

Lewis Acid-Promoted Imine Synthesis by the Insertion of Isocyanides into C–H Bonds of Electron-Rich Aromatic Compounds

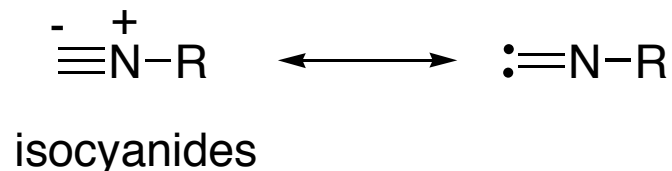
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Suita, Osaka 565-0871, Japan*

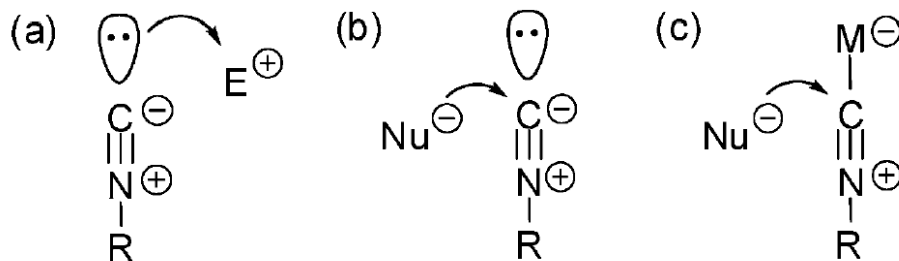
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Chatani, N. *et al.*, *Org. Lett.* **2007**, *9*, 3351-3353

Ambiphilic Reactivity of Isocyanides



- Isocyanides possess unique reactivity due to the presence of a formally divalent carbon atom

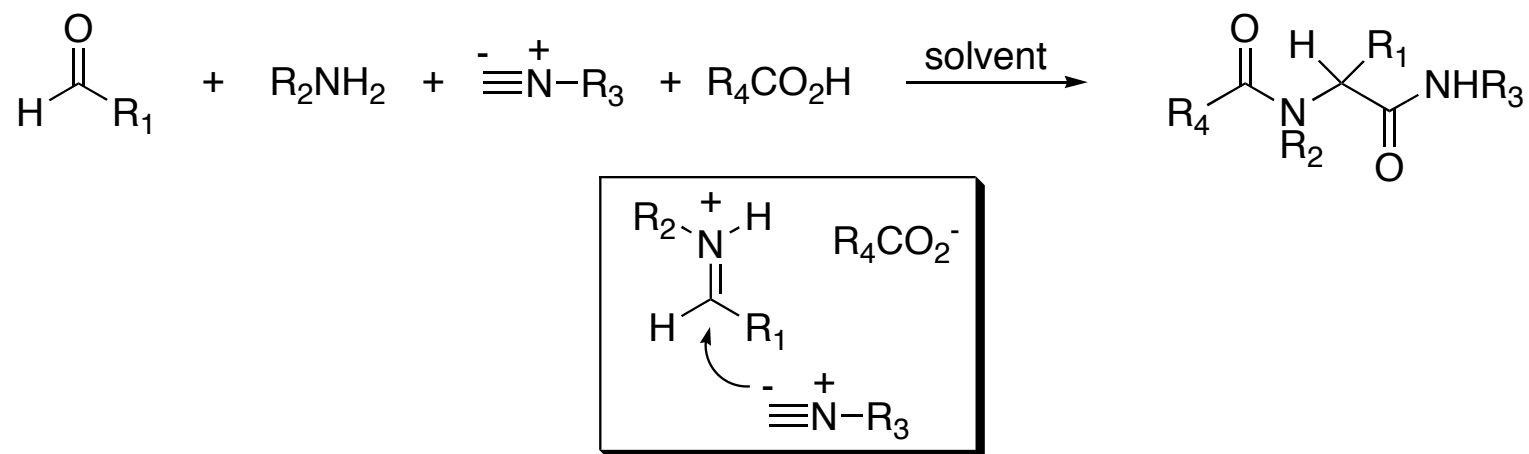


- Nucleophilic reactions dominate; electrophilic reactions commonly observed with strong nucleophiles

