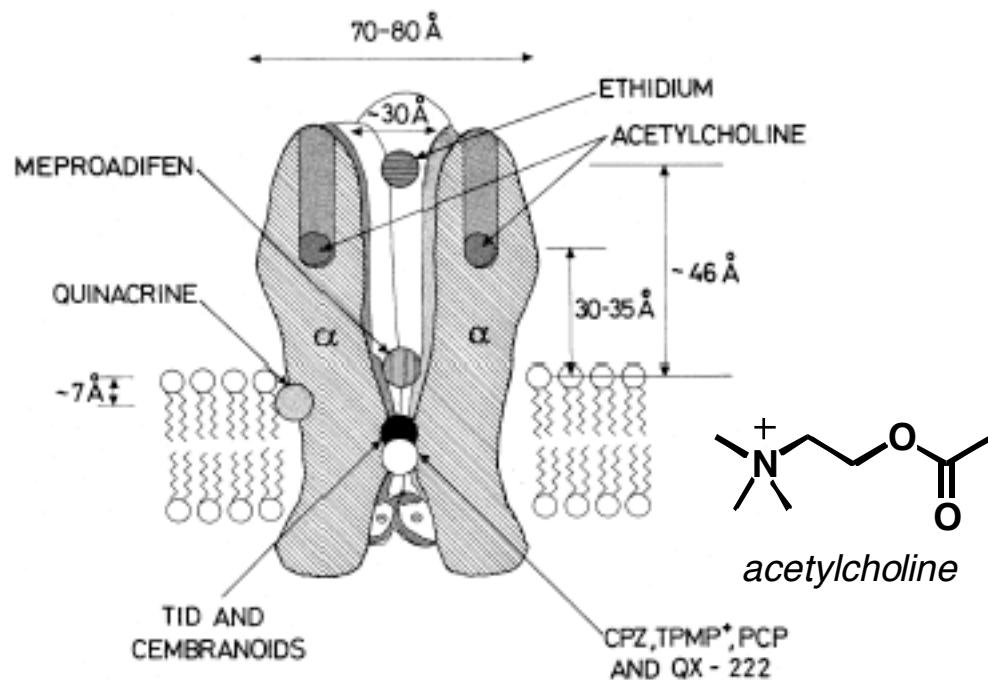
*Overall structure of the muscle-type AChR*



Arias, H. R. *Brain Res. Rev.* 1997, 25, 133

## *Example of Ionotropic Receptor: Nicotinic Acetylcholine Receptor (AChR)*

*Transverse schematic representation of the muscle-type AChR showing the most probable localisation of both acetylcholine and other ligands*



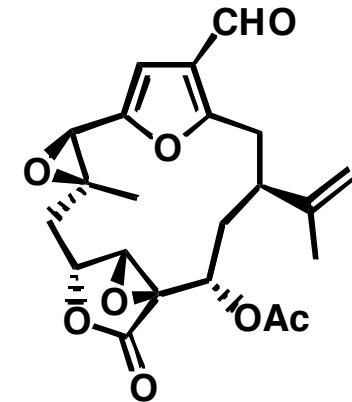
Arias, H. R. *Brain Res. Rev.* 1997, 25, 133

# *Presentation Outline*

## Cell signaling between neurons

*Mechanism*

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## Lophotoxin

*Isolation and structural features*

*Other members of the furanocembranolides*

*Bioactivity*

## Lophotoxin (Synthetic work)

*Synthetic work from other groups and previous work done by Pr. Wipf's research group*

*Current work done in the group*

## Lophotoxin

👉 First isolated<sup>1</sup> from Pacific Sea whips of the genus *Lophogorgia*

found mainly in tropical and subtropical waters (from Panama Bay northward to Point Conception, California)

relatively abundant (0.2% dry weight)

structure determined by spectral and chemical methods

👉 Produced in relatively large quantities by various species of gorgonian corals

758 g of freeze dried *Lophogorgia violacea* affords 163 mg (0.09%)

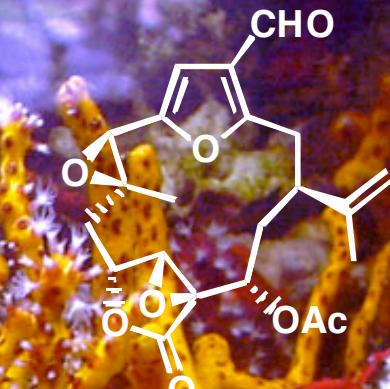
<sup>1</sup> Fenical, W.; Okuda, R. K.; Bandurraga, M. M.; Culver, P.; Jacobs, R. S. *Science* **1981**, *212*, 1512

### Structural Features

👉 Uncharged cyclic diterpene

👉 14-membered macrocycle incorporating a 2,3,5-trisubstituted furan, an epoxidized butenolide ring and a *trans*-trisubstituted epoxide

👉 5 stereogenic centers (2 epoxides, 3 chiral centers)

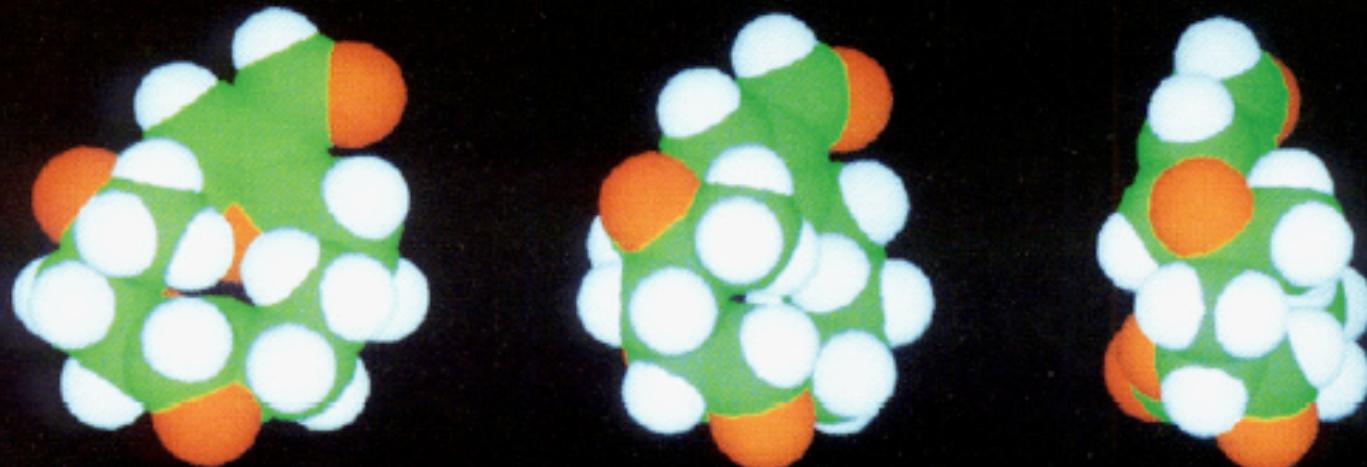


White needles

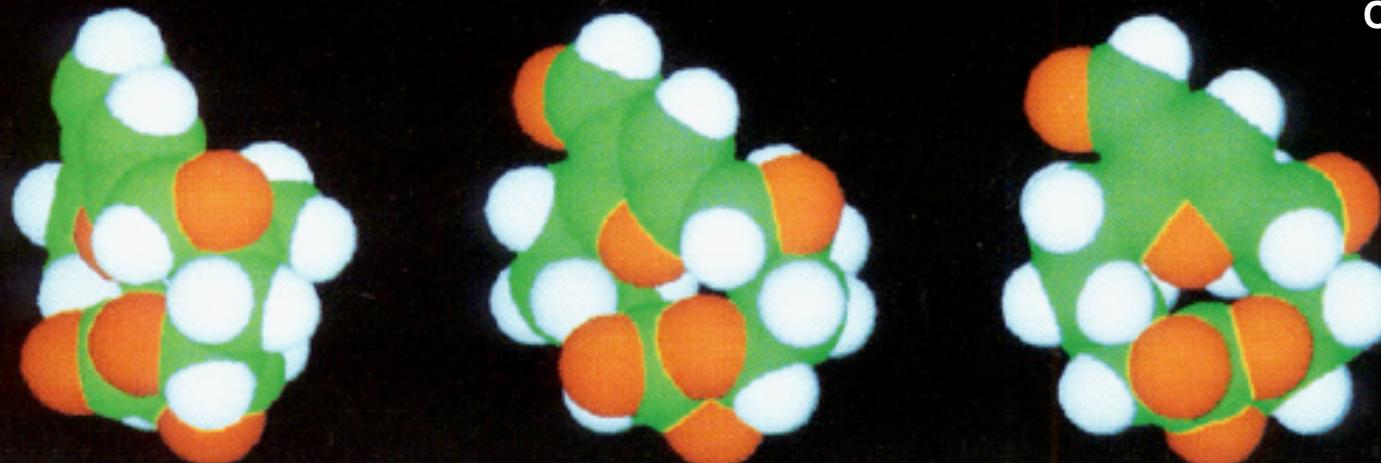
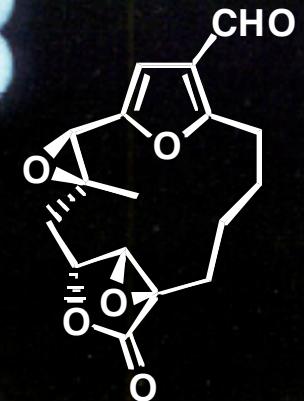
Mp 164–166 °C

$[\alpha]_D^{27} +14.2^\circ$  ( $c = 1.7$ ,  $\text{CHCl}_3$ )

