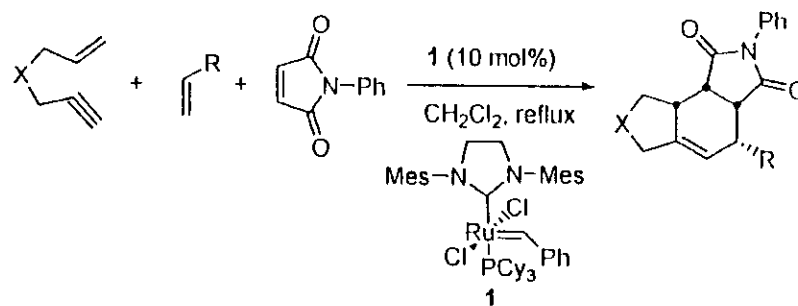


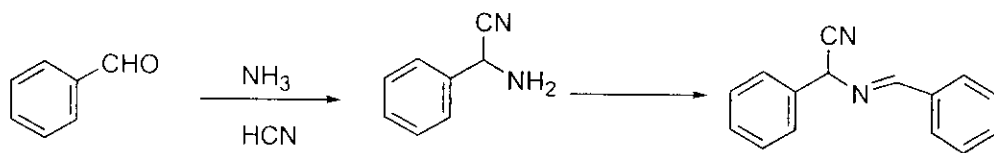
## One-Pot Three-Component Tandem Metathesis/Diels-Alder Reaction

Lee, H.; Kim, H.; Tae, H.; Kim, B.; Lee, J. *Org. Lett.* **2003**, *5*, 3439.



## Multi-Component Reactions (MCR)

- Chemical reactions that use three or more different starting materials and yields the final product in a one-pot procedure.
- Regarded as combinations or unions where the product of a two-component reaction reacts with a third component to give the next product and so on.....
- first MCR- Laurent and Gerhardt - 1838



- Today, about 300-400 different MCR's are known, 20-25% are based on isocyanides as one of the starting materials.

### Advantages:

- Highly efficient, generate more than 2 chemical bonds per operation
- Structure can be easily varied by diversifying starting material
- Starting materials are commercially available or easily prepared
- quick access to heterocyclic compounds
- atom economy

- Careful consideration of reactivities of starting materials and respective products must be examined
- Optimal conditions for reactions represent a significant challenge since the involved two-component reactions usually require different conditions.

## Multi-Component Reactions

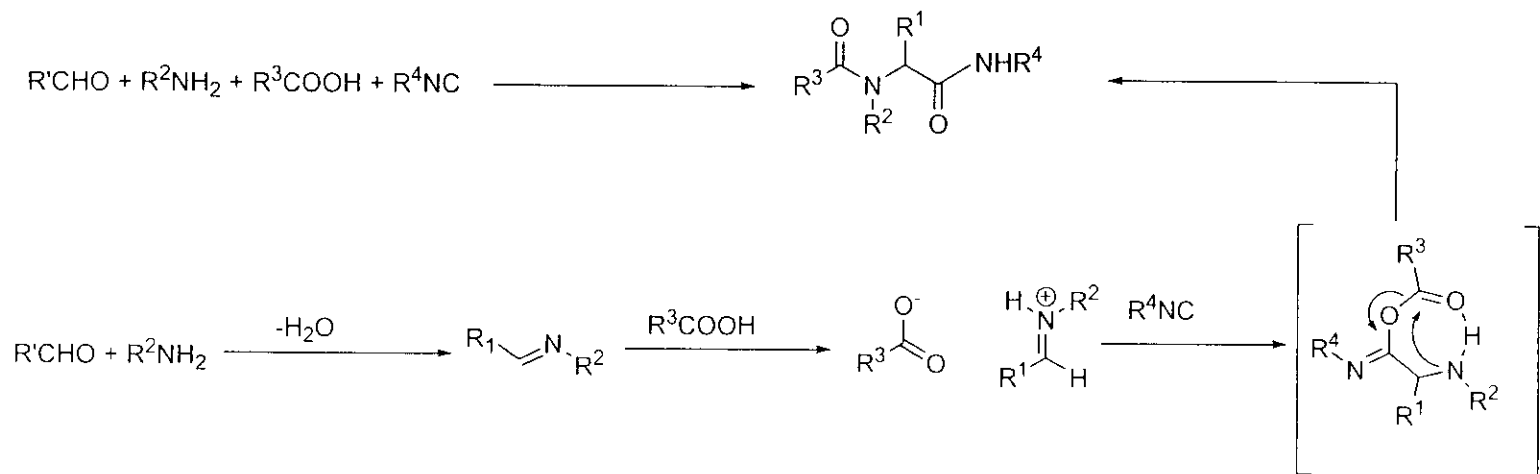
Table 1. Many of the historically significant MCRs are based on the reactivity of carbonyl or imine groups.