Domling, A., *Chem. Rev.* **2006**, *106*, 17-89

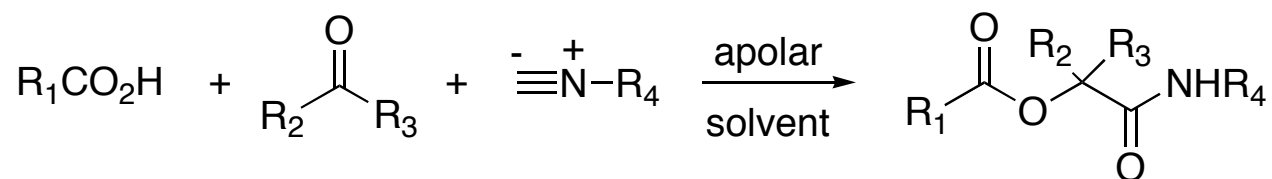
Isocyanides As Nucleophiles

Ugi Multicomponent Coupling Reaction



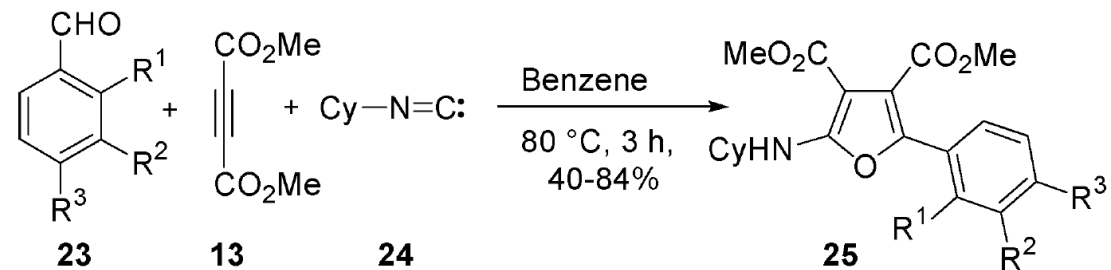
Ugi, I., *Pure Appl. Chem.* **2001**, 73, 187-191