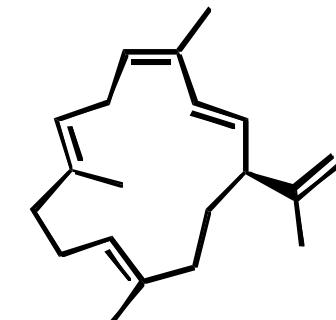
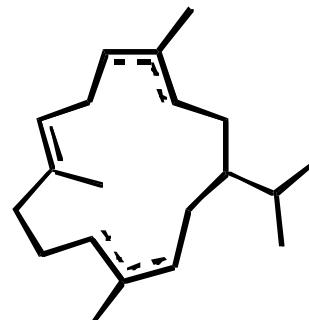
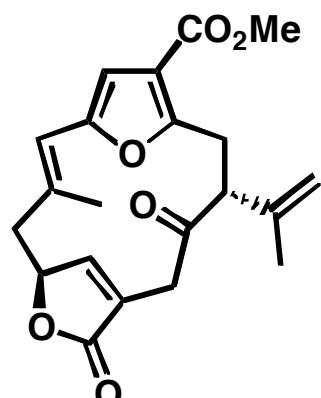
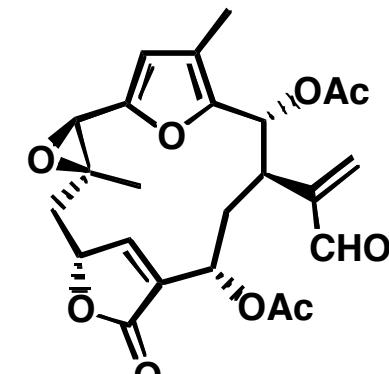
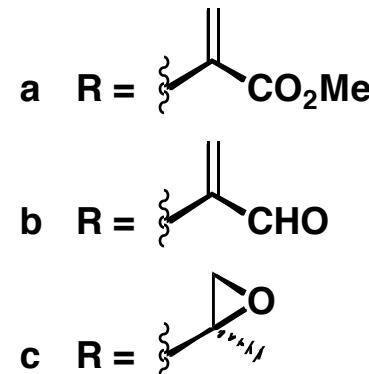
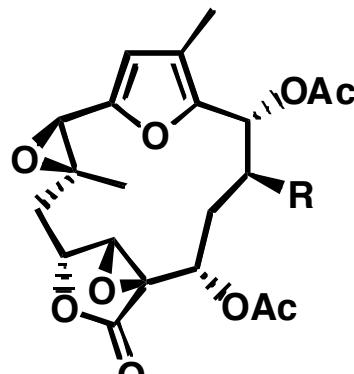
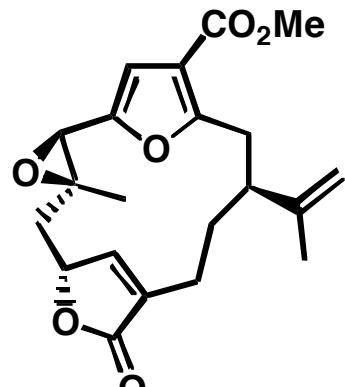
## Lophotoxin



The furan and lactone rings are oriented at *ca.* 90° angles to one another, with the lactone carbonyl pointing in the opposite direction as the C<sub>7</sub>-C<sub>8</sub> epoxide



## Other Members of the Furanocembranolides

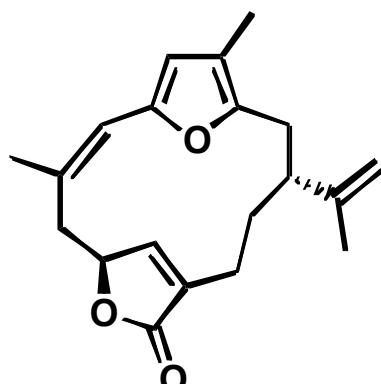


<sup>a</sup> Missakian, M. G.; Burreson, B. J.; Scheuer, P. J. *Tetrahedron* **1975**, *31*, 2513

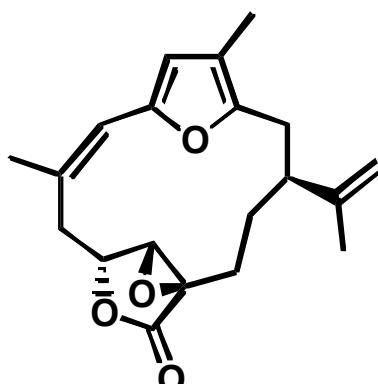
<sup>b</sup> Wright, A. E.; Burres, N. S.; Schulte, G. K. *Tetrahedron Lett.* **1989**, *30*, 3491

<sup>c</sup> Chan, W. R.; Tinto, W. F.; Laydoo, R. S.; Manchaud, P. S.; Reynolds, W. F.; McLean, S. *J. Org. Chem.* **1991**, *56*, 1773

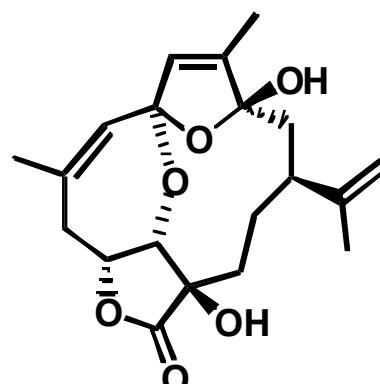
## *Other Members of the Furanocembranolides*



Rubifolide<sup>d</sup>



Coralloidolide A<sup>e</sup>



Coralloidolide B<sup>e</sup>

### *Total syntheses of Furanocembranes*

Bis-deoxylophotoxin, see; Cases, M.; de Turiso, F. G.-L.; Pattenden, G. *Synlett* **2001**, 1869

Deoxypukalide, see; Marshall, J. A.; Van Devender, E. A. *J. Org. Chem.* **2001**, 66, 8037

Rubifolide (enantiomer), see; Marshall, J. A.; Sehon, C. A. *J. Org. Chem.* **1997**, 62, 4313

Acerosolide (racemic), see; Paquette, L. A.; Astles, P. C. *J. Org. Chem.* **1993**, 58, 165

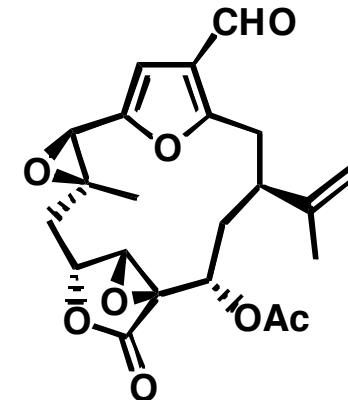
<sup>d</sup> Williams, D.; Andersen, R. J.; Van Duyne, G. D.; Clardy, J. *J. Org. Chem.* **1987**, 52, 332

<sup>e</sup> D'ambrosio, M.; Fabbri, D.; Guerriero, A.; Pietra, F. *Helv. Chim. Acta* **1987**, 70, 63

## Bioactivity of Lophotoxin

- 👉 Causes neuromuscular paralysis by inhibition of nicotinic acetylcholine receptors

Causes paralysis and asphyxiation (LD<sub>50</sub> in mice is 8.0 µg/g)



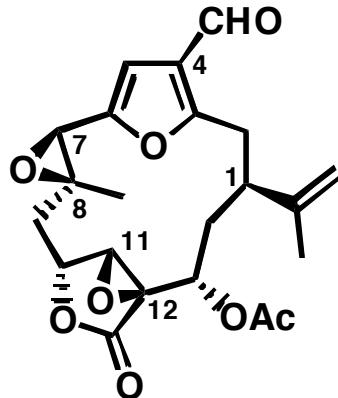
Lophotoxin acts as a competitive antagonist, by reacting covalently with the Tyr<sup>190</sup> residue (used as a probe to study the role of Tyr<sup>190</sup> in binding acetylcholine)

- 👉 Also inhibits neuronal and peripheral nicotinic acetylcholine receptors

- 👉 Lophotoxin and other furanocembranolides are responsible for the chemical defense displayed by the brazilian octocoral *Lophogorgia violacea* when submitted to feeding experiments to predatory fishes

Epifanio, R de A.; Maia, L. F.; Fenical, W. *J. Braz. Chem. Soc.* **2000**, 11, 584

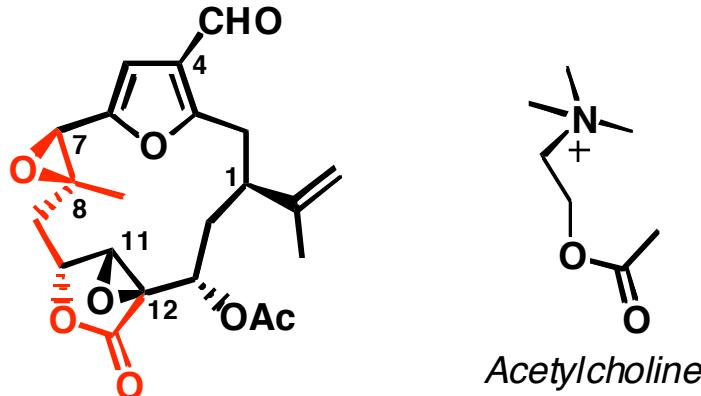
## *Structure/Activity Studies of the Lophotoxin Family*



- Substituents at C<sub>1</sub>, C<sub>2</sub> and C<sub>4</sub> are not essential to retain activity
- Epoxide at C<sub>11</sub>–C<sub>12</sub> is not critical either (however, the involvement of C<sub>11</sub> in the irreversible binding has not been ruled out)
- Epoxide at C<sub>7</sub>–C<sub>8</sub> is very important and is likely to be involved in the covalent reaction with the receptor
- C<sub>13</sub> acetate is *not absolutely required* (12-fold decrease in activity for the hydrolized derivative)
- Reduction of the lactone carbonyl to a cyclic hemiacetal produces a dramatic decrease in activity, which apparently results from a decrease in affinity for the recognition site

Abramson, S. N.; Trischman, J. A.; Tapiolas, D. M.; Harold, E. E.; Fenical, W.; Taylor, P.  
*J. Med. Chem.* **1991**, *34*, 1798