Name of the reaction	Year of discovery	Example <sup>[a]</sup>
Strecker synthesis <sup>[24]</sup>	(1838) 1850	<chem>O=Cc1ccc2ccccc12.C#N&gt;&gt;NC(C#N)c1ccc2ccccc12</chem>
Hantzsch dihydropyridine synthesis <sup>[25]</sup>	1882	<chem>CCOC(=O)C=C + NH3 + O=Cc1ccc(C(F)(F)F)cc1&gt;&gt;CCOC(=O)C1=CC(=C(C(F)(F)F)N1)C(=O)OCC</chem>
Radziszewski imidazole synthesis <sup>[26]</sup>	1882	<chem>CC(=O)C + CH2O + CN + NH3&gt;&gt;CC1=C(C)N=CN1</chem>
Hantzsch pyrrole synthesis <sup>[27]</sup>	1890	<chem>O=C(C)C(=O)OCC + PhNH2 + CCOC(=O)C(Br)C(=O)C&gt;&gt;CCOC(=O)C1=C(C)N(Ph)C1C(=O)OCC</chem>
Biginelli reaction <sup>[28, 29]</sup>	1891	<chem>NC(=O)N + CCOC(=O)C=C + O=Cc1ccc2ccccc12&gt;&gt;O=C1NC(=O)C2=CC=CC=C2C1C(=O)OCC</chem>
Mannich reaction <sup>[29]</sup>	1912	<chem>CC(=O)C + 2 CH2O + CN&gt;&gt;CC(=O)C(C)CN(C)C(C)C(=O)C</chem>
Bucherer-Bergs hydantoin synthesis <sup>[29, 30]</sup>	1941	<chem>C1=NC(=O)NC(=O)C1 + NH3 + CO2 + HCN&gt;&gt;NC(=O)C(C)C(=O)N</chem>

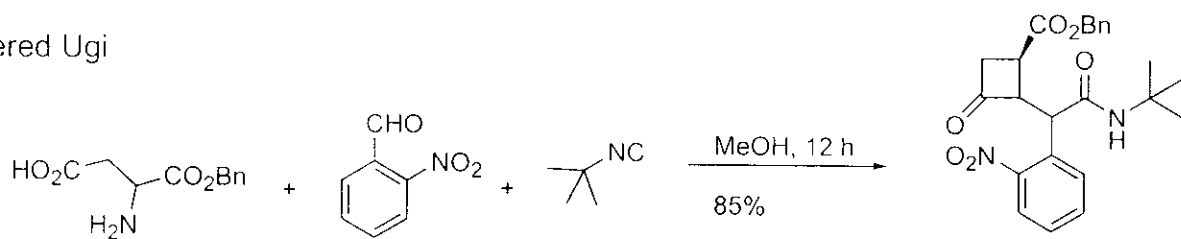
[a] T = thymine.

## Ugi Reaction

-discovered in 1959, most utilized MCR



-tethered Ugi

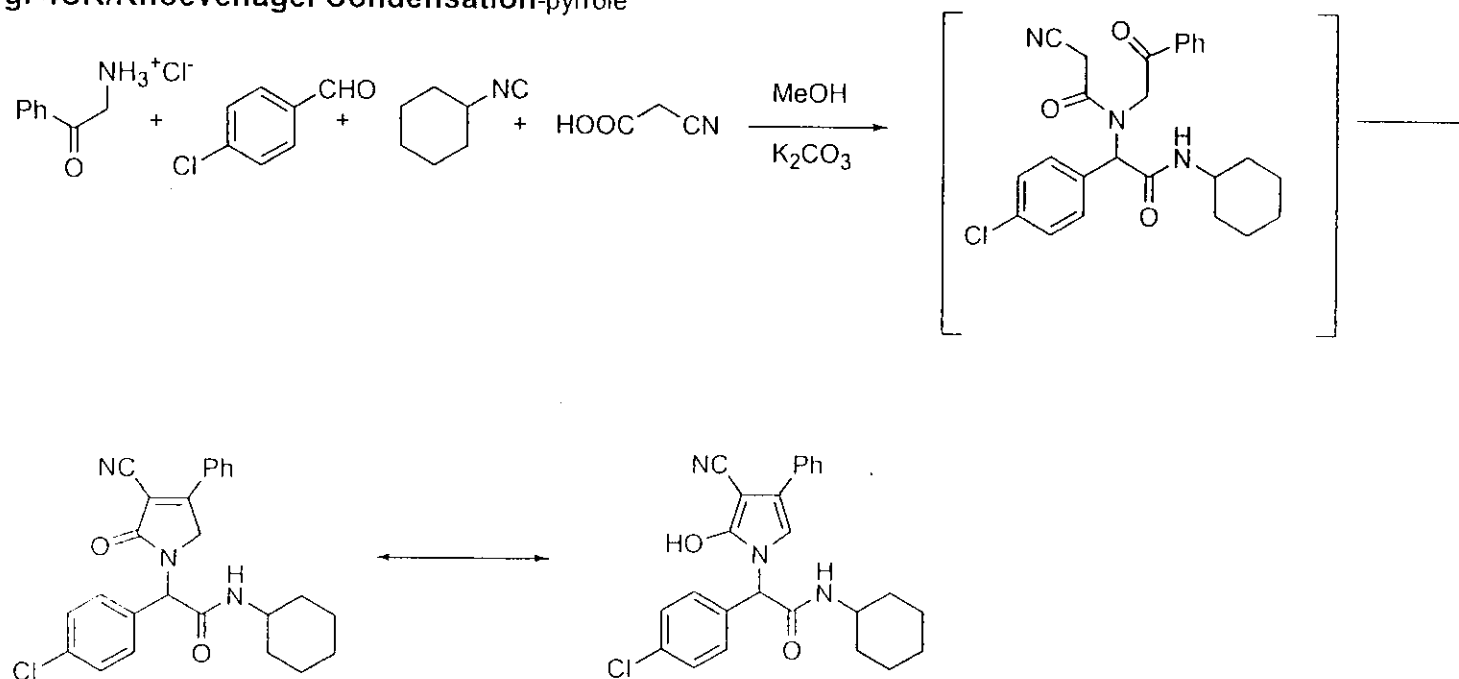


Weber, L. *Current Medicinal Chemistry* 2002, 9, 1241  
Fukuyama, I. *et al. J. Am. Chem. Soc.* 2002, 124, 6552