Passerini Reaction

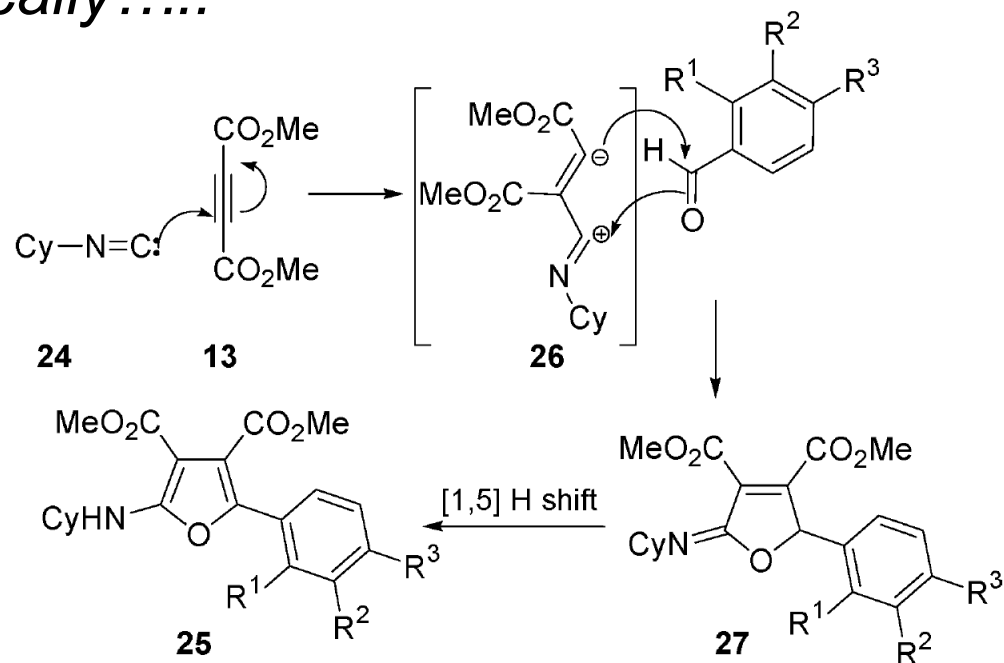


Ostaszewski, R. *et al.*, *Pure Appl. Chem.* **2003**, 75, 413-419

Synthesis of substituted Furans

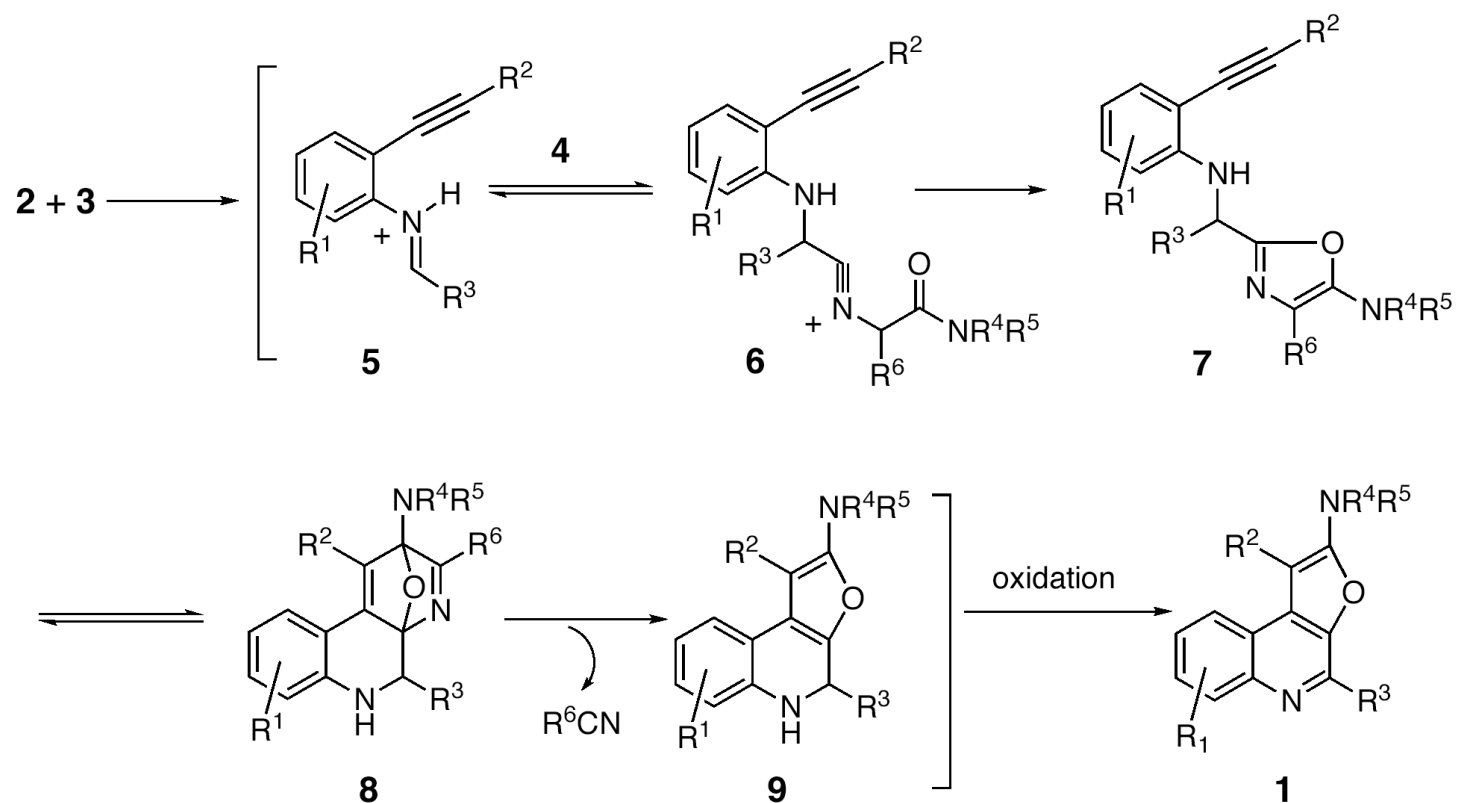
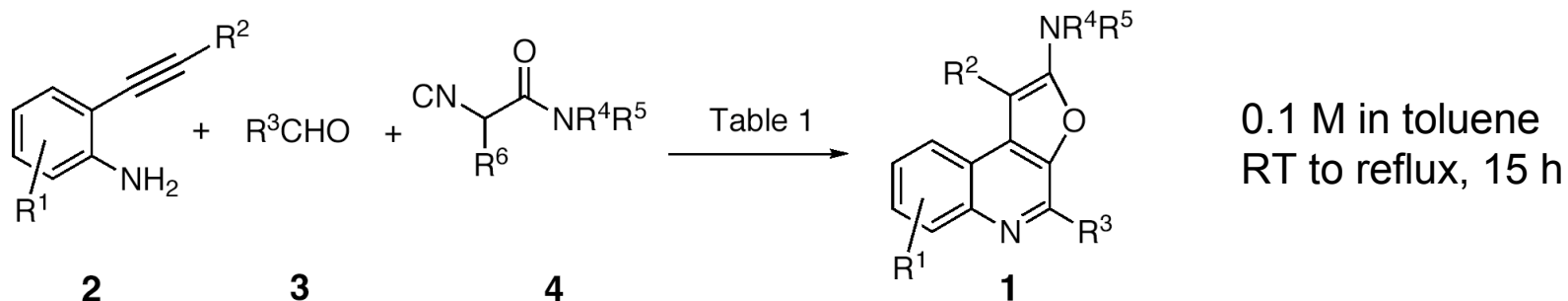


Mechanistically.....



Nair, V.; Vinod, A. U., *Chem. Commun.* **2000**, 1019-1020

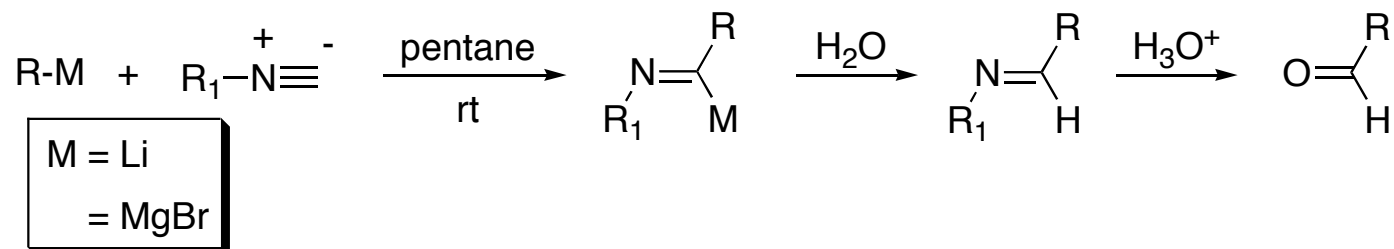
Synthesis of Highly Substituted Furo[2,3-c]quinolines