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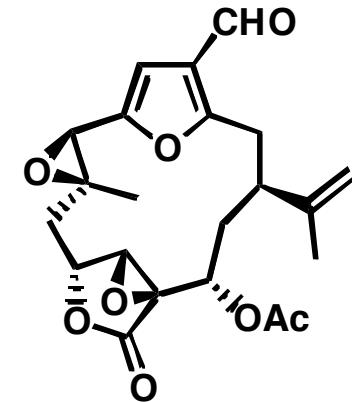
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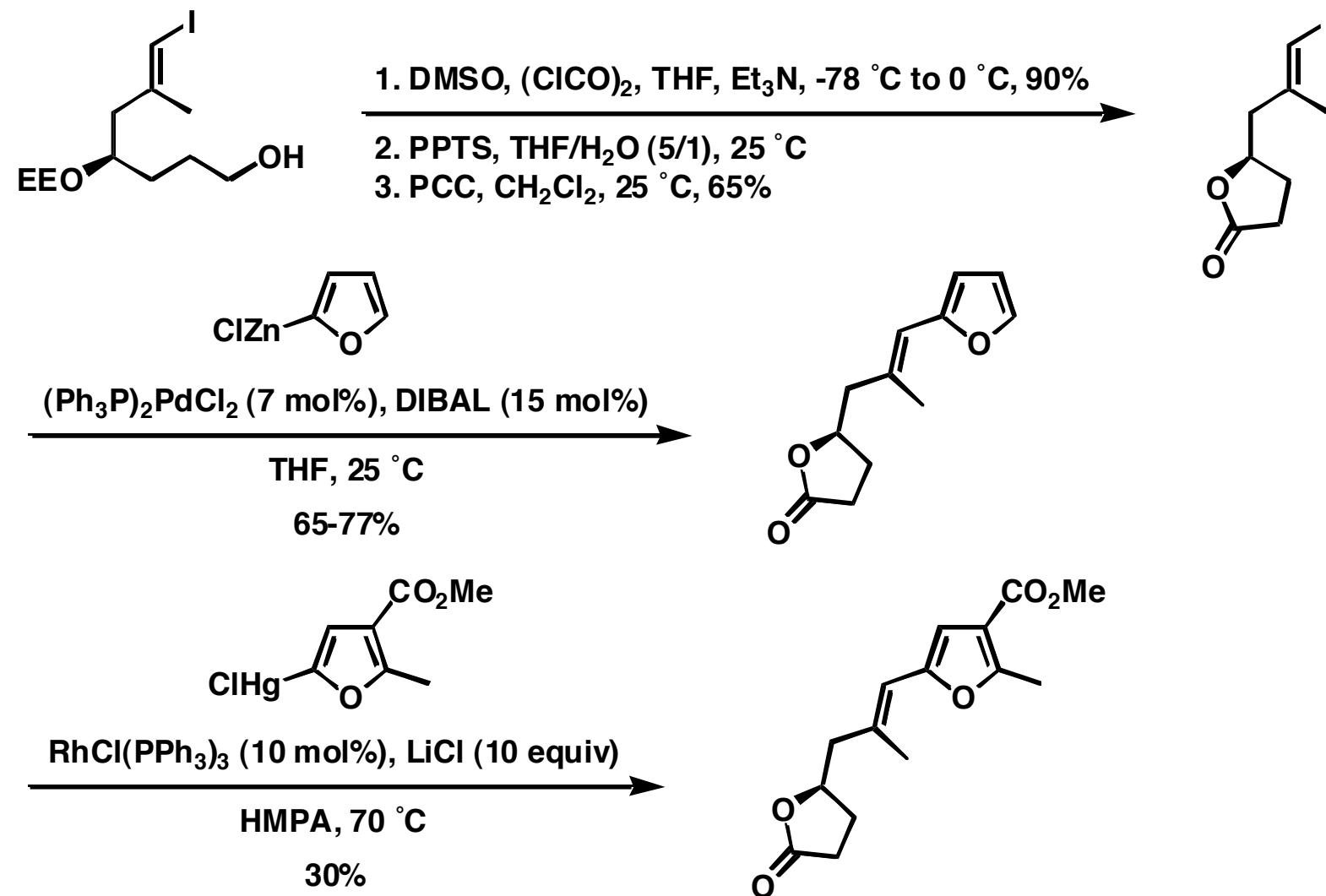
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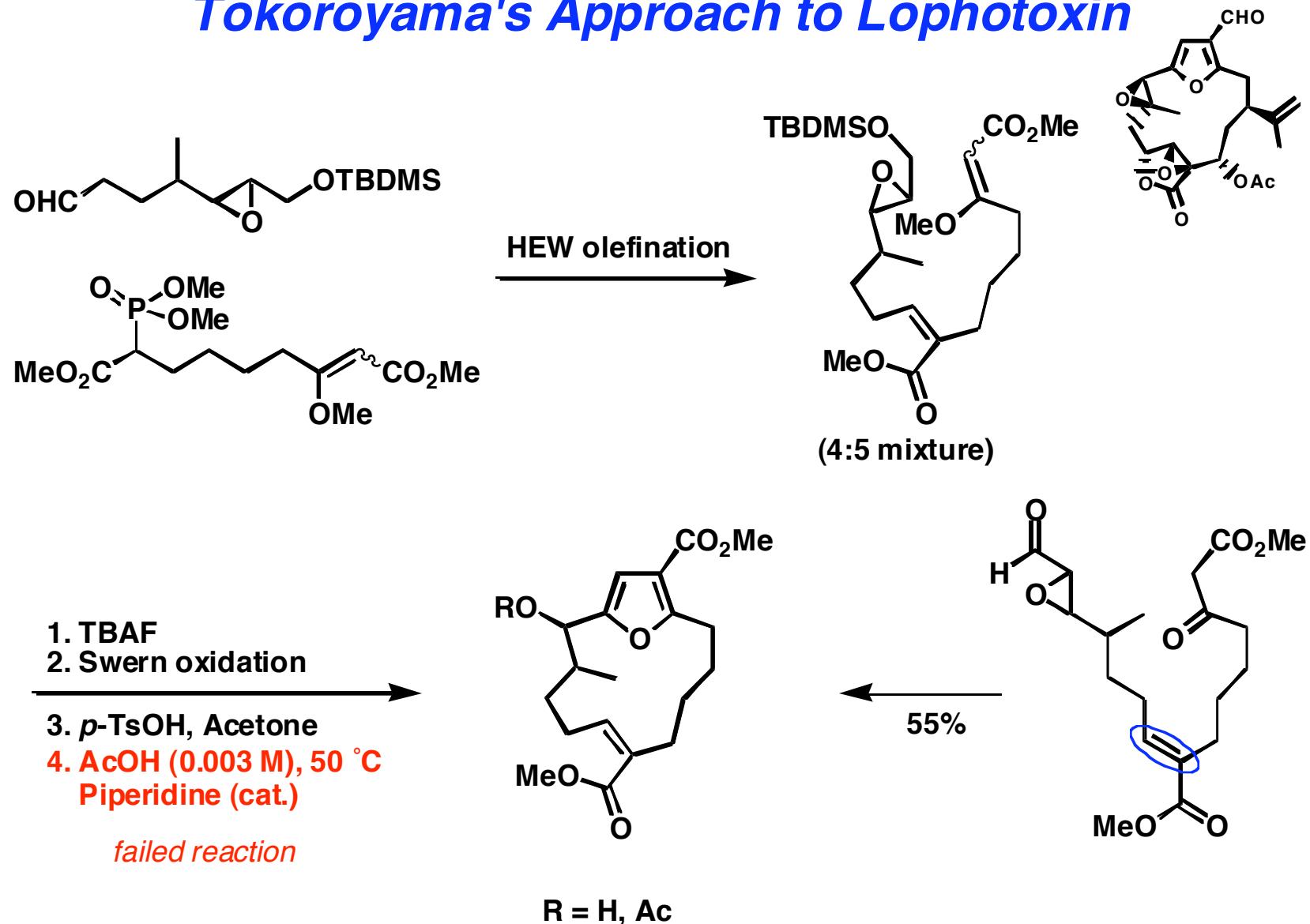
*Current work done in the group*

## Tius' Approach to Lophotoxin



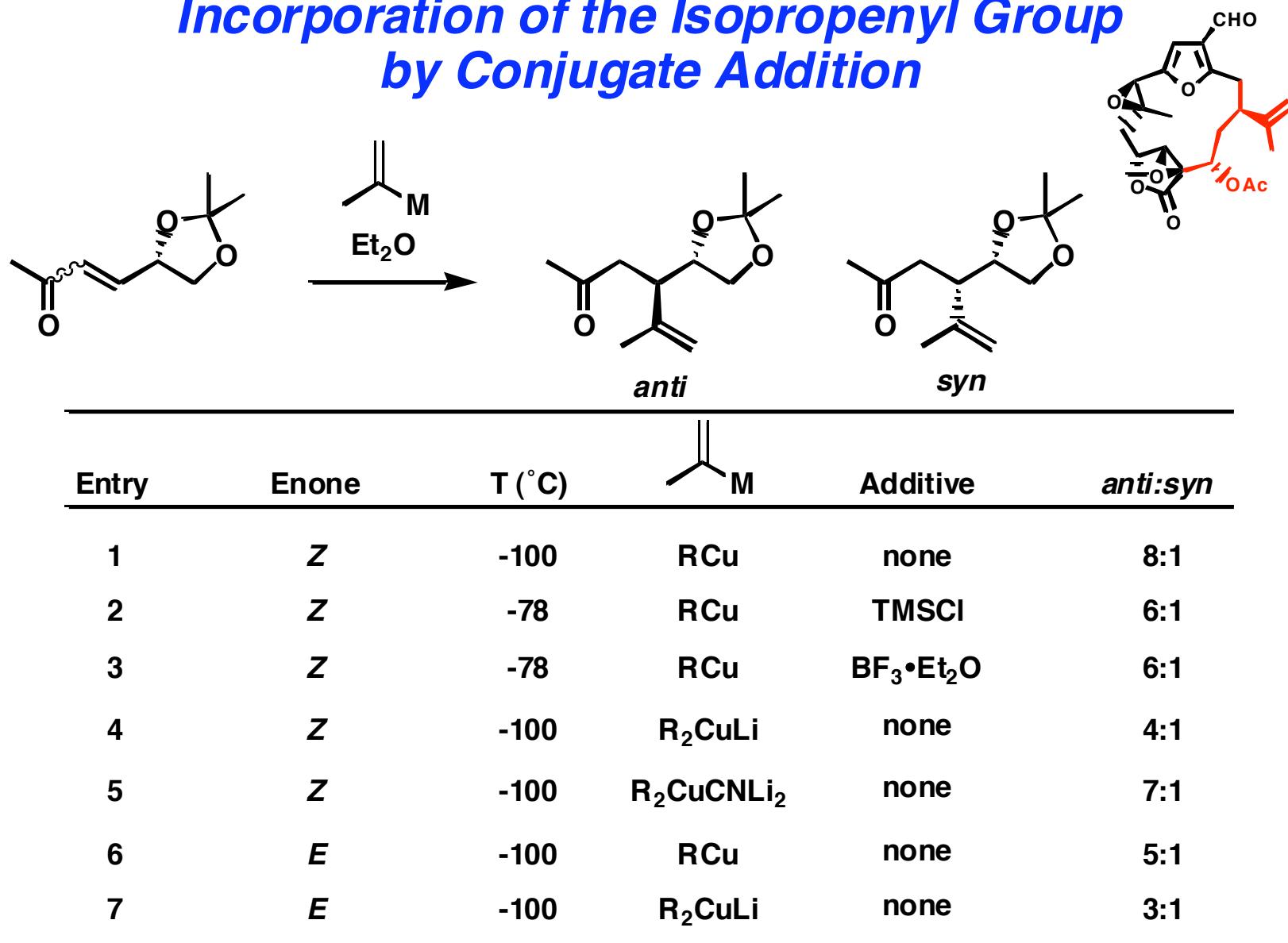
Tius, M. A.; Trehan, S. *J. Org. Chem.* **1986**, *51*, 767