## Combination of Ugi Reaction with Other Transformations:

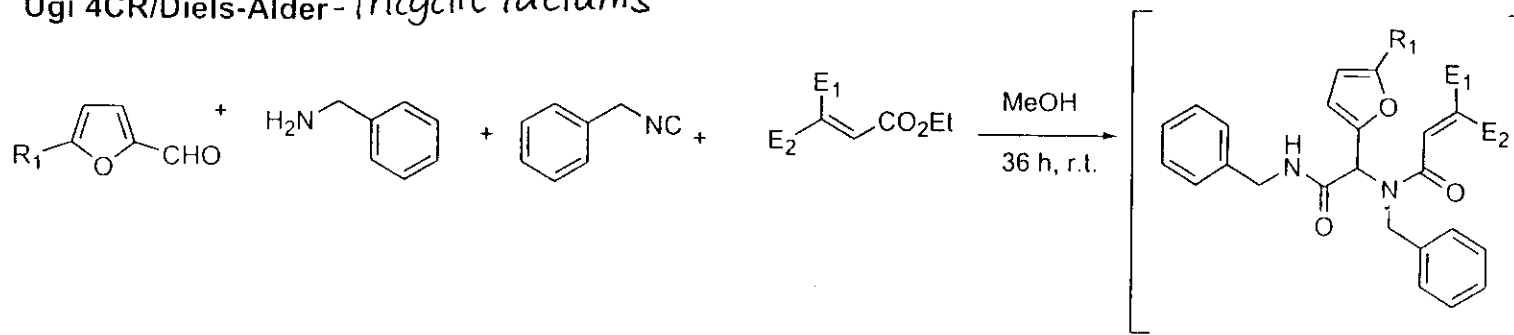
- possible when starting materials bear other functional groups, that react after initial formation of Ugi adduct
- constraints--Ugi requires protic solvents- select a transformation compatible with protic solvents or develop novel conditions for Ugi

### Ugi 4CR/Knoevenagel Condensation-pyrrole

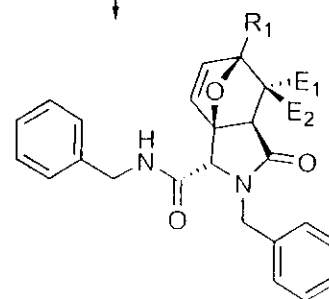


## Combination of Ugi Reaction with Other Transformations:

### Ugi 4CR/Diels-Alder-tricyclic lactams



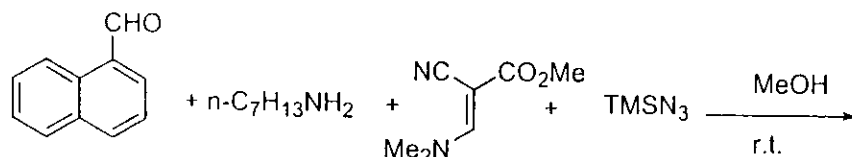
Entry	$R_1$	$E_1$	$E_2$	Yield %	Isomer Ratio
a	H	CO <sub>2</sub> Et	H	89	92:8
b	H	H	CO <sub>2</sub> Et	70	92:8
c	H	H	CONH-Bn	85	86:14
d	Me	CO <sub>2</sub> Et	H	78	92:8
e	Me	H	CO <sub>2</sub> Et	72	93:7
f	Me	H	CONH-Bn	81	83:17



-can be used with polymer support

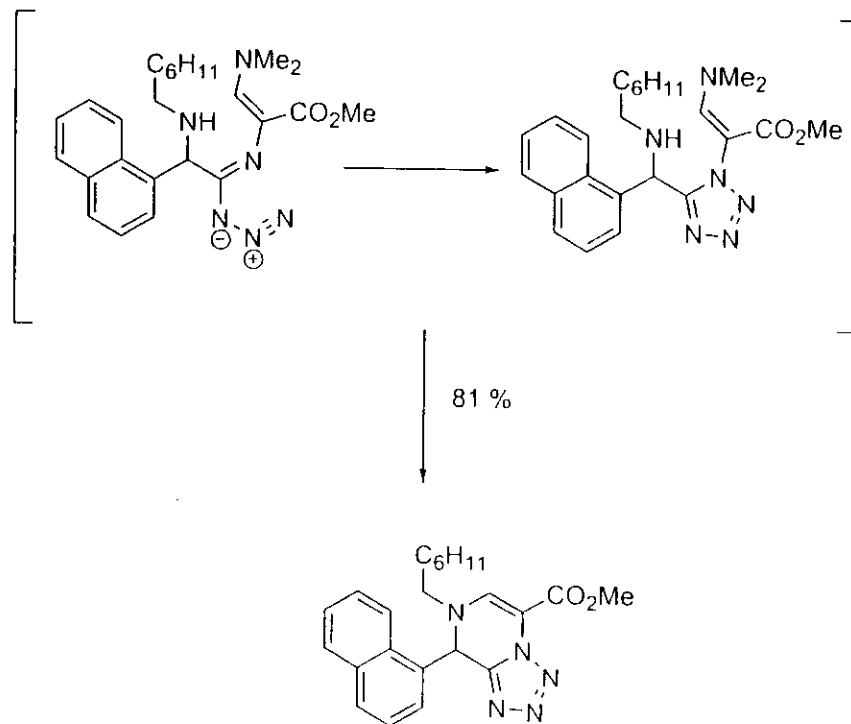
## Combination of Ugi Reaction with Other Transformations:

### Ugi 4CR/Intramolecular 1,4-Addition and Elimination



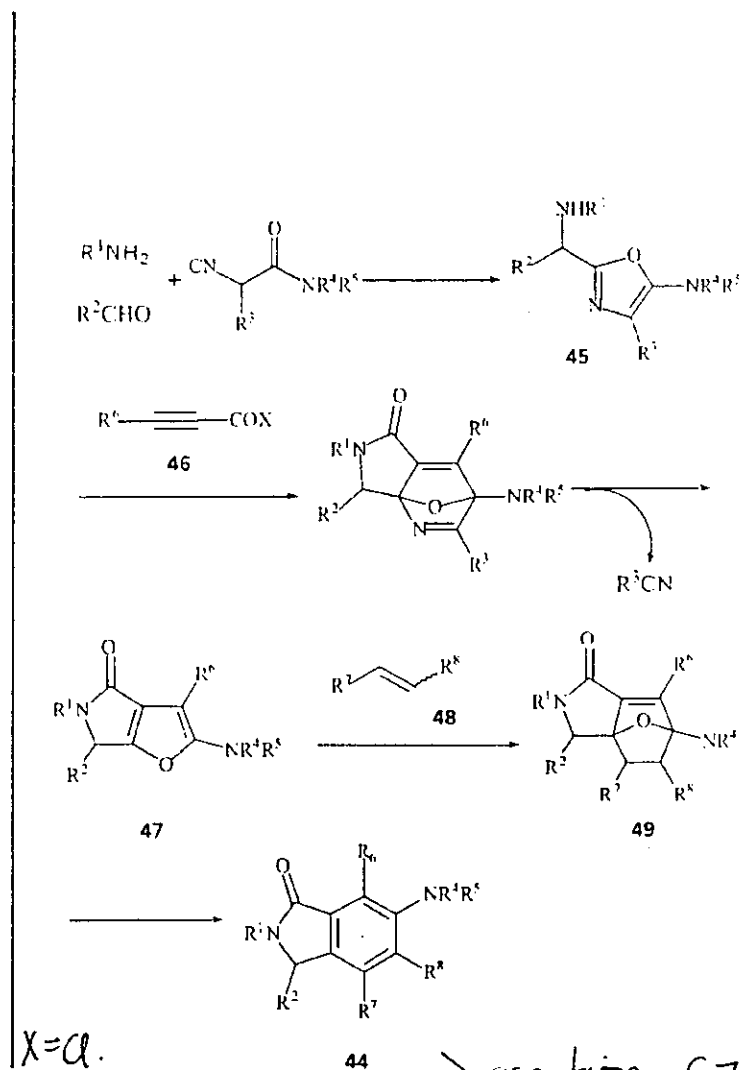
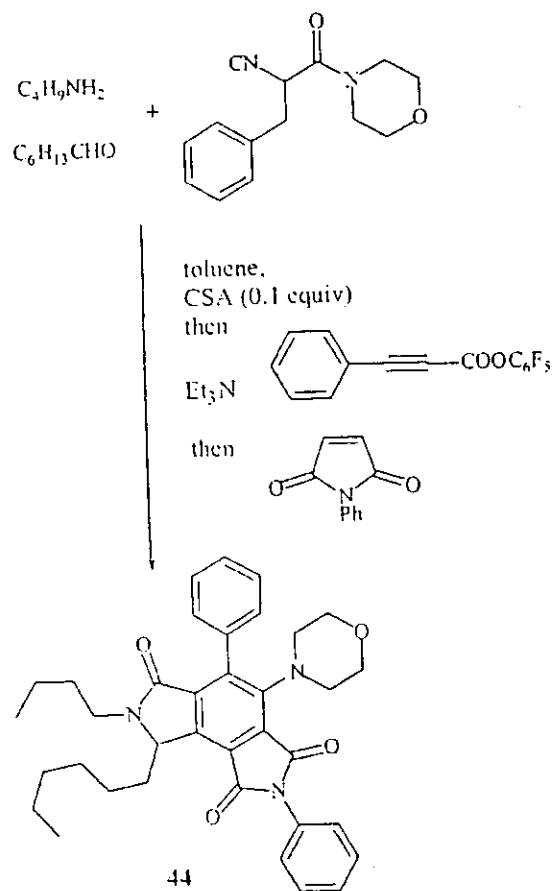
Entry	R1	R2	R3	Yield (A + B) Step 1 - %	Yield (C) Step 2 - %	Yield (C) Step 1+2 - %
1			Me	42	---	---
2			Me	68	100	---
3			Me	27 (cryst.)	100	---
4			Me	65	---	---
5			Me	46 (cryst.)	99	---
6			Me	---	---	81
7			Me	---	---	59
8			Me	---	---	46
9			Me	---	---	48
10			Me	---	---	46
11			Me	---	---	29
12			Me	---	---	72