Zhu, J.; Fayol, A., *Angew. Chem. Int. Ed.* **2002**, 41, 3633-3635

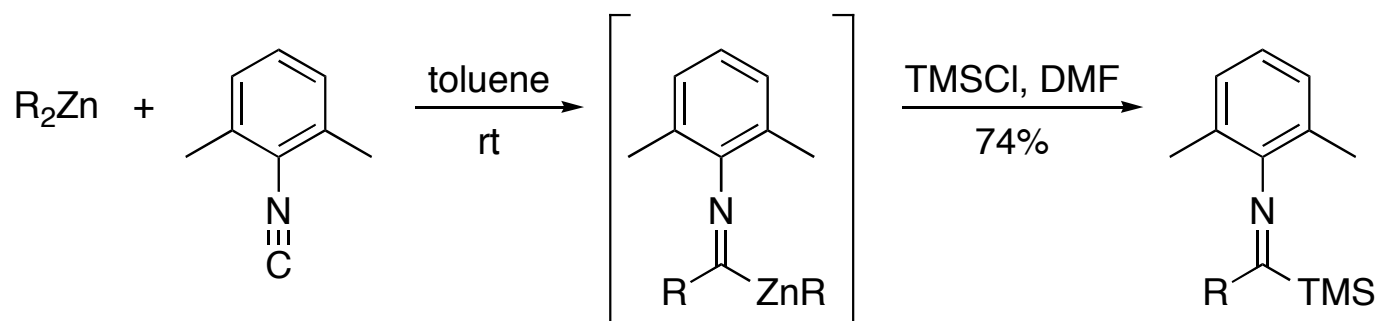
Isocyanides As Electrophiles

Addition of Alkyl Lithium and Alkyl Grignard Reagents



Walborsky, H. M. *et al.*, *J. Org. Chem.* **1974**, 39, 600-604

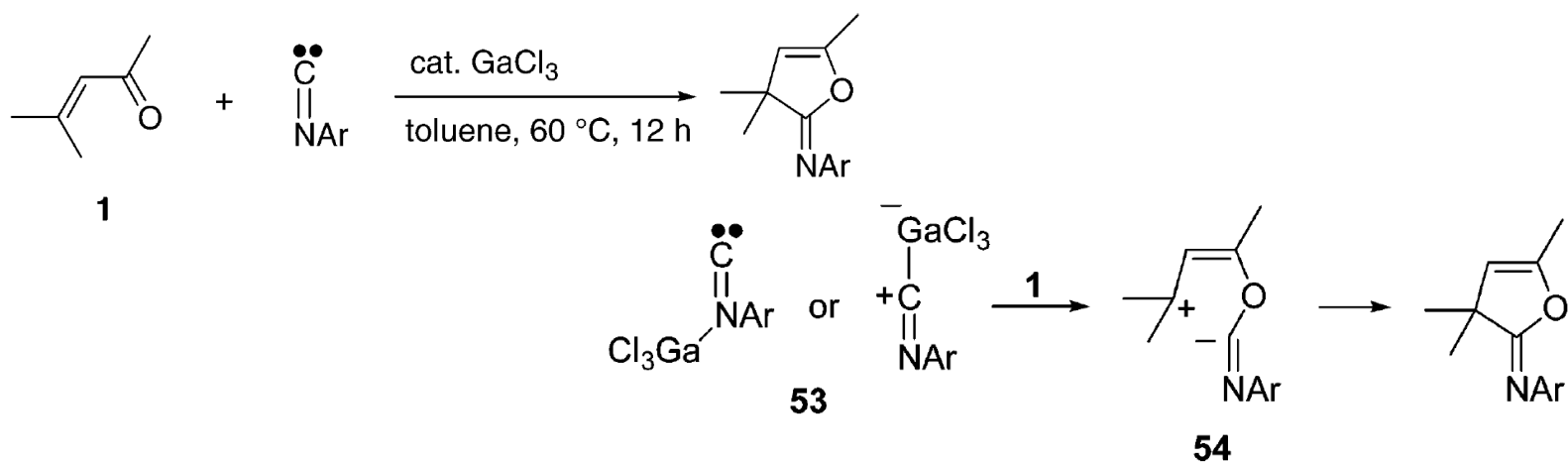
Addition of Dialkyl Zinc Reagents



Ito, Y. *et al.*, *J. Org. Chem.* **1988**, 53, 4158-4159

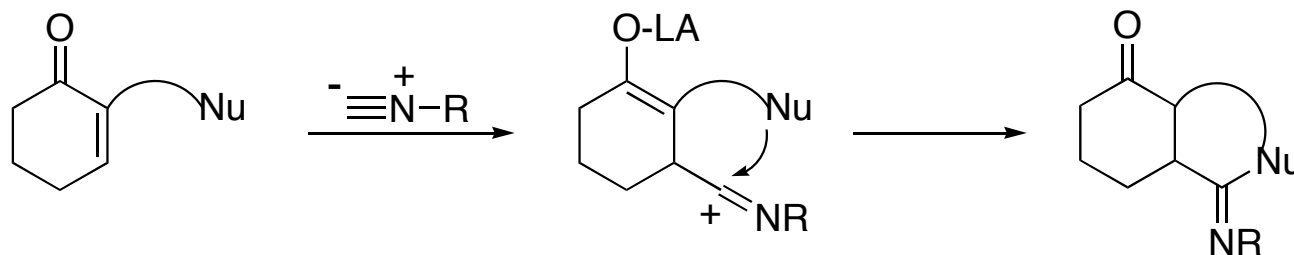
Isocyanides As Electrophiles

Activation with Lewis Acids: *s-cis* ketones



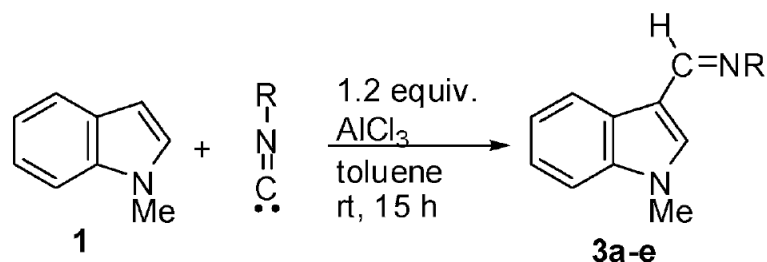
Chatani, N. *et al.*, *J. Am. Chem. Soc.* **2005**, 127, 761-766

Activation with Lewis Acids: *s-trans* ketones

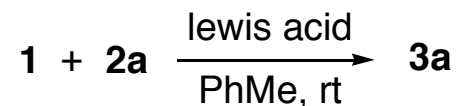


Winkler, J. D.; Asselin, S. M., *Org. Lett.* **2006**, 8, 3975-3977

Initial Experiments

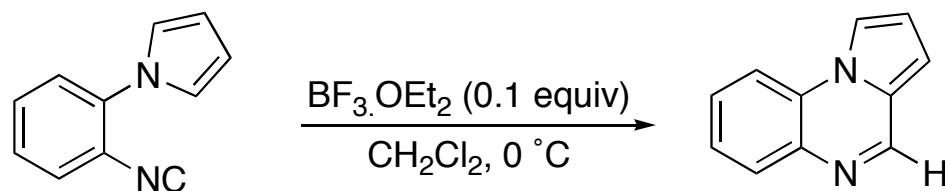