## Tokoroyama's Approach to Lophotoxin



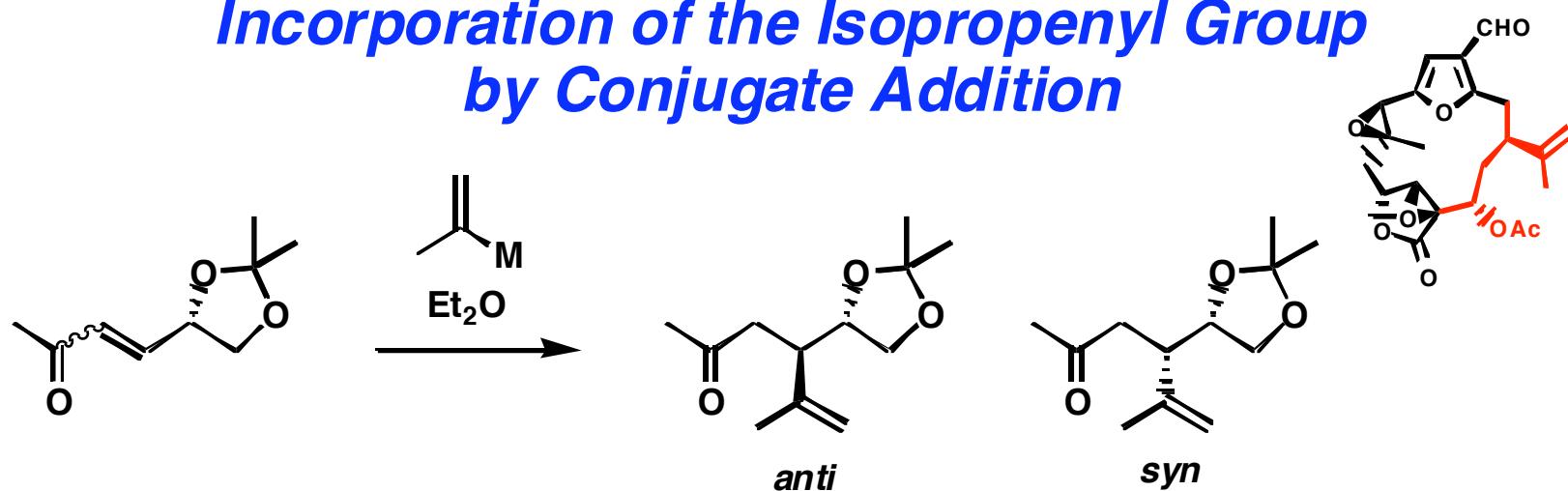
Kondo, A.; Ochi, T.; Ito, H.; Tokoroyama, T.; Siro, M. *Chem. Lett.* **1987**, 1491

## Incorporation of the Isopropenyl Group by Conjugate Addition



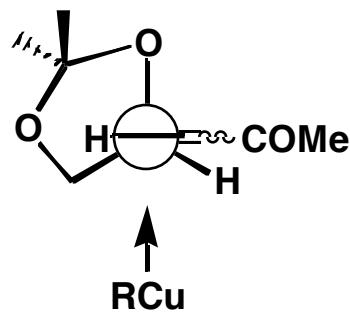
Leonard, J.; Ryan, G. *Tetrahedron Lett.* **1987**, 28, 2525

## Incorporation of the Isopropenyl Group by Conjugate Addition



Entry	Enone (E:Z)	T (°C)	$\text{CH}_2=\text{CH}-\text{M}$	Yield (%)	anti:syn
1	6:1	-78	RCu	79	7:1
2	7:2	-100	RLi	60	1:36

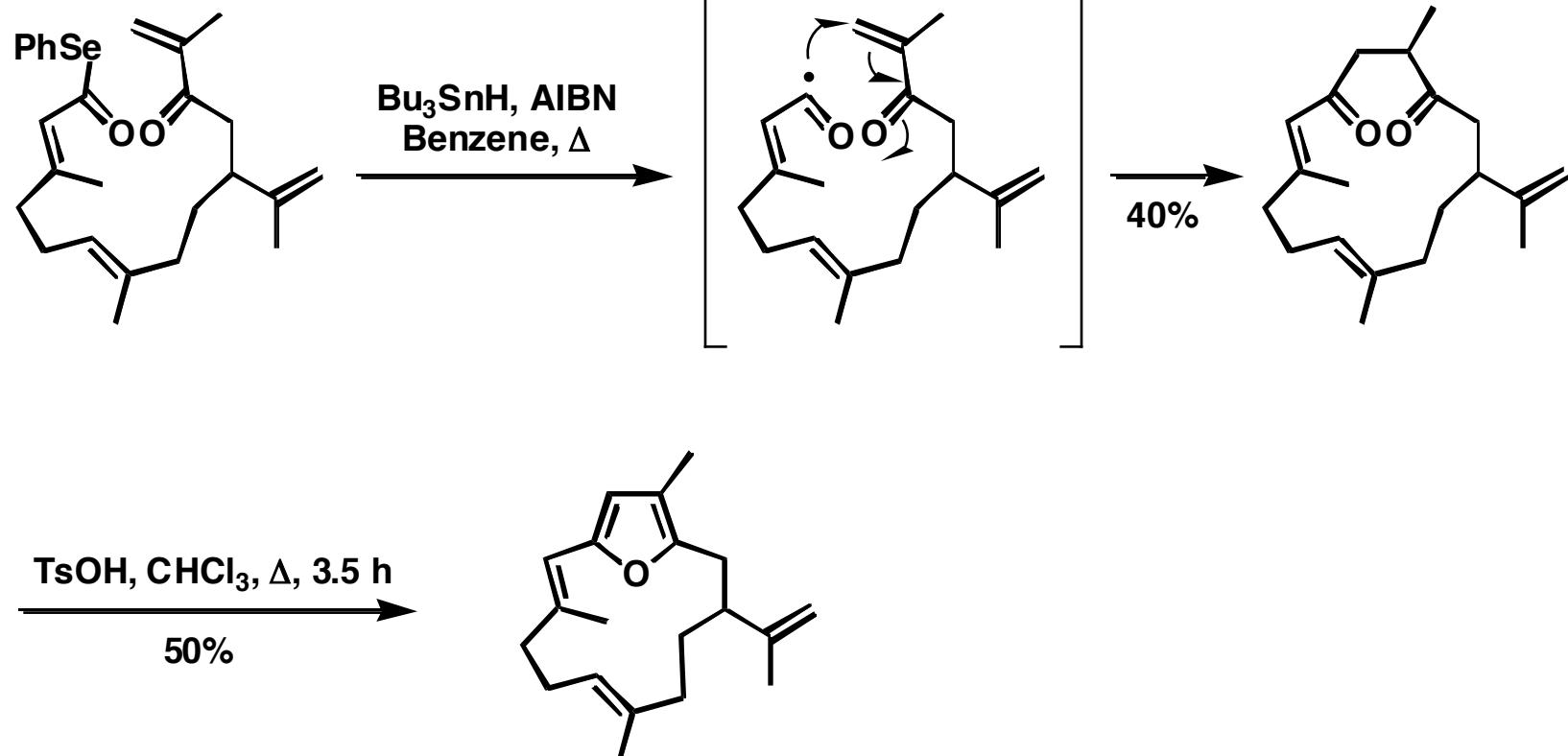
Felkin-Anh transition state



Leonard, J.; Ryan, G. *Tetrahedron Lett.* **1987**, *28*, 2525

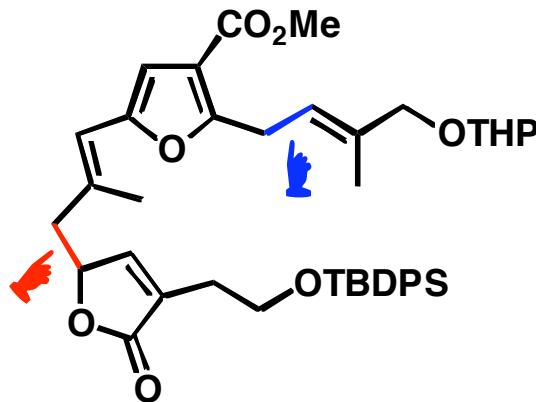
## Pattenden's First Approach to Lophotoxin

Acyl Radical Macrocyclization Strategy

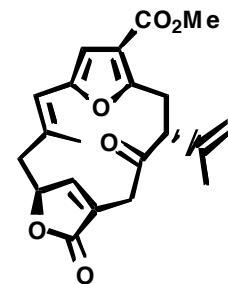


Astley, M. P.; Pattenden, G. *Synthesis* **1992**, 101

## Paquette's Approach to Acerosolide



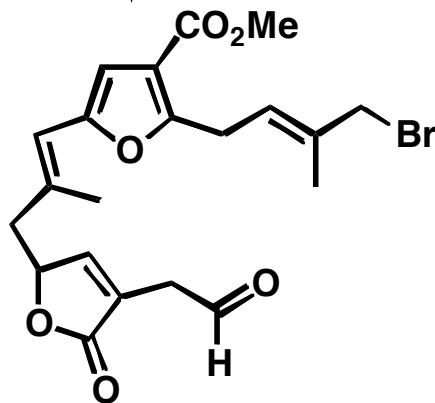
*Allylstannane addition to aldehyde ( $\text{SnCl}_4$ )  
 $sp^3-sp^2$  Stille coupling  
racemic synthesis*