Bemayme, H. Boucrid, K. *Tetrahedron Lett.* 1998, 2735

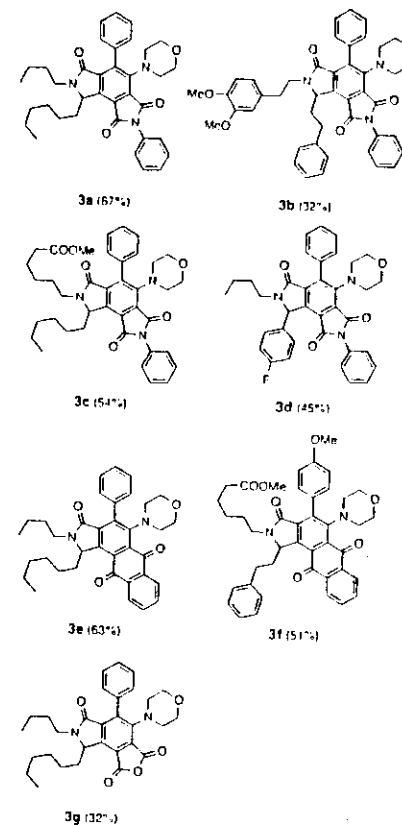


## 5-Component Reaction

-Hexasubstituted benzenes

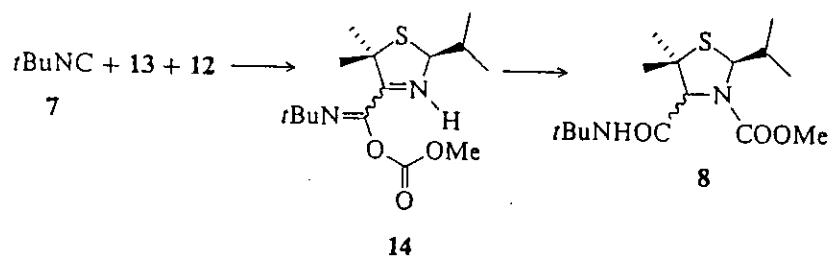
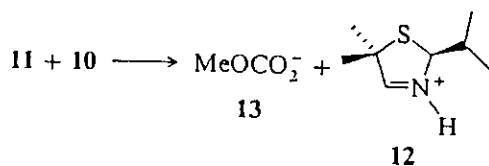
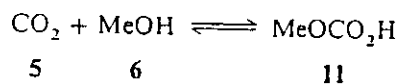
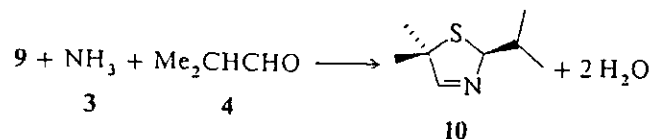
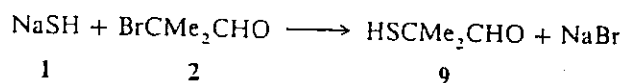
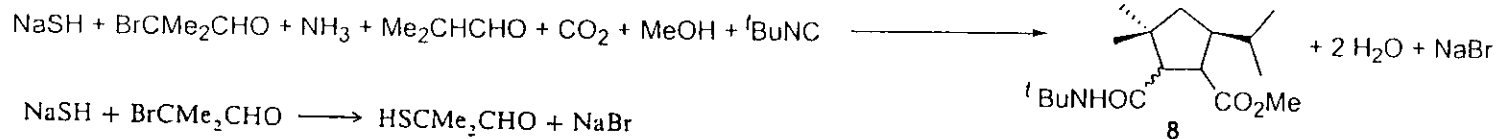


$\Rightarrow$  creation of 7-chemical bonds  
 5 points for diversity





## Seven-Component Reaction



Scheme 2. Mechanism of the 7CC.

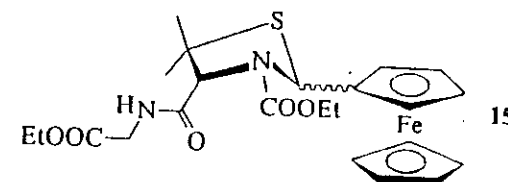
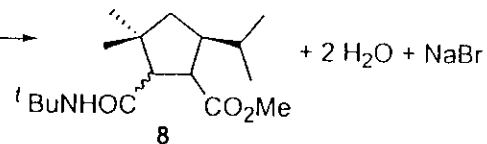
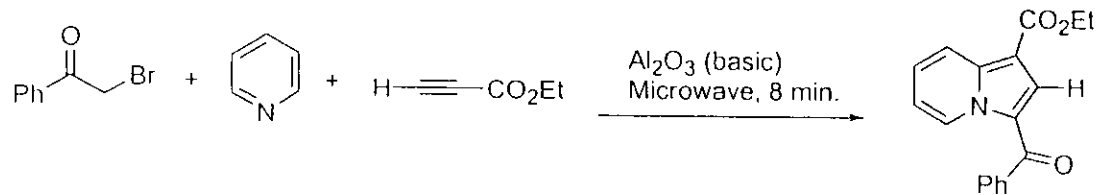


Table 1. Conditions and results of the syntheses of 8 and 15.

Product	Conversion [%] [a]	Yield [%] [b]	Diastereo- meric ratio [c]	Solvent [d]	<i>p</i> [bar]
8	70	43	2:1	MeOH	16
15	58	18	14:9	EtOH	48

[a] Based on unchanged Asinger product (determined from the <sup>1</sup>H NMR spectrum of the reaction mixture). [b] Pure substance after flash chromatography. [c] Determined from the <sup>1</sup>H NMR spectrum of the reaction mixture. [d] Alcoholic component.

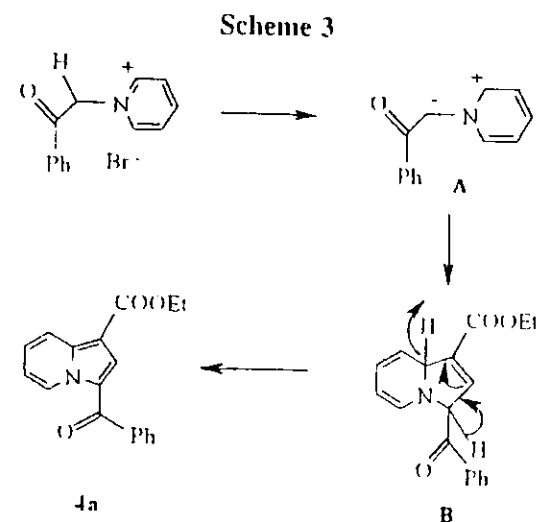
Microwave-Mediated MCR's:  
Indolizines



**Table 1.** Microwave-Promoted Three-Component Reaction<sup>a</sup> According to Scheme 1 Catalyzed by Alumina<sup>b</sup>

entry	catalyst	yield <sup>c</sup> of <b>4a</b> (%)
1	$\text{Al}_2\text{O}_3$	92
2	Nil	12
3	pyridine <sup>d</sup>	48
4	pyridine/ $\text{Al}_2\text{O}_3$	78
5	toluene/ $\text{Al}_2\text{O}_3$	68
6	THF/ $\text{Al}_2\text{O}_3$	60
7	DMF/ $\text{Al}_2\text{O}_3$	75
8	toluene/ $\text{Et}_3\text{N}/\text{Al}_2\text{O}_3$	80
9	THF/ $\text{Et}_3\text{N}/\text{Al}_2\text{O}_3$	76
10	DMF/ $\text{Et}_3\text{N}/\text{Al}_2\text{O}_3$	82

