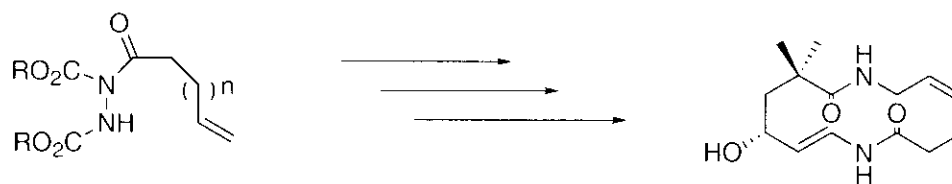


### Use of N-N Bond Stereodynamics in Ring-Closing Metathesis to Form Medium-Sized Rings and Macrocycles

Yi Jin Kim and Daesung Lee

*Org. Lett.* **2004** ASAP



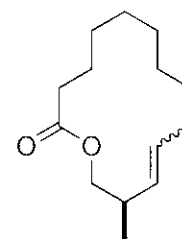
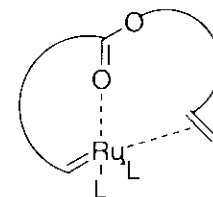
## Ring Closing Metathesis

Macrocyclic framework plays a significant role in natural products.

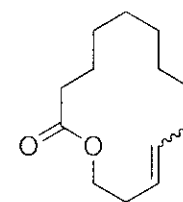
- RCM is one way to gain entry into such systems and has been widely used.

Factors that affect the efficiency of RCM:

- i) presence of functional groups that serve to assemble the reacting sites
- ii) low steric congestion near double bonds
- iii) appropriate distance b/n polar groups and alkenes to be metathesized



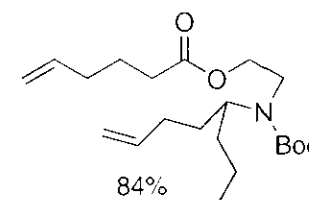
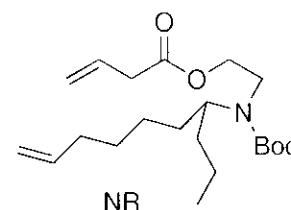
10%