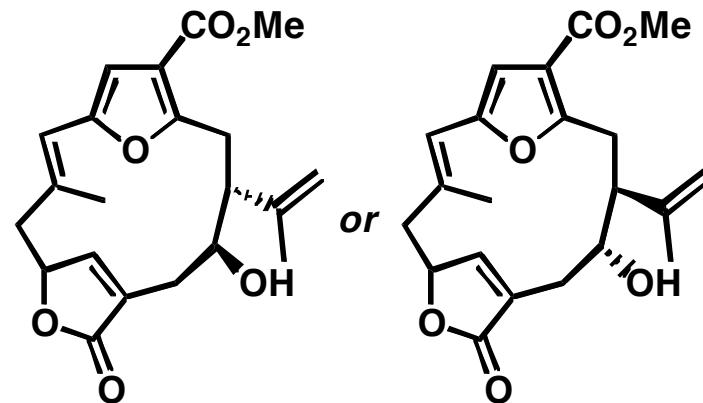
1.  $\text{Ph}_2\text{PCH}_2\text{CH}_2\text{PPh}_2 \bullet \text{Br}_2$   
 $\text{CH}_2\text{Cl}_2, 0^\circ\text{C}$  (64%)  
2. HF, ACN/H<sub>2</sub>O (68%)  
3. PDC, MS, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C (46%)

*Oxidation with PDC in DMF affords Acerosolide*

*Possibility of epimerization during the oxidation step*

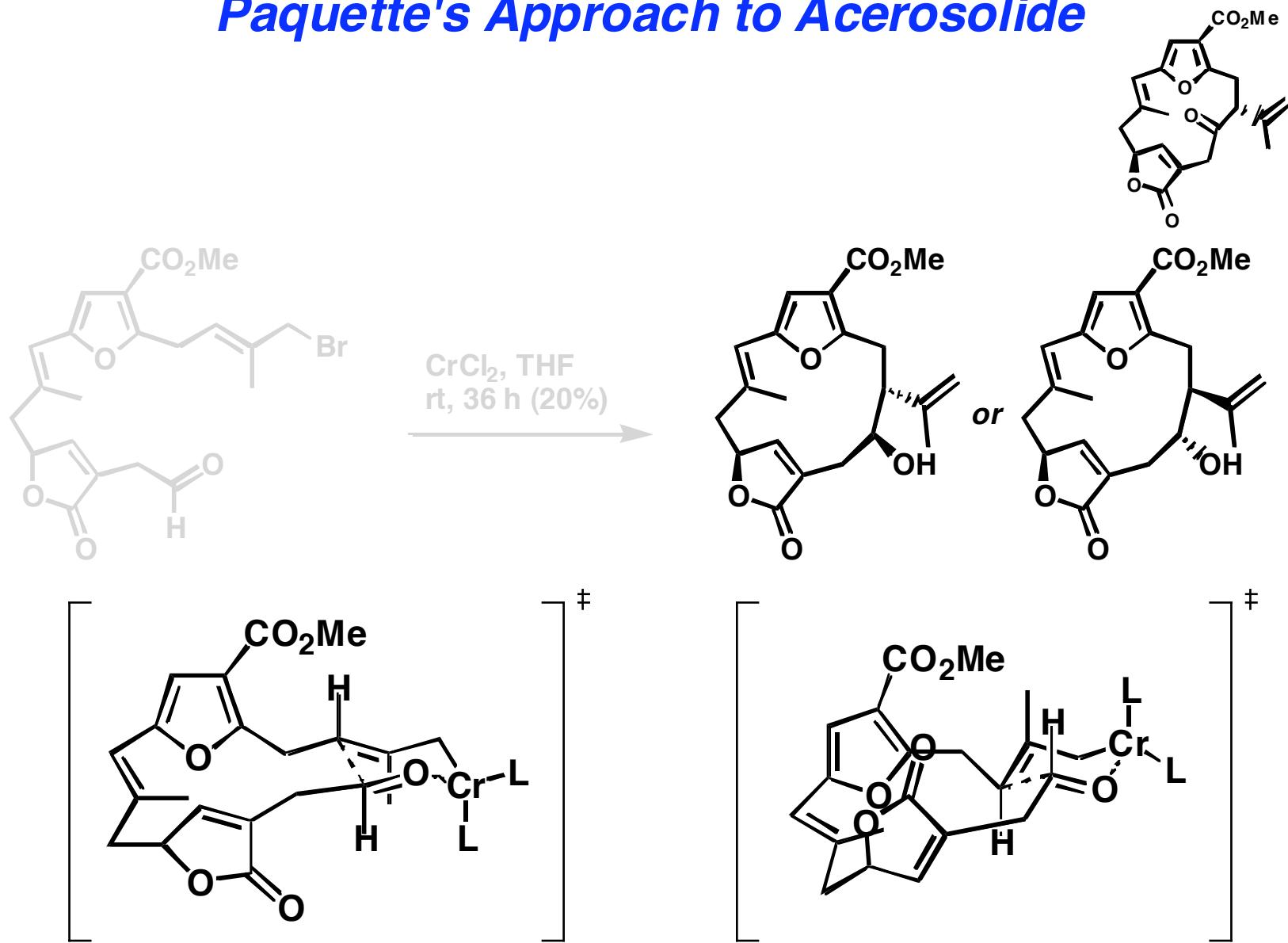


$\text{CrCl}_2, \text{THF}$   
rt, 36 h (20%)



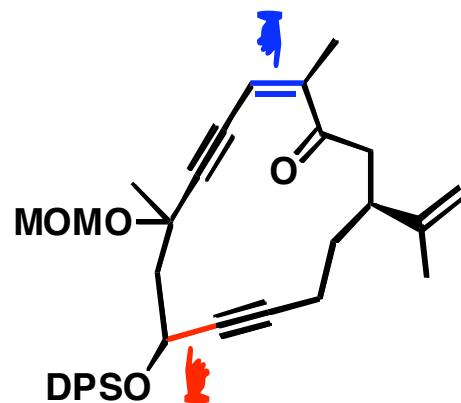
Paquette, L. A.; Astles, P. C. *J. Org. Chem.* **1993**, *58*, 165

## Paquette's Approach to Acerosolide

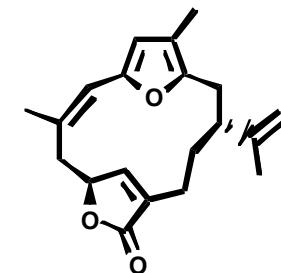


Paquette, L. A.; Astles, P. C. *J. Org. Chem.* **1993**, *58*, 165

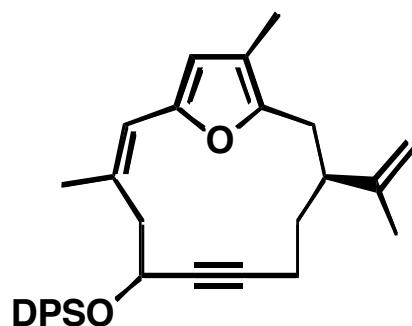
## Marshall's Approach to Rubifolide



Lithium acetylide addition  
HEW olefination

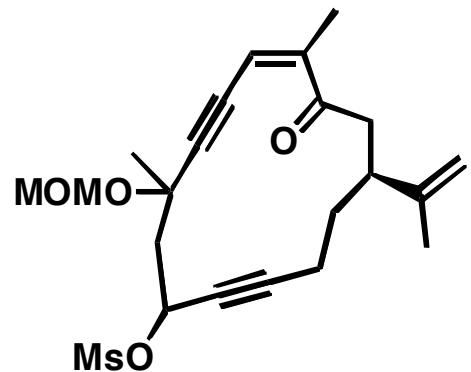


↓  
1. DIBALH  
2. KOt-Bu, *t*-BuOH  
THF, 18-C-6  
*failed reaction*

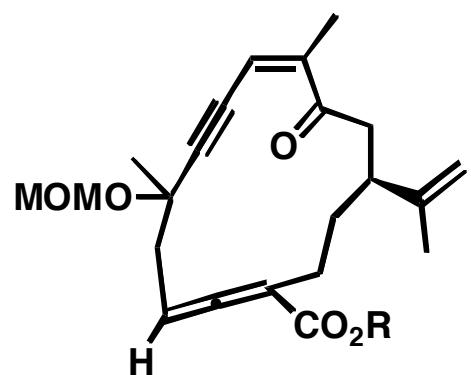


Marshall, J. A.; Sehon, C. A. *J. Org. Chem.* **1997**, *62*, 4313

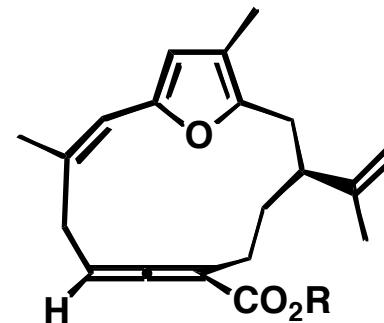
## Marshall's Approach to Rubifolide



Pd( $\text{PPh}_3$ )<sub>4</sub>, CO  
TMSCH<sub>2</sub>CH<sub>2</sub>OH, THF

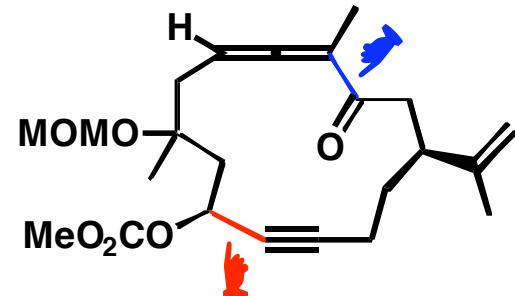


1. LiBH<sub>3</sub>N(*i*Pr)<sub>2</sub>  
2. AgNO<sub>3</sub>  
*failed reaction*

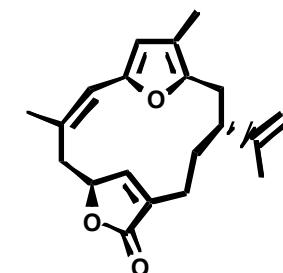


Marshall, J. A.; Sehon, C. A. *J. Org. Chem.* **1997**, *62*, 4313

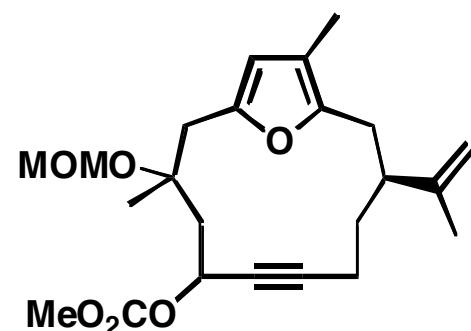
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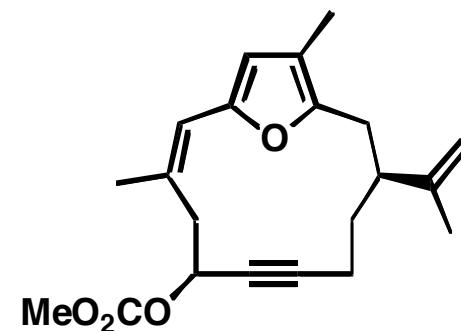
Lithium acetylide addition  
Allenylstannane addition/oxidation/isomerization