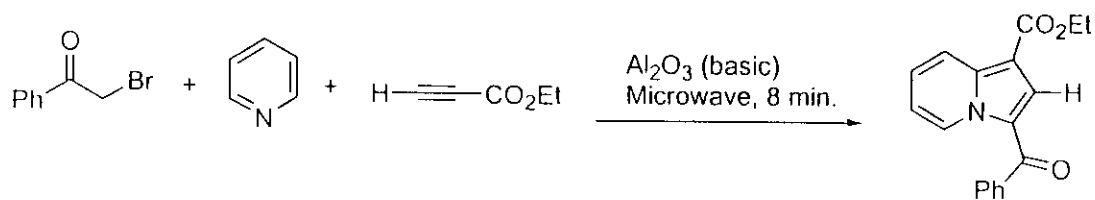
<sup>a</sup> Reactions were carried using 1 mmol of phenylacetyl bromide, 1.2 equiv of pyridine, 1.2 equiv of ethyl propiolate, and 1 g of basic alumina. <sup>b</sup> Basic alumina was activated at 450 °C for 12 h. <sup>c</sup> Isolated yields. <sup>d</sup> An excess of pyridine was used without alumina.



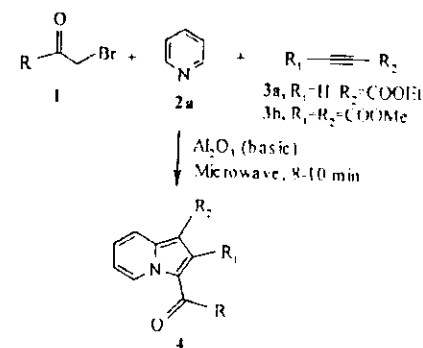
**Table 3.** Three-Component Reaction of **1a**, **2**, and **3a** under Thermal Condition

entry	solvent	base	yield of <b>4a</b> (%)
1	toluene	$\text{Et}_3\text{N}$	60
2	THF	$\text{Et}_3\text{N}$	55
3	$\text{CH}_3\text{CN}$	$\text{Et}_3\text{N}$	50
4	pyridine	$\text{Et}_3\text{N}$	80
5	toluene	$\text{Al}_2\text{O}_3$	45
6	THF	$\text{Al}_2\text{O}_3$	42
7	$\text{CH}_3\text{CN}$	$\text{Al}_2\text{O}_3$	35
8	pyridine	$\text{Al}_2\text{O}_3$	88

Microwave-Mediated MCR's:  
Indolizines



**Table 2.** Microwave-Mediated Three-Component Reaction of Acyl Halide **1**, Pyridine **2a**, and Acetylene **3** Catalyzed by Alumina

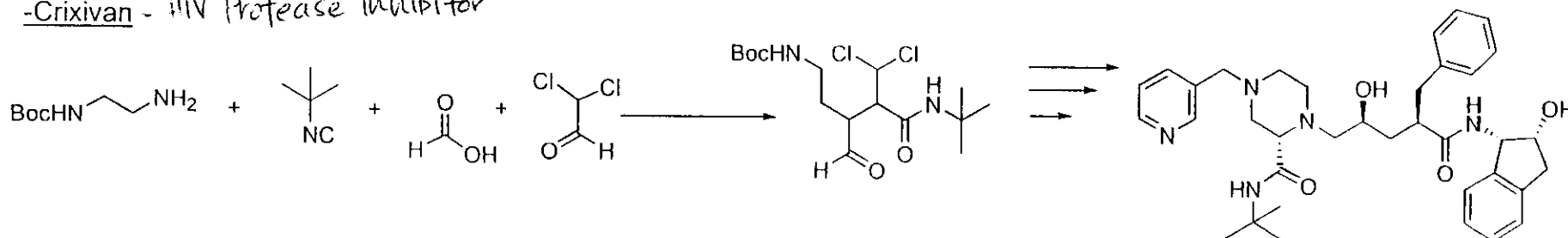


Entry	Product			Yield* (%)			
	R	R <sub>1</sub>	R <sub>2</sub>	Solid <sup>b</sup>	Solvent		
1	Ph	1a	COOMe	COOMe	4b	94	71
2	<i>p</i> -Tolyl	1b	H	COOEt	4c	93	65
3	<i>p</i> -Tolyl	1b	COOMe	COOMe	4d	91	62
4	Styryl	1c	H	COOEt	4e	90	66
5	Styryl	1c	COOMe	COOMe	4f	87	60
6		1d	H	COOEt	4g	94	70
7	AcO		COOMe	COOMe	4h	93	67
8		1d	H	COOEt	4i	92	64
9	BzO		COOMe	COOMe	4j	90	62