R = 2,6-Me ₂ C ₆ H ₃ (2a)	88%
3,5-Me ₂ C ₆ H ₃ (2b)	68%
4-MeOC ₆ H ₄ (2c)	79%
4-ClC ₆ H ₄ (2d)	64%
benzyl (2e)	82%



entry	lewis acid	yield (%)
1	BF ₃ .OEt ₂	68
2	GaCl ₃	86
3	In(OTf) ₃	90

- A formal insertion of the isocyanide into the aromatic C-H bond
- Group 13 elements: excellent promoters
- Intermolecular reaction is unprecedented



Kobayashi, K., *et al.*, *Chem. Lett.* **1998**, 9, 551-552

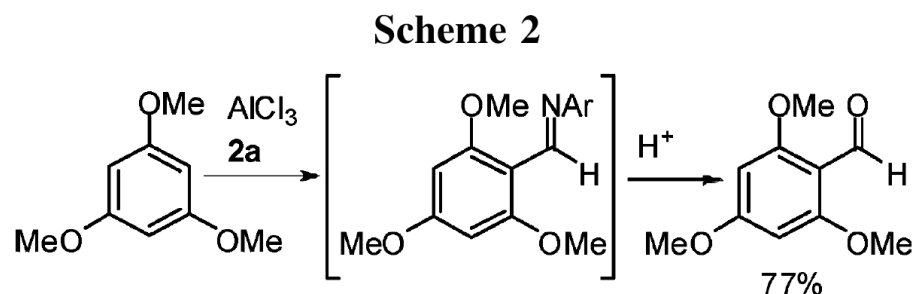
Chatani, N. *et al.*, *Org. Lett.* **2007**, 9, 3351-3353

Substrate Scope

entry	aromatics	product ^b	yield (%) ^c
1			R = Me 88
2			R = CH ₂ Ph 81
3			R = 4-MeC ₆ H ₄ 93
4			R = H 78
5			R' = Me 76
6			R' = Ph 73
7			R' = OMe 80
8			R' = Br 89
9			R' = CO ₂ Me 83
10 ^d			R'' = Me 85 run at
11			R'' = Ph 92 60 °C
12			84 ^e + 3-subst. isomer (6%)
13			78
14 ^d			82 run at 60 °C