↓  
 $\text{AgNO}_3, \text{SiO}_2$   
Hexane (84%)



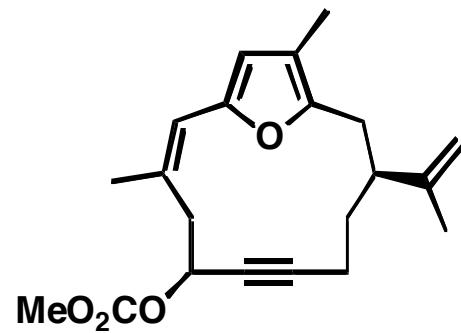
$p\text{-TsOH}, \text{CH}_2\text{Cl}_2$   
73%



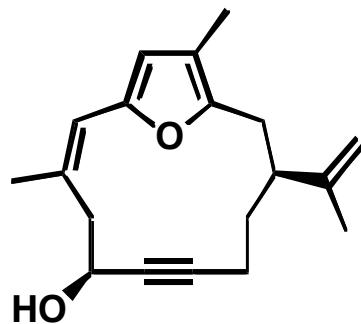
1:1 mixture

Marshall, J. A.; Sehon, C. A. *J. Org. Chem.* **1997**, *62*, 4313

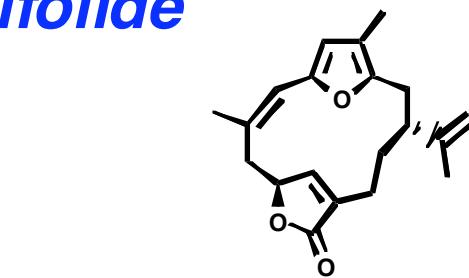
## Marshall's Approach to Rubifolide



- ↓
1.  $\text{K}_2\text{CO}_3$ ,  $\text{MeOH}$  (96%)
  2.  $\text{MnO}_2$ ,  $\text{CH}_2\text{Cl}_2$  (98%)
  3. K-Selectride,  $\text{THF}$ ,  $-78^\circ\text{C}$  (98%)

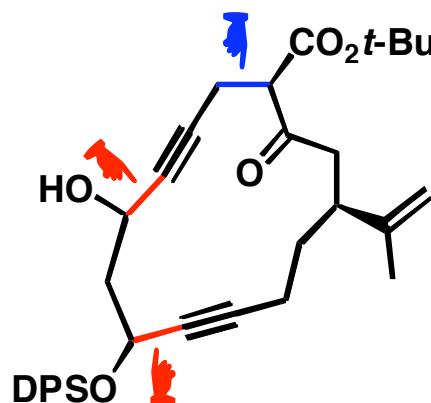


1. TFAA, 2,6-Lutidine  
 $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{H}_2\text{O}$ ,  $\text{THF}$ ,  $\text{CO}$  atm  
2.  $\text{AgNO}_3$ ,  $\text{SiO}_2$ , Hexane (49%)

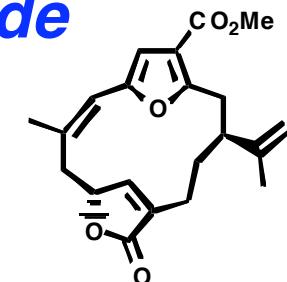


Marshall, J. A.; Sehon, C. A. *J. Org. Chem.* **1997**, *62*, 4313

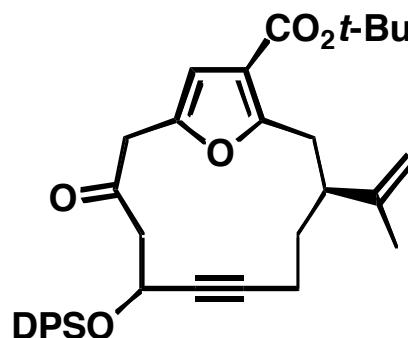
## Marshall's Approach to Deoxypukalide



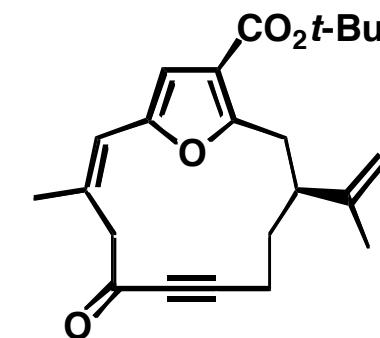
Lithium acetylide addition  
 $\beta$ -keto ester alkylation  
mixture of 8 diastereoisomers



↓  
1. DMP oxidation  
2. SiO<sub>2</sub>, Hexanes (96%)

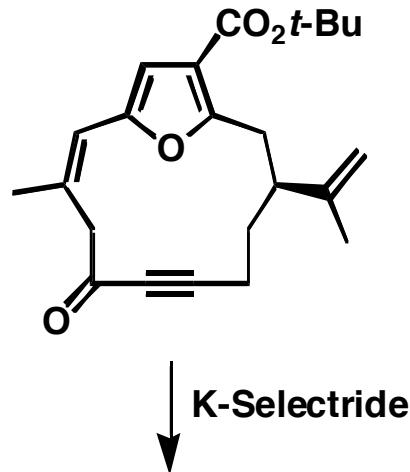


1. LiHMDS, Comins' Reagent (75%)  
2. Pd(PPh<sub>3</sub>)<sub>4</sub>, Me<sub>2</sub>Zn (91%)  
3. TBAF (85%)  
4. DMP oxidation

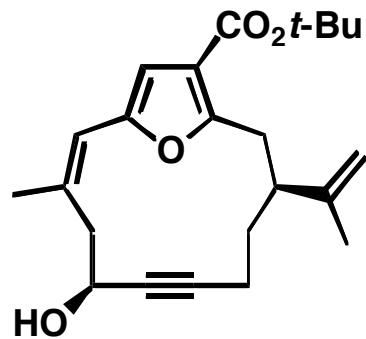


Marshall, J. A.; Van Devender, E. A. *J. Org. Chem.* 2001, 66, 8037

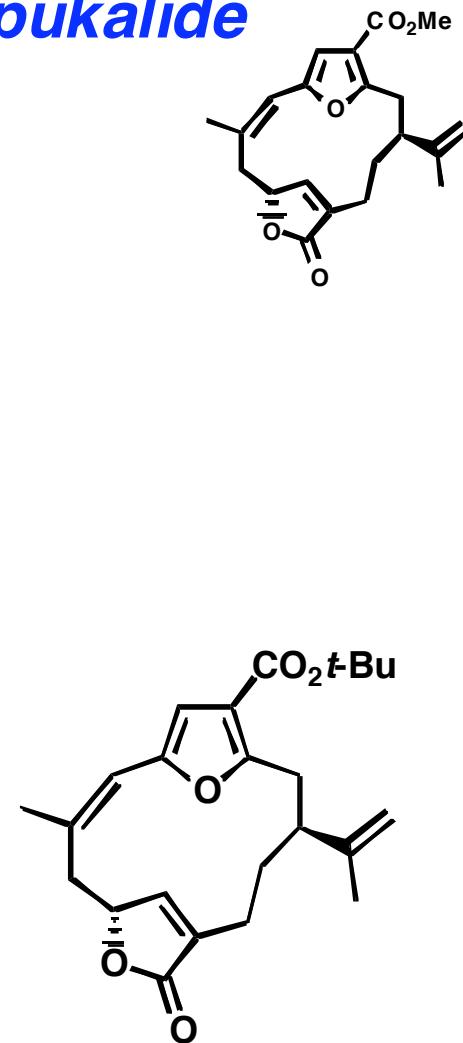
## Marshall's Approach to Deoxypukalide



K-Selectride



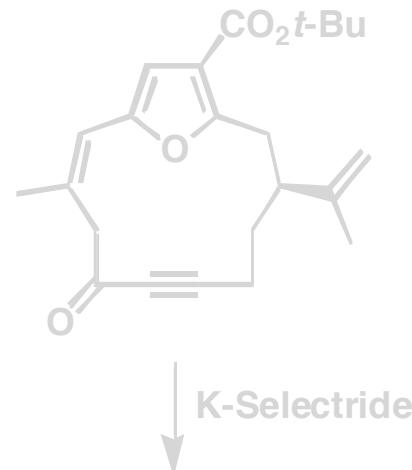
1. TFAA, 2,6-Lutidine  
 $\text{Pd}(\text{PPh}_3)$ ,  $\text{H}_2\text{O}$ , THF, CO atm  
2.  $\text{AgNO}_3$ ,  $\text{SiO}_2$ , Hexane (58%)



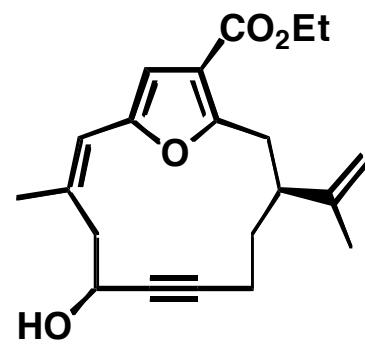
*Thermolysis (210 °C) and treatment with  $\text{TMSCHN}_2$  affords Deoxypukalide*