52%

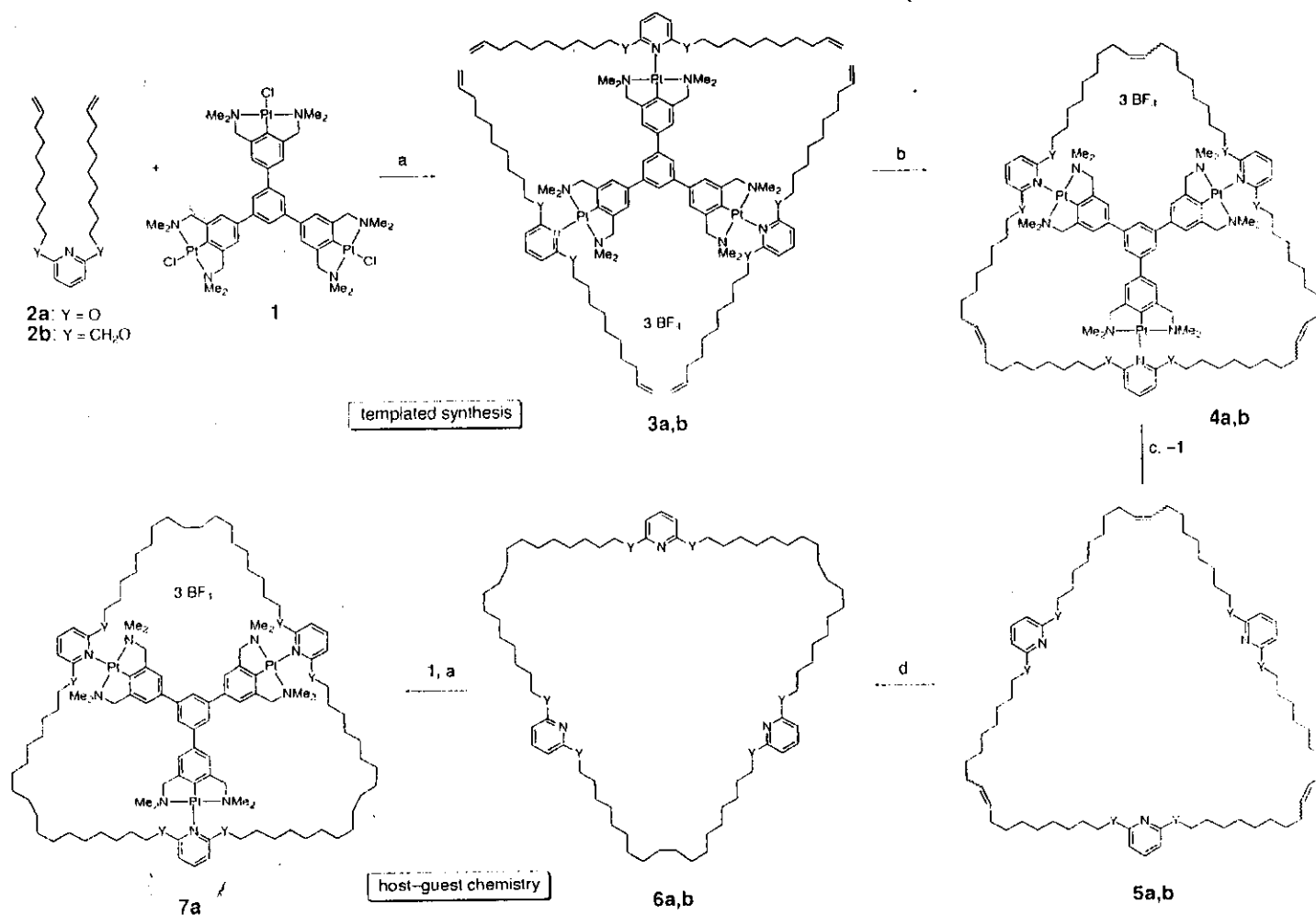
Recently,

- i) template effect
- ii) conformational and stereoelectronic constraints



84%

## Shape-Persistent Tricationic Platinum Template

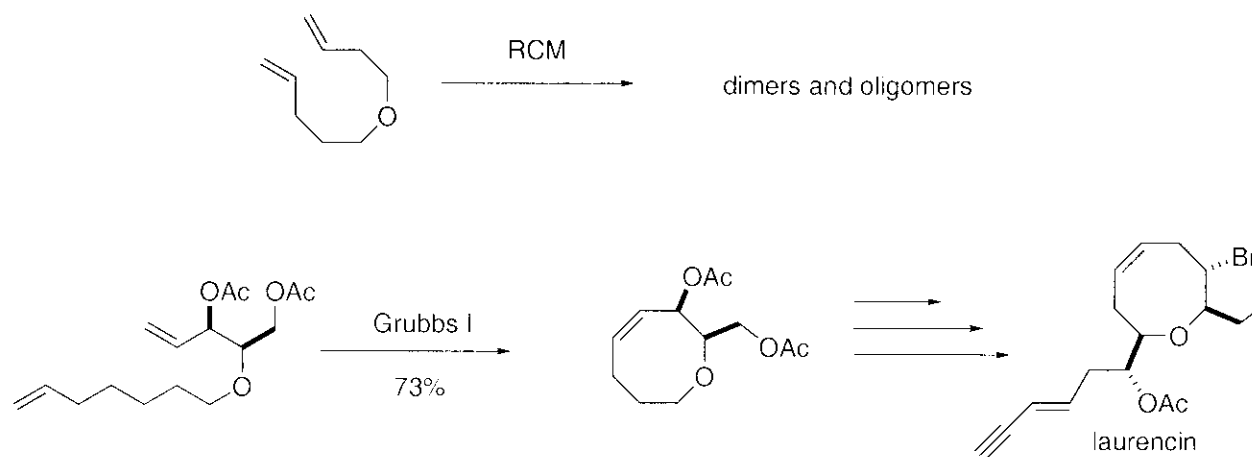


-69 membered macrocycle

-this template synthesis is applied to large heteroatomic crown ether rings:

Scheme 1. a) AgBF<sub>4</sub>, CH<sub>2</sub>Cl<sub>2</sub>; b) [Cl<sub>2</sub>(PCy<sub>3</sub>)<sub>2</sub>Ru=CHPh] 5 mol%, CH<sub>2</sub>Cl<sub>2</sub>; c) NaCl, H<sub>2</sub>O/CH<sub>2</sub>Cl<sub>2</sub>; d) H<sub>2</sub>, Pd/C.

## Stereoelectronic Constraints

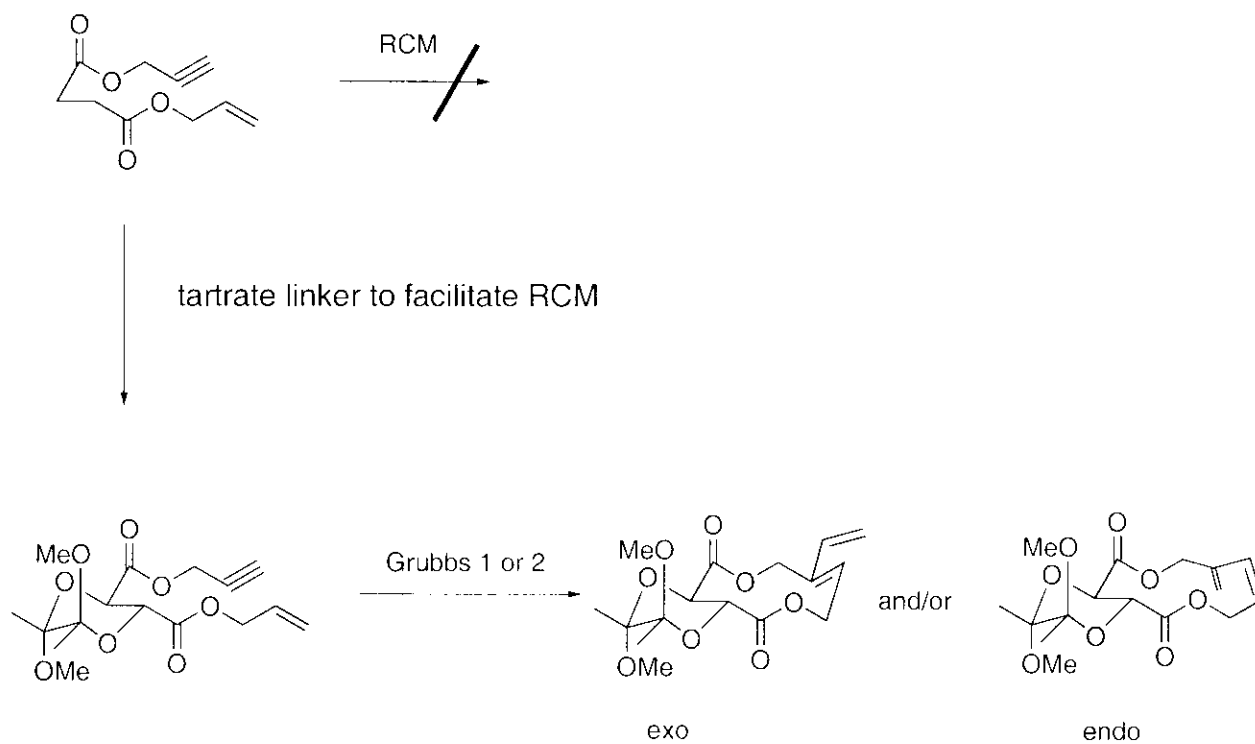


- vicinal stereogenic centers provide access to conformations where the olefinic chains are gauche which facilitate ring closing -----> gauche effect of 1,2 dioxigen substituents

Hoveyda, A. H. *J. Am. Chem. Soc.* **1996**, 118, 4291.