Ar = 2,6-dimethylphenyl

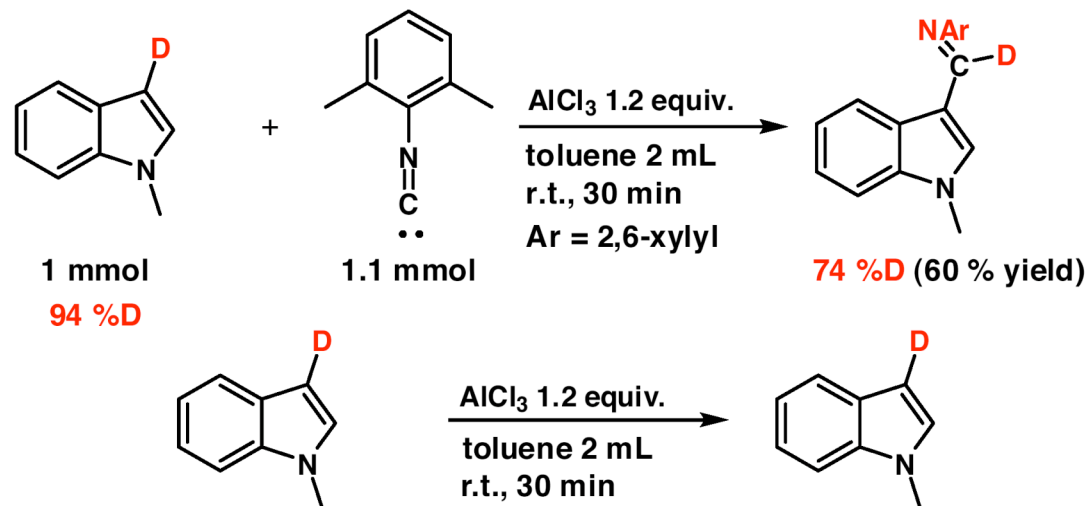
Heterocycle (1.0 mmole)
Isocyanide (1.2 mmole)
AlCl₃ (1.2 mmole),
toluene, rt, 15 h



- AlCl₃-mediated insertion exhibits a wide substrate scope
- Electron withdrawing groups on indole: *not* tolerated

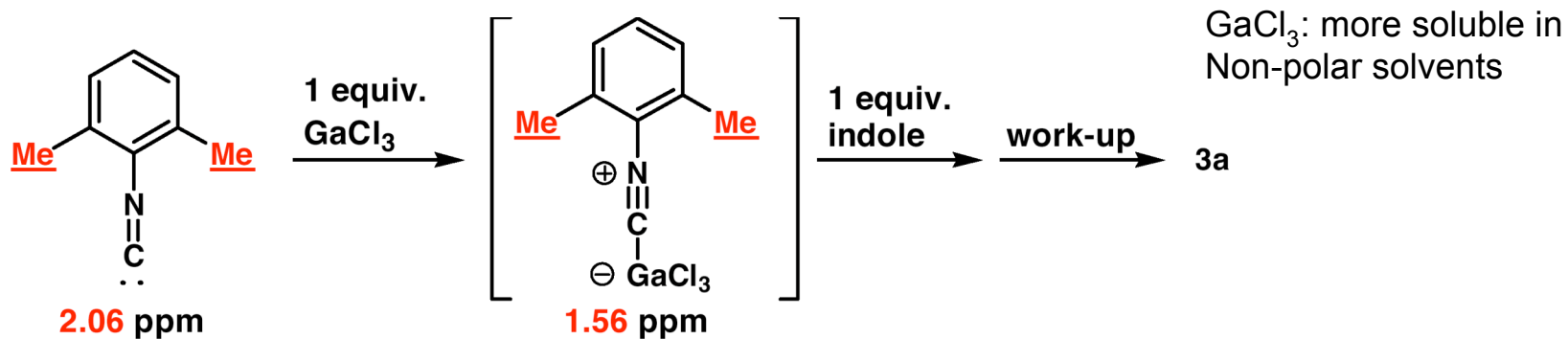
Chatani, N. *et al.*, *Org. Lett.* **2007**, *9*, 3351-3353