Marshall, J. A.; Van Devender, E. A. *J. Org. Chem.* **2001**, *66*, 8037

## Marshall's Approach to Deoxypukalide



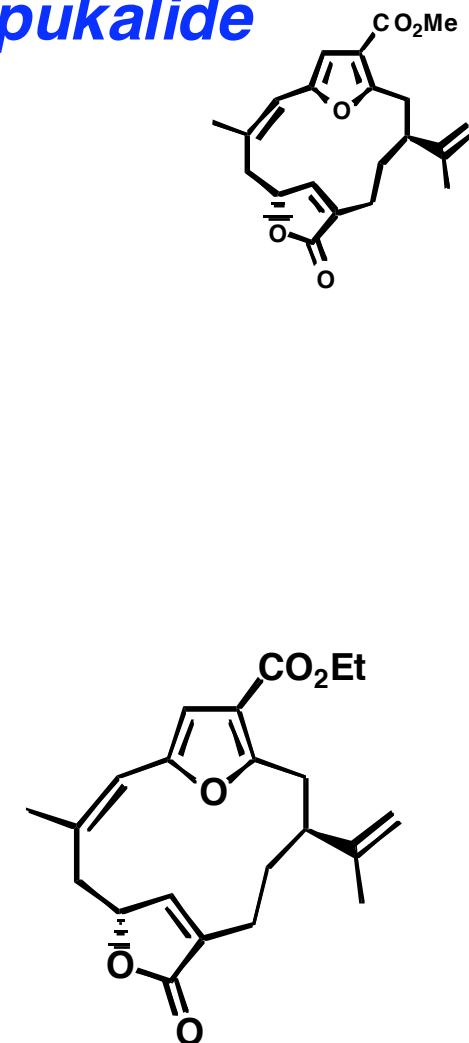
K-Selectride



1:1 mixture

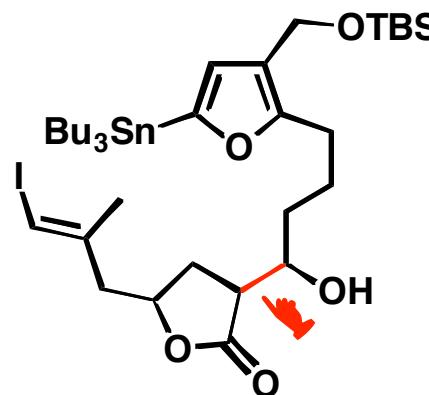
1. TFAA, 2,6-Lutidine  
 $\text{Pd}(\text{PPh}_3)$ ,  $\text{H}_2\text{O}$ , THF, CO atm  
2.  $\text{AgNO}_3$ ,  $\text{SiO}_2$ , Hexane (20-30%)

Stoichiometric amount of Pd

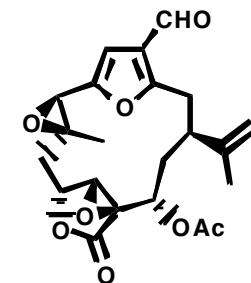


Marshall, J. A.; Van Devender, E. A. *J. Org. Chem.* **2001**, *66*, 8037

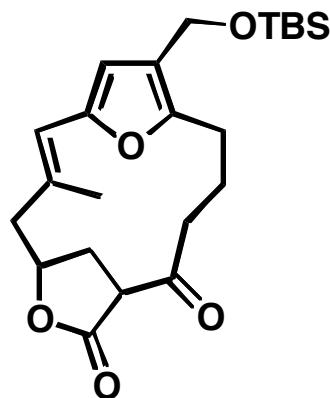
## *Paterson's Approach to Lophotoxin*



*Aldol reaction*



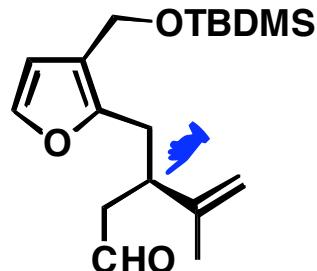
↓  
1.  $\text{Pd}_2(\text{dba})_3, \text{AsPh}_3$   
NMP, 40 °C, 24 h  
2. DMP,  $\text{CH}_2\text{Cl}_2$  (15%)



*1:1 mixture of diastereoisomers*

Paterson, I.; Brown, R. E.; Urch, C. J. *Tetrahedron Lett.* **1999**, *40*, 5807

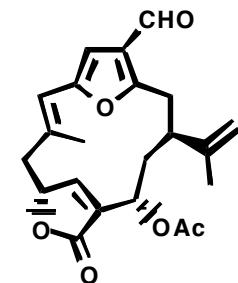
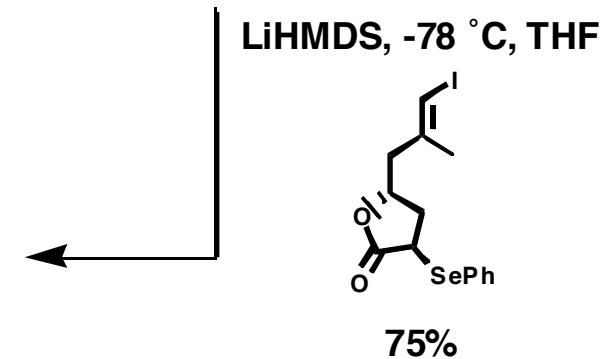
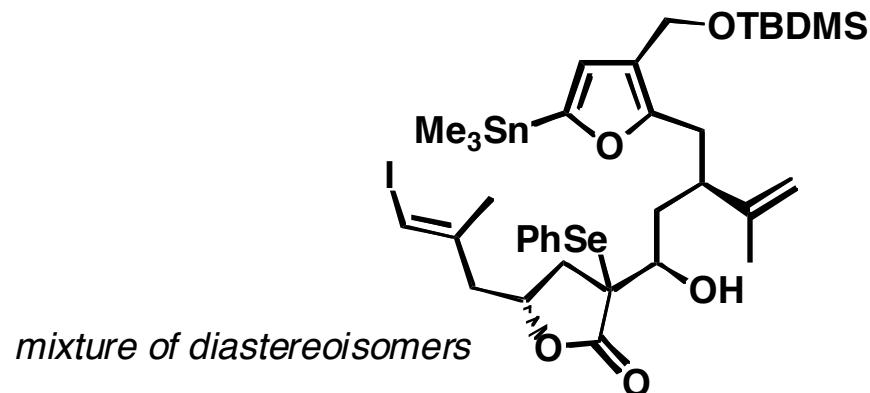
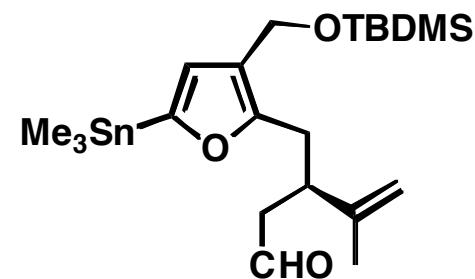
# Pattenden's Synthesis of bis-Deoxylophotoxin



*Deconjugative alkylation using  
Evans' oxazolidinone*

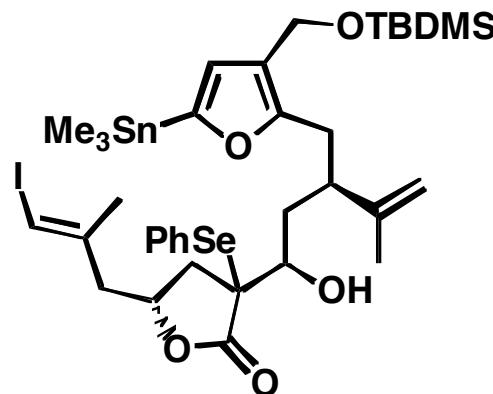
1.  $\text{NaBH}_4$ , MeOH, 0 °C, 70%

2. BuLi, TMEDA, rt, 6 h  
then  $\text{Me}_3\text{SnCl}$ , 0 °C to rt, 16 h, 80%  
3. TPAP, NMO, MS,  $\text{CH}_2\text{Cl}_2$ , 1 h, 75%

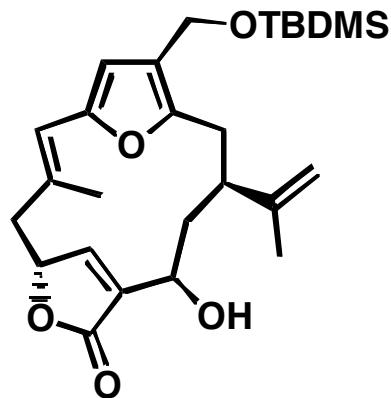
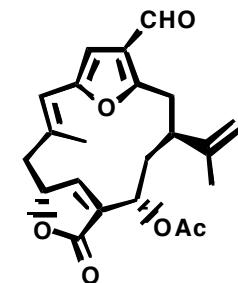


Cases, M.; de Turiso, F. G.-L.; Pattenden, G. *Synlett* 2001, 1869

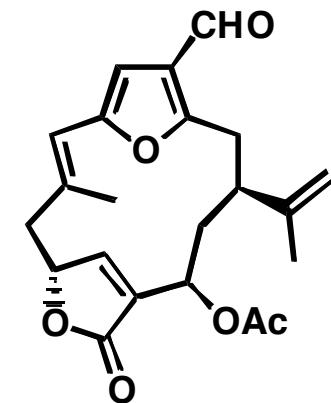
## Pattenden's Synthesis of bis-Deoxylophotoxin



1.  $\text{H}_2\text{O}_2$ ,  $\text{CH}_2\text{Cl}_2/\text{Pyridine}$ , 1 h, 0 °C  
2.  $\text{AsPh}_3$ ,  $\text{Pd}_2\text{dba}_3$ , NMP, 40 °C, 14 h  
20%



1.  $\text{Ac}_2\text{O}$ ,  $\text{Et}_3\text{N}$ , DMAP, rt, 4 h, 40%  
2. CSA,  $\text{MeOH}/\text{CH}_2\text{Cl}_2$ , 3 h, 0 °C  
3. Dess-Martin, Pyridine,  $\text{CH}_2\text{Cl}_2$ , 3 h, 0 °C  
80%