Choy, A.; Crimmins, M. *J. Am. Chem. Soc.* **1999**, 121, 5653.

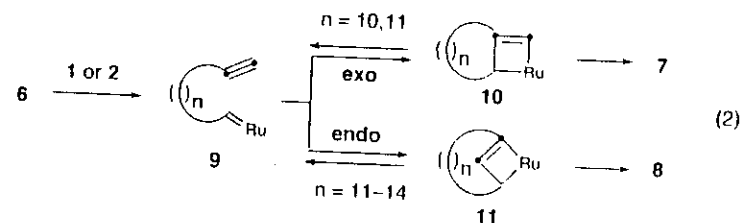
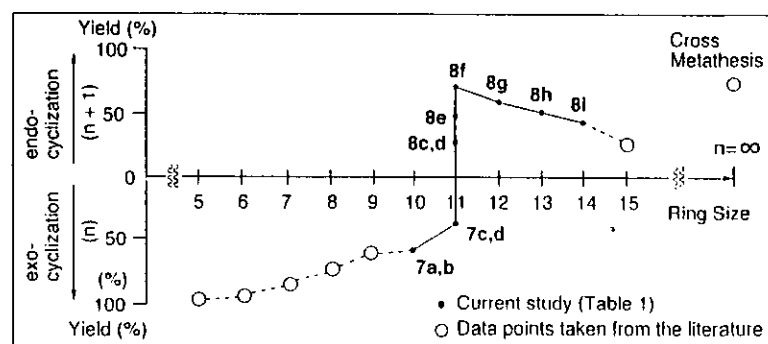
## Tartrate based Enyne Metathesis



## Tartrate based Enyne Metathesis:

Table 1. RCM of Enynes to Form Macrocycles<sup>a</sup>

Entry	Enyne substrate	Macrocycle product		E:Z	Yield (%) <sup>b</sup>
		exo	endo		
1	$m = n = 1$ $X = Y = O$ $R = H$			0:1	52
2	$R = Me$			0:1	61
3	$m = 1, n = 2$ $X = Y = O$ $R = H$			1:1 (2:1 for 8c) (E-8c only)	50 (91) <sup>d</sup> (90) <sup>e</sup>
4	$R = Me$			1:1 for 8d	92
5	$m = 1, n = 2$ $X = NH, Y = O$ $R = H$			1:0	35 (60) <sup>e</sup>
6	$m = 2, n = 1$ $X = Y = O$ $R = H$			1:1	70
7	$m = 2, n = 2$ $X = Y = O$ $R = H$			0:1	61
8	$m = 3, n = 2$ $X = Y = O$ $R = H$			2:1	54
9	$m = 3, n = 3$ $X = Y = O$ $R = H$			1:1	55



Hansen, E.; Lee, D. *J. Am. Chem. Soc.* **2003**, *125*, 9582-9583.

<sup>a</sup> Reactions performed with 5 mol % of 2 at 0.2 M in refluxing  $\text{CH}_2\text{Cl}_2$ .  
<sup>b</sup> Isolated yield. <sup>c</sup> The stereochemistry of 7a-d was determined by nOe, and that of 8e-i was determined by coupling constant. <sup>d</sup> RCM under ethylene in refluxing  $\text{CH}_2\text{Cl}_2$ . <sup>e</sup> RCM under ethylene at 25 °C.

- Unlike the 6-membered cyclic hydrazines, only limited synthetic methods are available for 7-10 membered rings.
- 1st. RCM utilizing a 1,2-diaza skeleton