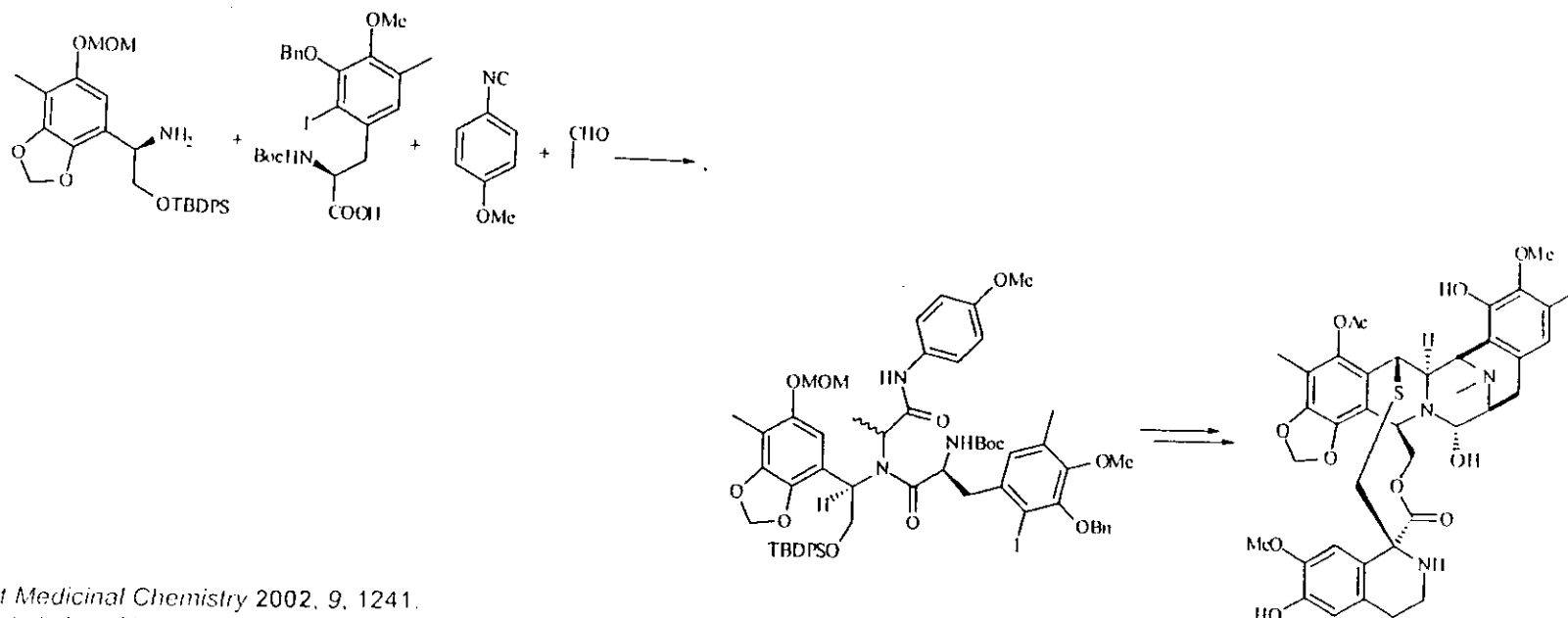
## Multi-Component Reactions: Applications in Drug Discovery

-useful for generating diverse libraries quickly

-Crixivan - HIV Protease Inhibitor



-Ecteinascidin 743 - anti-cancer



Weber, L. *Current Medicinal Chemistry* 2002, 9, 1241.  
 Fukuyama, T. *et al. J. Am. Chem. Soc.* 2002, 124, 6552  
 Nilu Jayasuriya @ Wipf Group

Scheme 12 Total synthesis of Ecteinascidin 743 with Ugi-4CR as key step.

12/20/2003

## Tandem Multi-Component Reaction

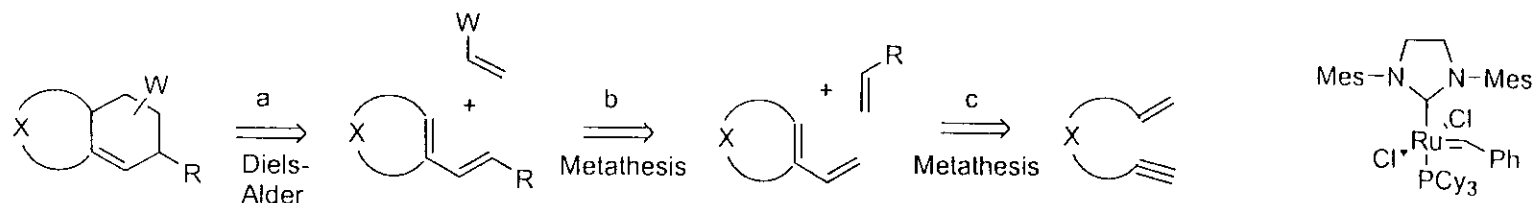
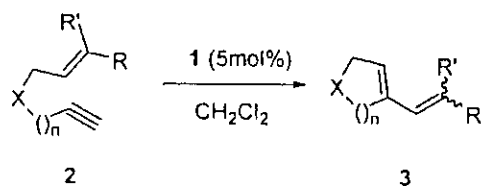


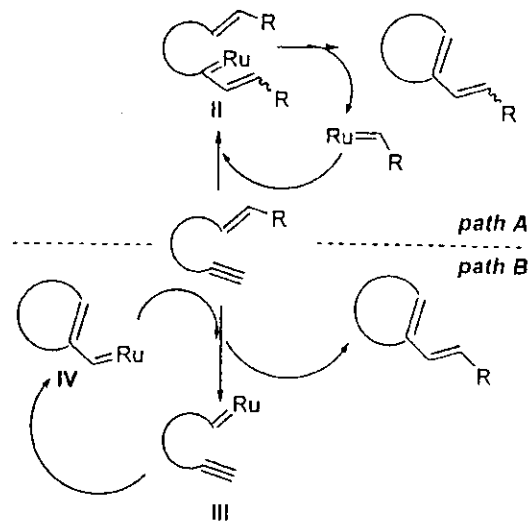
Table 1. Enyne Metathesis Reaction<sup>11</sup>