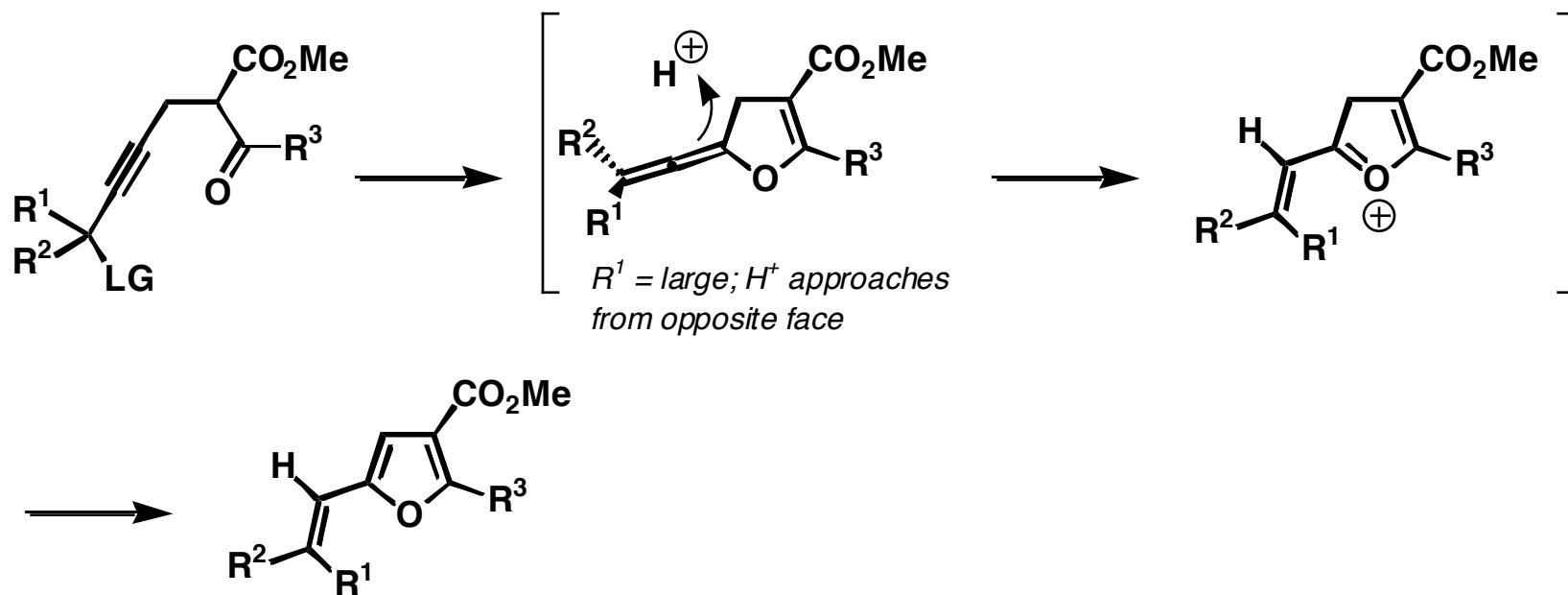
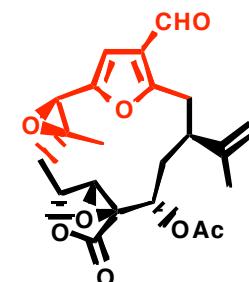


For a short review on the application of the intramolecular Stille reaction in some target natural product syntheses

Pattenden, G.; Sinclair, D. J. *J. Organomet. Chem.* 2002, 653, 261

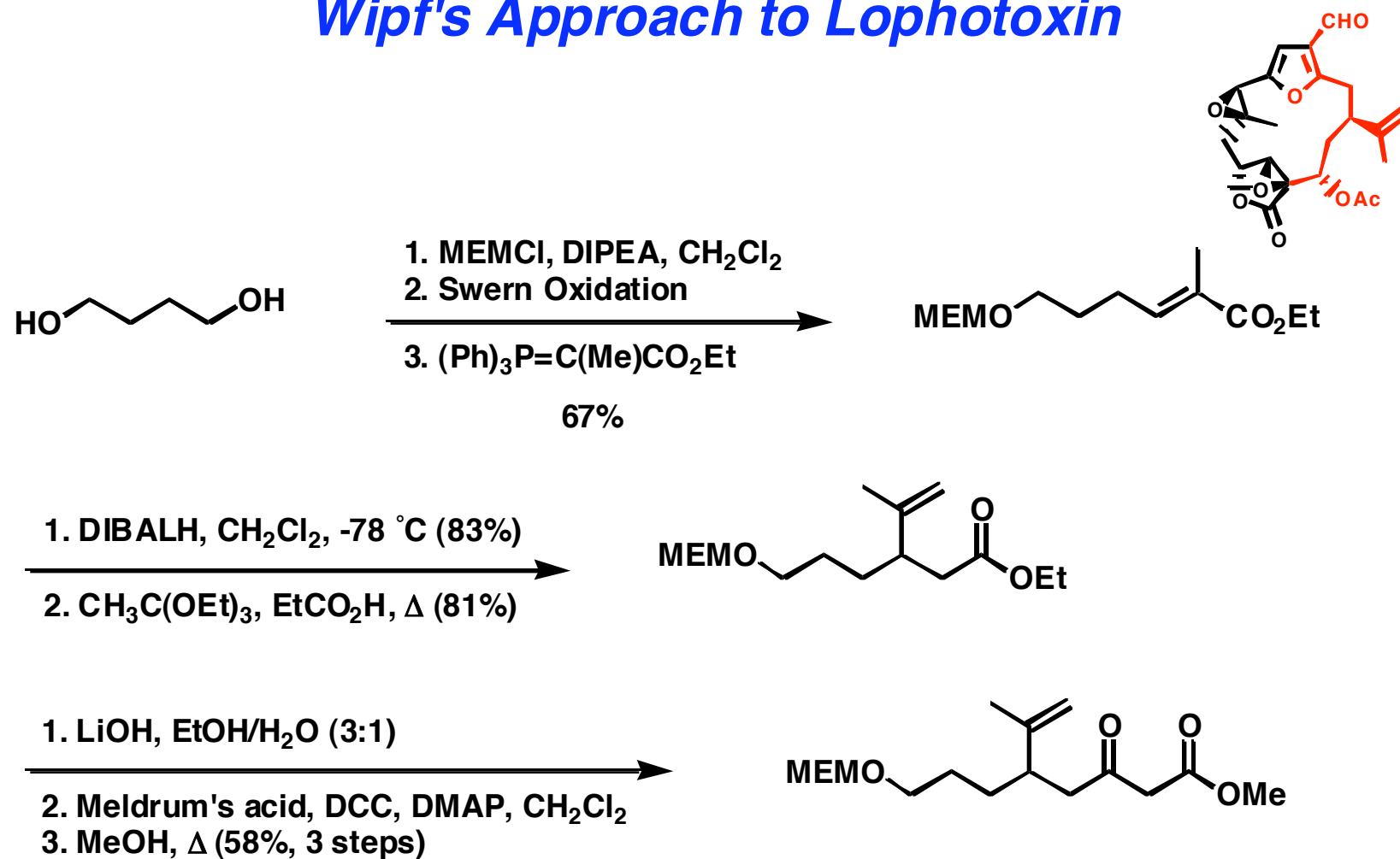
## Wipf's Approach to Lophotoxin

Key step: cyclization of an  $\alpha$ -propargyl  $\beta$ -keto ester



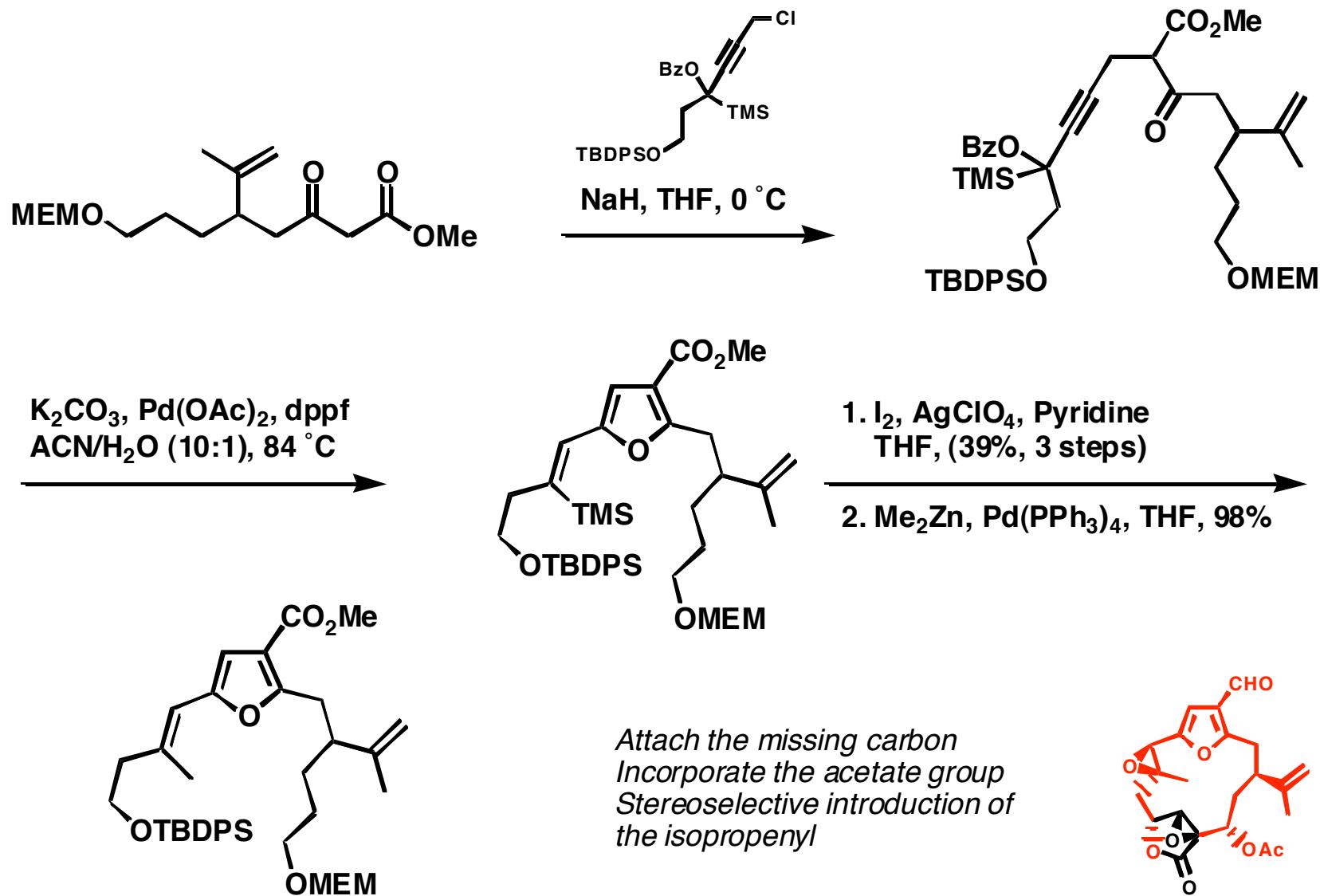
Wipf, P.; Rahman, L. T.; Rector, S. R. *J. Org. Chem.* **1998**, *63*, 7132

## Wipf's Approach to Lophotoxin



Wipf, P.; Soth, M. J. Org. Lett. 2002, 4, 1787

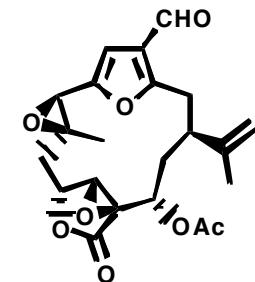
## Wipf's Approach to Lophotoxin



Wipf, P.; Soth, M. J. Org. Lett. 2002, 4, 1787

## Future Work

👉 Develop a way to introduce the isopropenyl group in a diastereoselective way



👉 Find an efficient way to convert the methyl ester to a terminal alkyne