Table 1. Ring-closing metathesis of **5** and deprotection of the Boc in **8**

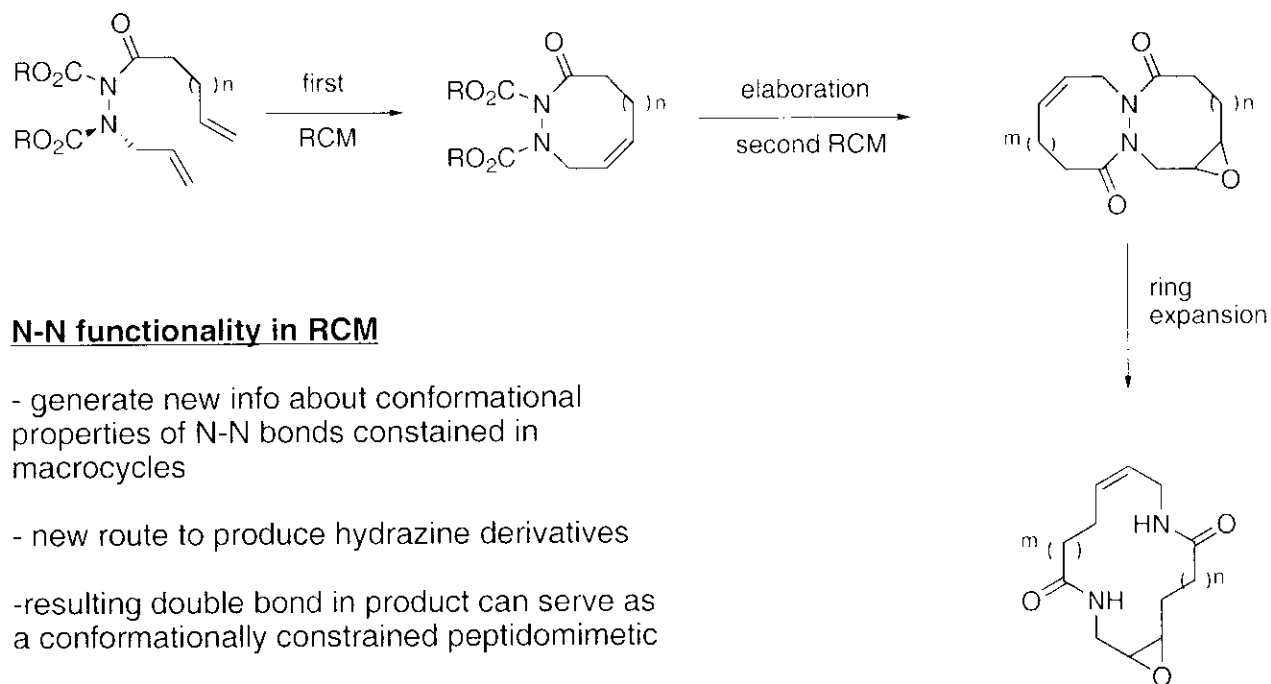
Entry	5	Conditions <sup>a</sup>	Yields (%) <sup>b</sup>	
			8	9
1	<b>5a</b> <i>m</i> = 1 <i>n</i> = 1	8 h 0.02 M	<b>8a</b> (93)	<b>9a</b> (96)
2	<b>5b</b> <i>m</i> = 1 <i>n</i> = 2	8 h 0.02 M	<b>8b</b> (98)	<b>9b</b> (80)
3	<b>5c</b> <i>m</i> = 2 <i>n</i> = 2	16 h 0.008 M	<b>8c</b> (74)	<b>9c</b> (97)
4	<b>5d</b> <i>m</i> = 3 <i>n</i> = 2	16 h 0.006 M	<b>8d</b> (72)	<b>9d</b> (95)
5	<b>5e</b> <i>m</i> = 3 <i>n</i> = 3	16 h 0.005 M	<b>8e</b> (70)	<b>9e</b> (84)

<sup>a</sup> Reaction time and concentration for the RCM.<sup>b</sup> Isolated yields.<sup>c</sup> Conditions for the deprotection of the Boc in **8e**: BF<sub>3</sub>·OEt<sub>2</sub>, 4 Å molecular sieves, CH<sub>2</sub>Cl<sub>2</sub>.Table 2. Ring-closing enyne metathesis of **6**

Entry	Substrate	Conditions <sup>a</sup>	Product	Yield (%) <sup>b</sup>
1	<b>6a</b> <i>m</i> = 1	4 h 0.02 M	<b>10a</b>	99
2	<b>6b</b> <i>m</i> = 2	8 h 0.02 M	<b>10b</b>	70
3	<b>6c</b> <i>m</i> = 3	10 h 0.02 M	<b>10c</b>	70

<sup>a</sup> Reaction time and concentration.<sup>b</sup> Isolated yields.Hahn, D-W.; Tae, J. *Tetrahedron Lett.* 2004, 45, 3757.