Labeling Experiments



- H at C-3 is transferred to the imino carbon in the product

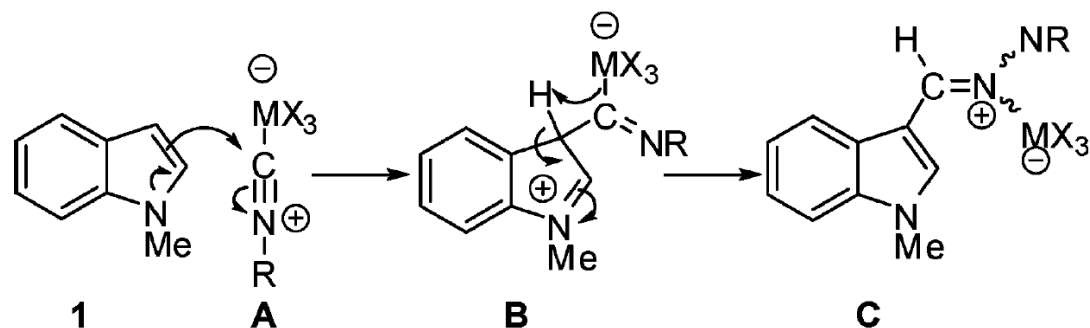
¹H-NMR experiment with GaCl₃



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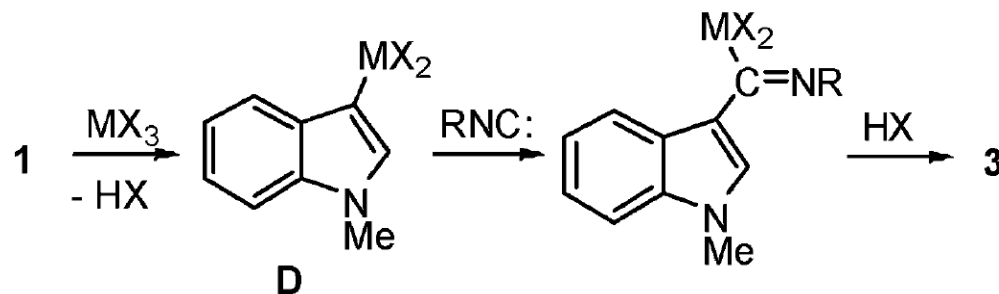
Proposed Mechanism

Scheme 3



Nucleophilic attack of the indole onto LA-activated isocyanide
Rearomatization *via* deprotonation and protonolysis of C-M bond

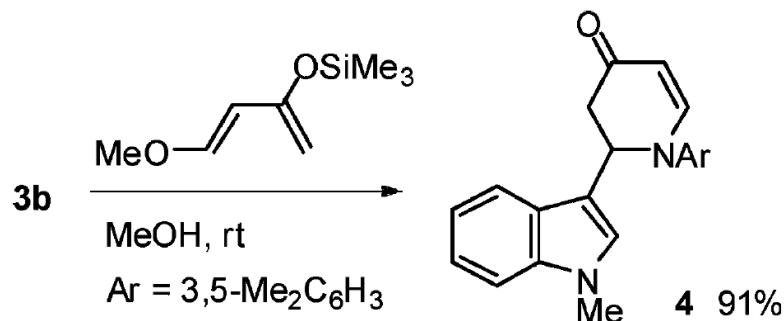
Alternative Mechanism



- Treatment of a solution of indole + isocyanide with LA: no interaction between indole and LA

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Further Structural Elaboration

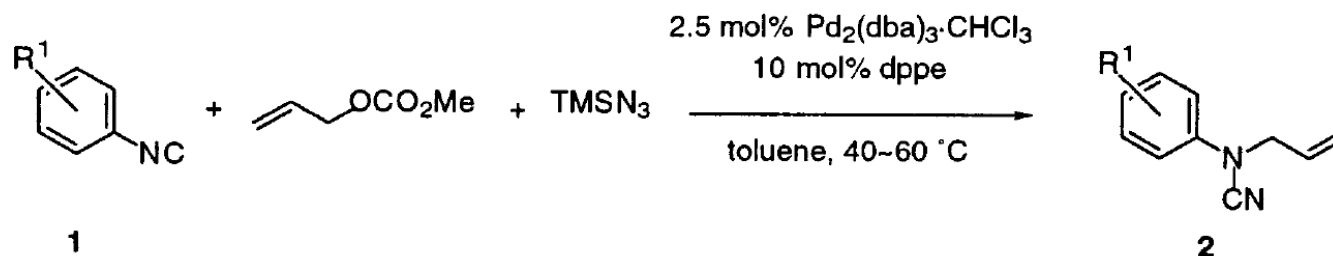


Conclusions

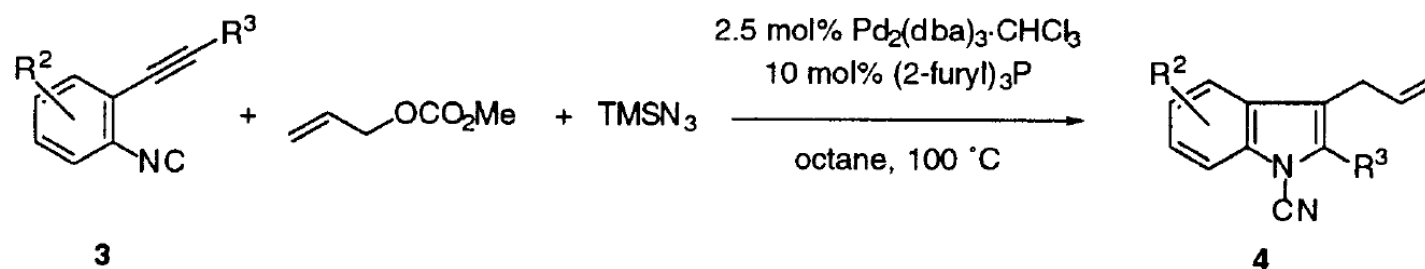
- A novel protocol for the α -addition of isocyanides to aromatic compounds promoted by inexpensive LAs has been developed
- This method provides a direct and practical access to a wide variety of imines via aromatic C-H bond functionalization
- Ongoing research seeks to explore this new isocyanide insertion process into other chemical bonds