enyne	X	n	R	R'	product	yield <sup>a</sup> (E/Z) <sup>b</sup>
2a	NTs	1	CH <sub>3</sub>	H	3a	85% (3/1)
2b	NTs	1	CH <sub>2</sub> OH	H	3b	72% (3/1)
2c	NTs	1	CH <sub>3</sub>	CH <sub>3</sub>	3c	98%
2d	NTs	2	CH <sub>3</sub>	H	3d	82% (6/1)
2e	NTs	2	CH <sub>2</sub> OH	H	3e	86% (6/1)
2f	NTs	2	CH <sub>3</sub>	CH <sub>3</sub>	3f	92%
2g	CE <sub>2</sub> <sup>c</sup>	1	CH <sub>3</sub>	H	3g	83% (6/1)
2h	CE <sub>2</sub>	1	CH <sub>2</sub> OH	H	3h	88% (16/1)
2i	CE <sub>2</sub>	1	CH <sub>3</sub>	CH <sub>3</sub>	3i	90%
2j	CE <sub>2</sub>	2	CH <sub>3</sub>	H	3j	82% (16/1)
2k	CE <sub>2</sub>	2	CH <sub>2</sub> OH	H	3k	96% (12/1)
2l	CE <sub>2</sub>	2	CH <sub>3</sub>	CH <sub>3</sub>	3l	93%

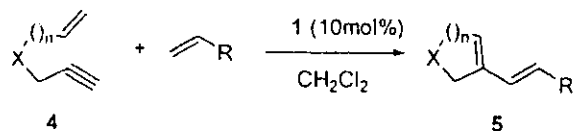
<sup>a</sup> Isolated yield. <sup>b</sup> E/Z ratio was determined by <sup>1</sup>H NMR. <sup>c</sup> E = COOEt.

Scheme 3. Plausible Mechanistic Pathways



## Tandem Enyne, Diene-ene Metathesis Reaction

Table 2. Tandem Enyne, Diene-ene Metathesis Reaction<sup>15</sup>



enyne	X	n	R	product	yield <sup>a,b</sup>
4a	NTs	1	Bu	5a	86%
4a	NTs	1	(CH <sub>2</sub> ) <sub>2</sub> ac	5b	88%
4a	NTs	1	Ph	5c	86%
4a	NTs	1	Bn	5d	84%
4a	NTs	1	(CH <sub>2</sub> ) <sub>2</sub> OBz	5e	73%
4a	NTs	1	(CH <sub>2</sub> ) <sub>2</sub> Br	5f	78%
4a	NTs	1	CH <sub>2</sub> TMS	5g	93%
4b	NTs	2	Bu	5h	85%
4b	NTs	2	(CH <sub>2</sub> ) <sub>2</sub> ac	5i	74%
4b	NTs	2	Ph	5j	68%
4b	NTs	2	Bn	5k	86%
4b	NTs	2	(CH <sub>2</sub> ) <sub>2</sub> OBz	5l	66%
4b	NTs	2	(CH <sub>2</sub> ) <sub>2</sub> Br	5m	74%
4b	NTs	2	CH <sub>2</sub> TMS	5n	81%
4c	CE <sub>2</sub>	1	Bu	5o	85%
4c	CE <sub>2</sub>	1	(CH <sub>2</sub> ) <sub>2</sub> ac	5p	63%
4c	CE <sub>2</sub>	1	Ph	5q	87%
4c	CE <sub>2</sub>	1	Bn	5r	86%
4c	CE <sub>2</sub>	1	(CH <sub>2</sub> ) <sub>2</sub> OBz	5s	73%
4c	CE <sub>2</sub>	1	(CH <sub>2</sub> ) <sub>2</sub> Br	5t	82%
4c	CE <sub>2</sub>	1	CH <sub>2</sub> TMS	5u	65%

<sup>a</sup> Isolated yield. <sup>b</sup> E/Z ratio was determined to be >20:1 by <sup>1</sup>H NMR.  
<sup>c</sup> E = COOEt. <sup>d</sup> ac = CH<sub>3</sub>CO

## Three-Component Metathesis/Diels-Alder Reaction

Scheme 4. Multicomponent Tandem Reaction Sequence

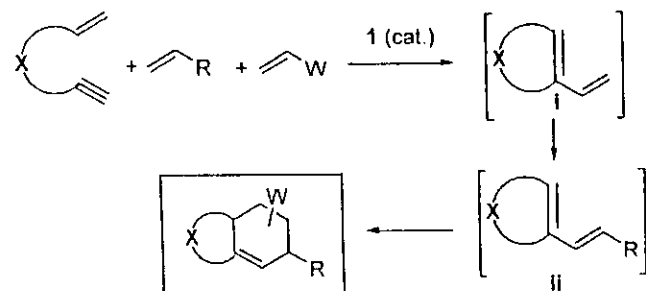


Table 3. Three-Component Metathesis/Diels-Alder Reaction<sup>17</sup>

enyne	X	n	R	product <sup>a</sup>	yield <sup>b</sup>
4a	NTs	1	Bu	6a	83%
4a	NTs	1	Bn	6b	75%
4a	NTs	1	(CH <sub>2</sub> ) <sub>2</sub> Br	6c	71%
4a	NTs	2	Bu	6d	78%
4a	NTs	2	Bn	6e	78%
4a	NTs	2	(CH <sub>2</sub> ) <sub>2</sub> Br	6f	71%
4a	CE <sub>2</sub>	1	Bu	6g	74%
4b	CE <sub>2</sub>	1	Bu	6h	74%
4b	CE <sub>2</sub>	1	(CH <sub>2</sub> ) <sub>2</sub> Br	6i	77%

<sup>a</sup> Relative stereochemistry of the product was confirmed through NOE experiment. <sup>b</sup> Isolated yield. <sup>c</sup> E = COOEt.