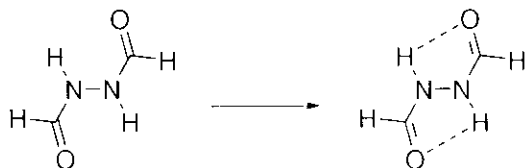
## Double RCM





## Stereodynamic Behavior of N-N

unsubstituted

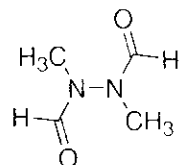


-  $E_a = 1$  kcal/mol

- Planar structure is more stable due to the hydrogen bonding.

- Due to the hydrogen bonding, electron density is withdrawn from the N-lone pair, which reduces repulsive interactions between the nitrogens.

substituted



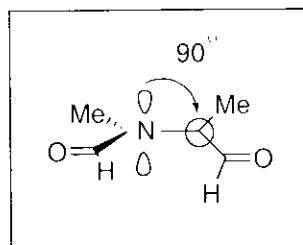
-  $E_a = 19$  kcal/mol

- No Hydrogen bonding ----> Planar structure is much less stable.

- Much more repulsive interactions between the nitrogens.

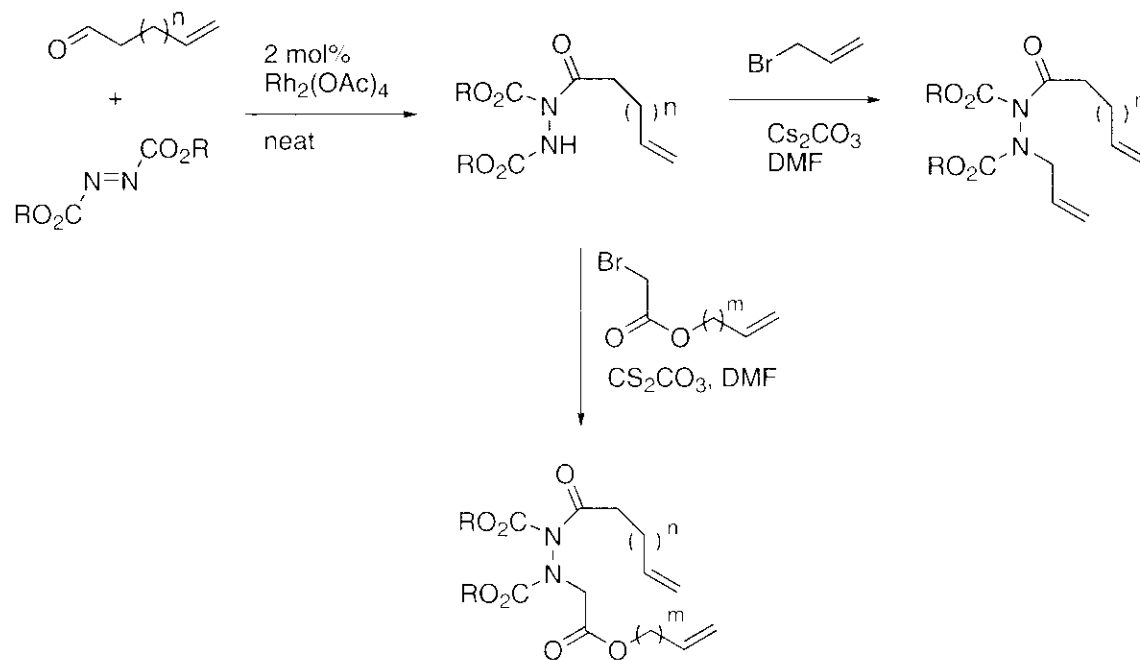
- Maintains a CO-N-N-CO near 90 °

- vicinal N, N' substituents will arrange in a gauche like arrangement ----> facilitate RCM