Isocyanides: Metal-Mediated MCRs

Synthesis of allyl aryl cyanamides

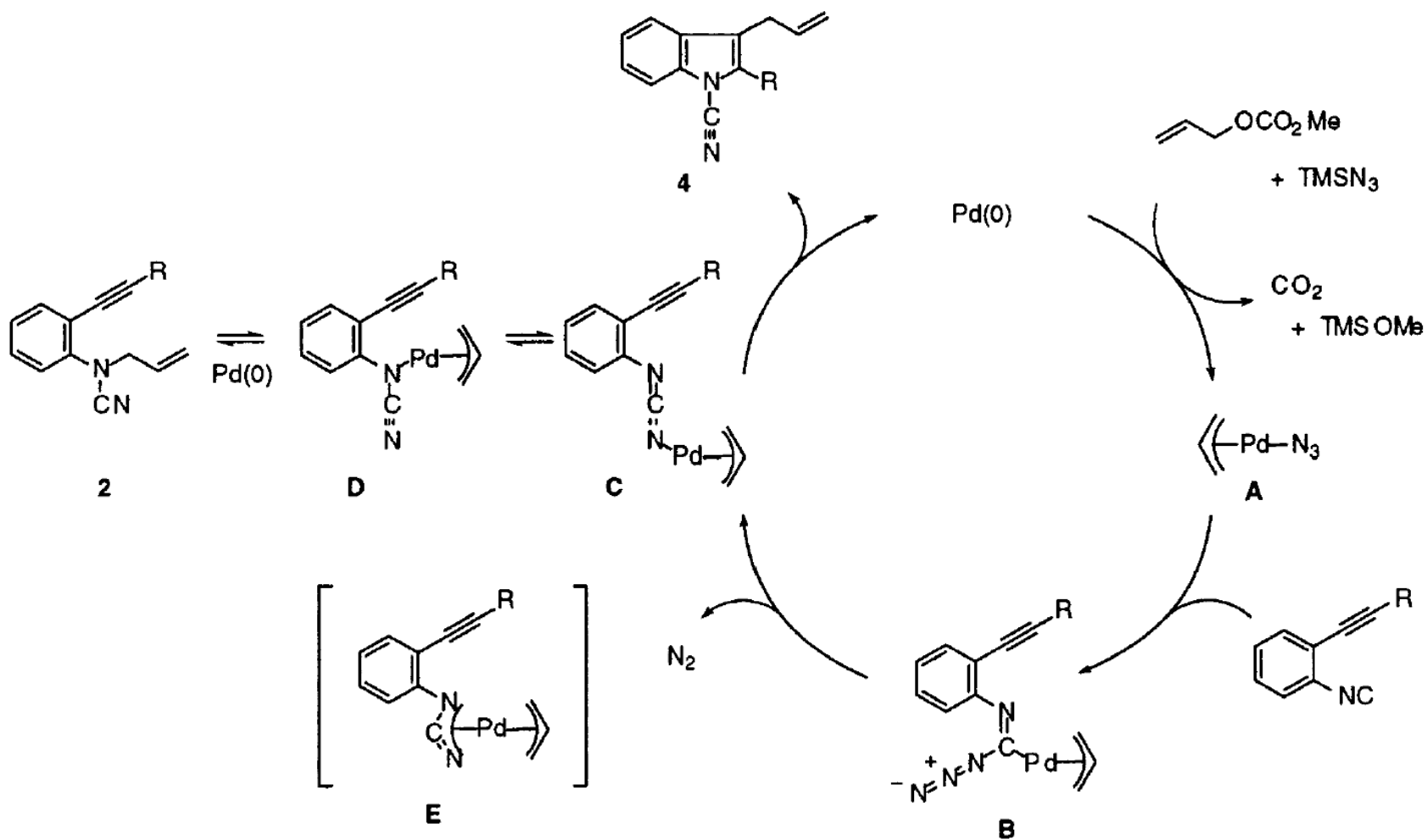


Synthesis of N-cyanoindoles



Kamijo, S.; Yamamoto, Y., *J. Am. Chem. Soc.* **2002**, *124*, 11940-11945

Mechanism of the formation of *N*-cyanoindoles



Kamijo, S.; Yamamoto, Y., *J. Am. Chem. Soc.* **2002**, *124*, 11940-11945