$E_a = \sim 19$  kcal/mol

### Synthesis of RCM Substrates



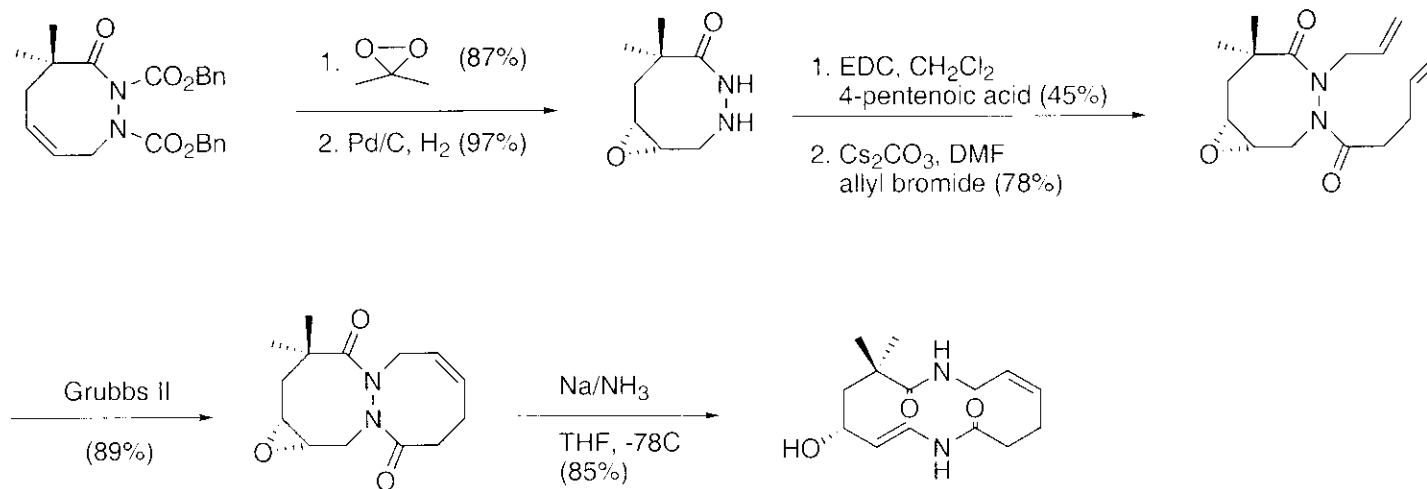
# Formation of Cyclic Hydrazines via RCM

Table 1. Formation of Cyclic Hydrazine Derivatives via RCM<sup>a</sup>

entry	RCM substrate (2)	product (3)	yield (%) <sup>b</sup>
1			93
2			88 (65) <sup>c</sup>
3			42
4			46
5			65
6			72 2:1 Z:E
7			94 3:2 Z:E
8			47 <sup>d,e</sup>
9			60 <sup>d,f</sup>

<sup>a</sup> Reactions are carried out in CH<sub>2</sub>Cl<sub>2</sub> (0.002 M) with 5–10 mol % of **8** for 2–5 h. <sup>b</sup> Isolated yield. <sup>c</sup> Yield for **2b**. <sup>d</sup> Reaction under ethylene. <sup>e</sup> 1:1 for **3h**:**3h'**. <sup>f</sup> 2:1:1 for **3i**:**3i'**.

### Synthesis of Macrocycle



### **Future Work:**

- Elaboration of final macrocycles
- study of biological performance as cyclic peptide mimics
- nanotube-forming propensity

### **Summary:**

- RCM utilizes the conformational constraints of N-substituted diacyhydrazines.
- RCM of N-substituted diacyhydrazines result in 8-14 membered cyclic hydrazines in good yield.
- New strategy to gain rapid access to macrocyclic amides from hydrazines